One century of experiments on electron-atom and molecule scattering: A critical review of integral cross-sections

III. - Hydrocarbons and halides

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1. – Introduction

Electron scattering on less common molecular species both of industrial and atmospheric interest, like halocarbons or halosilanes, have been studied for the first time in recent years. The present paper (part III) is a continuation of the review of cross-sections (CS) for atomic and diatomic molecules, part I (Zecca, Karwasz and Brusa 1996) and polyatomic molecules including HCl, part II (Karwasz, Brusa and Zecca 2001). The present paper deals with hydrocarbons (apart from CH_4 discussed in part II) and halides, including substituted hydrocarbons, substituted silanes (SiF₄ and so on) and hexafluorides (SF₆, WF₆, UF₆).

As in part II, some targets, like more complex organic molecules, for which little experimental work exists, are grouped in sections where it seemed to be most appropriate. In this manner, the reader will find a few comments and references on CH_3OH in subsect. **3**'3, "Methyl halides"; on BCl_3 in subsect. **3**'5 entitled "Trihalides (NF₃ and others)"; on C_6H_5Cl and C_6F_5Cl in subsect. **3**'7 "Hexafluorobenzene (C_6F_6)" and so on.

The review focuses on integral cross-sections, in particular on total cross-sections (TCS). Differential elastic cross-sections or integral cross-sections for electron attachment are present for some targets, where we have found these data significant for the physics of scattering. In figures presenting integral CS, the same energy span of 7 decades is maintained, in order to facilitate the comparisons. In targets, like CCl_4 , for which the CS (usually electron attachment CS) measurements extend below 1 meV, the energy scale was additionally expanded. In other gases, like SiF_4 , where TCS were measured in a limited energy range only, the E scale is reduced. In these cases the format of the drawings is changed, in order to stress the difference with the adopted standard.

Numerous "specialized" reviews have been dedicated to the detailed study of previous experimental and theoretical papers regarding integral and differential cross-sections on a limited number of targets (very often a single gas). Tawara et al. (1992) reviewed cross-sections for hydrocarbons. Christophorou et al. (1997a, b) reviewed extensively total, integral and differential cross-sections for CHF₃ and CF₂Cl₂, Christophorou and Olthoff (1998a, b) have made the same work for C_2F_6 and C_3F_8 , respectively. Large progress has also been achieved in theoretical treatment of electron-complex targets scattering, see for example Gulley and Buckman (1999) and Gianturco and Lucchese (1998) for benzene.

A sort of "reader's guide" has been published in the Introduction sections of part I and part II of the review. The scope of this guide is to help the user in reading the paper. The reader of the present work is referred to the Introduction of part II. There, he can find information regarding:

- numerical tables
- organization and layout of the figures
- organization of the reference lists
- special terminology
- experimental errors
- the methods used by us for the analysis of differential and vibrational cross-section data
- units.

The reference list of this subsection also includes a number of general references to review papers (Shimamura (1989), van de Burgt *et al.* (1989), Kauppila and Stein (1990), Hayashi (1992), Inokuti (1994), Smith and Španel (1994), Chutjian *et al.* (1996), Christophorou and Olthoff (1998a, b, 1999), Kimura *et al.* (2000).

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2. - Hydrocarbons

Electron scattering on hydrocarbons was studied in relatively few laboratories. Saturated hydrocarbons (alkanes) show similar features in TCS and for this reason will be presented in a single subsection (2[.]3). Cross-sections for unsaturated hydrocarbons, like acetylene (subsect. 2[.]1) and benzene (subsect. 2[.]2), show a wider variety of resonant structures, mainly in the total and vibrational excitation channels. An example of alkenes (ethene) is included in the "linear hydrocarbons" subsection (2[.]3). In subsect. 2[.]3 the reader will also find a comprehensive list of measurements on hydrocarbons in general.

2.1. Acetylene (C_2H_2). – The acetylene molecule is isoelectronic with N_2 and CO and it triggers a continuous theoretical interest. In particular, numerous calculations (Tossell 1985, Krumbach *et al.* 1989, Szmytkowski 1989, Khurana and Jain 1992, Jain 1993, Gianturco and Stoecklin 1994, Venkatnathan and Mishra 1998) predicted a shape ${}^2\Pi_{\rm g}$ resonance in analogy to the low-energy shape resonances in N_2 and CO, see figs. 21 and 25 in part I. Integral cross-sections for C_2H_2 are shown in fig. 1.

Total cross-sections. The first measurements of TCS by Brüche date to 1929 and cover the 1–40 eV energy range. Normalized TCS have been obtained by Sueoka and Mori (1989) between 0.7 and 400 eV; absolute TCS by Xing et al. (1994) between 400 eV and 2600 eV. The TCS shows a peak of $35.8 \cdot 10^{-20}$ m² at 2.5 eV and another bump with a maximum of $24.4 \cdot 10^{-20}$ m² at 7.5 eV (Sueoka and Mori 1989). Dressler and Allan (1987) measured the relative TCS in the 0.05-5 eV energy range. In fig. 1 we report their data normalized to the experiment of Sueoka and Mori (1989) at 2.5 eV: the two TCS curves coincide in the overlapping energy range.

In the 3–40 eV energy range the data of Sueoka and Mori (1989) are, on the average, 10% lower than those of Brüche (1929) and at 400 eV they are lower than the data of Xing et al. (1994) by the same amount. Below 2 eV the data of Sueoka and Mori are higher than the measurements of Brüche (1929); the use of a longitudinal magnetic field in Sueoka's and Mori apparatus can cause such an effect. The TCS of Sueoka and Mori (1989) are lower than the elastic CS calculation of Jain (1993) at the position of the maximum (2.5 eV) and than the theoretical elastic CS of Gianturco and Stoecklin (1994) in the whole 2–50 eV energy range.

At high energies the measurements of Xing et al. (1994) agree well with the optical model of Jain and Baluja (1992) and with the additivity rule of Jiang et al. (1995). Between 400 eV and 2600 eV the $\rm C_2H_2$ TCS is by 20–25% higher than both the $\rm N_2$ and CO total cross-sections (Xing et al. 1995).

Elastic cross-sections. Relative DCS at 10°-150° angular, 10-225 eV energy, ranges were reported by Hughes and McMillen (1933); DCS in the 0°-130° angular and 100-1000 eV energy ranges, normalized to the independent-atoms model, were given by Fink et al. (1975); differential and integral CS at 5-100 eV, normalized by the relative flow method, were obtained by Khakoo et al. (1993). Absolute DCS at a single energy 2.0 eV, 10°-105° angles were determined by Kochem et al. (1985). The DCS of Khakoo et al. agree well in shape with the relative measurements of Hughes and McMillen, except at small scattering angles. The values of DCS at 100 eV given by Fink et al. (1975) and Khakoo et al. differ roughly by a factor of two at 30°-130° scattering angles. At 100 eV the DCS obtained by the three groups (Fink et al., Khakoo et al., Hughes and McMillen) and normalized at 90° to the measurements of Khakoo et al. (1993), diverge for scattering angles below 15°. This discrepancy is probably due to the worsening of the angular resolution in the three apparatuses, respectively.

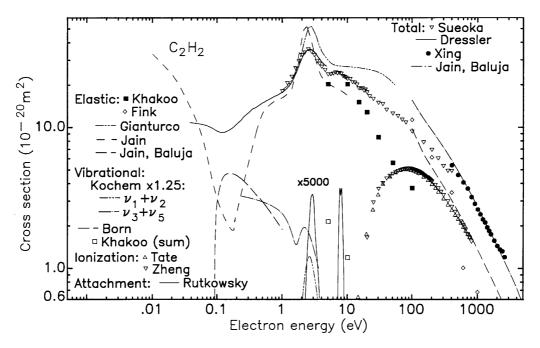


Fig. 1. – Integral cross-sections for electron scattering on C_2H_2 . Total: normalized data of Sueoka and Mori (1989); relative data of Dressler and Allan (1987) normalized to Sueoka and Mori at 2.5 eV; absolute of Xing et al. (1995); optical model of Jain and Baluja (1992). Elastic: absolute of Khakoo et al. (1993); Fink et al. (1975) normalized to the independent-atoms model and presently integrated ($\pm 15\%$ error); Jain (1993) and Jain and Baluja (1992), theoretical; Gianturco and Stoecklin (1994) theoretical. Vibrational: Kochem et al. (1985), multiplied by a factor of 1.25, see text; Khakoo et al. (1993), summed; Born approximation for ν_3 and ν_5 (eq. (15) in part I). Ionization: Tate and Smith (1932); Zheng and Srivastava (1996); data of Djurić et al. (1996), Gaudin and Hagemann (1967), Tian and Vidal (1998) and a single, 75 eV point of Lampe et al. (1957) are not shown as practically overlapping with the other data. Dissociative attachment: Rutkowsky et al. (1980), energy scale corrected by +0.3 eV according to the calibration of Dressler and Allan (1987).

The integral elastic CS of Khakoo et al. (1993) shows the same value of $20.3 \cdot 10^{-20}$ m² at 5 and 10 eV, then it descends monotonically with energy. This fall is faster than the one predicted by the low-energy calculation of Gianturco and Stoecklin (1994). The integral elastic CS obtained from the DCS of Fink et al. (1975) in the present work ($\pm 15\%$ integration and extrapolation uncertainty) are compatible at 100 eV and 200 eV with the calculations of Lee et al. (1990), not shown in the figure, and with those of Jain and Baluja (1982) within 10%. Between 400 eV and 600 eV the integral values obtained from the DCS of Fink et al. fall rapidly with energy, faster than the 1/E dependence. We suppose some error in their normalization to the independent-atoms model (Fink et al. 1975).

Low-energy calculations (Jain 1993) predict a Ramsauer minimum at about 0.15 eV. Measurements of the 90° elastic DCS at low energies (Kochem *et al.* 1985) show a fall up to the energy of 0.1 eV. Probably the elastic CS rises again at energies lower than 0.1 eV, as is indicated by the transmission experiment of Dressler and Allan (1987), see fig. 1. At 2 eV the DCS in C_2H_2 (Kochem *et al.*) resembles that in N_2 at 2.2 eV, see

fig. 23 in part I; this similarity could indicate scattering via a resonant state.

Swarm measurements. Drift velocities at 300–370 K were measured by Bowman and Gordon (1967). According to this work the momentum transfer CS falls from $100 \cdot 10^{-20}$ m² at 0.02 eV to about $20 \cdot 10^{-20}$ m² at 0.06 eV. This result disagrees with the analysis of Duncan and Walker (1972) who derived momentum transfer CS rising monotonically from $9.5 \cdot 10^{-20}$ m² at 0.01 eV to about $2.5 \cdot 10^{-20}$ m² at 1.0 eV.

Vibrational. Five fundamental vibrational modes are possible for acetylene. Due to the close spacing between vibrational levels, the experiments (Kochem et al. 1985, Khakoo et al. 1993) could not resolve the ν_1 (symmetric C-H stretching, 418 meV energy loss) from the ν_3 (asymmetric stretching, 409 meV) modes and the ν_4 (symmetric bending, 73 meV) from the ν_5 (asymmetric bending, 91 meV) modes. At 0.6 eV collision energy and 10° scattering angle, the excitation of the infrared-active ν_3 , ($\nu_4 + \nu_5$) and ν_5 modes only was observed (Kochem et al. 1985). For the ν_5 mode, the excitation functions in the near-threshold region can be well described by the Born approximation. At 2.6 eV the Born approximation reproduces well the low-angle ν_5 DCS but underestimates its values above 30°. For the $\nu_3 + \nu_1$ modes the Born approximation underestimates the experimental DCS even at 1.6 eV. The maximum of the low-energy vibrational CS as presently calculated (see eq. (15) in part I) from the Born approximation for ν_3 and ν_5 modes is rather high, $4.7 \cdot 10^{-20}$ m²; its position coincides with the minimum in the elastic CS as calculated by Jain (1993).

At 2.6 eV the DCS for the $\nu_3 + \nu_1$ modes and for the ν_2 (symmetric C \equiv C stretching, 245 meV) mode show a d-wave–like shape (Kochem et~al. 1985), similar to the vibrational excitation DCS in N₂ at 2.1 eV, see fig. 24 in part I. At 2.6 eV the branching ratio for the five modes (from ν_1 to ν_5) is 1:6:2:1:8. A 25% underestimation of the experimental absolute values (Kochem et~al. 1985), as indicated by the comparison with the Born approximation, could be caused by the normalization procedure. To take into account this possible underestimation, in fig. 1 we present the integral CS of Kochem et~al. multiplied by a factor of 1.25.

An enhancement of the ν_2 symmetric stretch mode with some quanta of the ν_4 bending mode was observed at 1.8 eV (Andrić and Hall 1988). It can be explained in terms of scattering via the ${}^2\Pi_{\rm g}$ shape resonance. More overtones were observed at 1.8 eV than in the maximum of the resonance at 2.5 eV. Measurements of Andrić and Hall (1988) and of Khakoo *et al.* (1993) also evidenced a rise of the $\nu_1 + \nu_3$ vibrational CS at 6 eV, pointing out to the existence of another resonant state at this energy.

Electronic excitation at 25 eV and 35 eV, $10^{\circ}-80^{\circ}$ scattering angles range was studied by Trajmar et al. (1968, 1970). DCS for the \tilde{C} $^{1}\Pi_{\rm u}$ (8.16 eV energy loss for $\nu_{2}'=0$) and the \tilde{D} (9.26 eV energy loss for $\nu_{2}'=0$) states are both of similar intensity at 25 eV and 35 eV, with the \tilde{C} state being sightly more forward-centered. DCS for the broad \tilde{B} , 7.2 eV energy loss band are by one order of magnitude smaller. DCS for \tilde{a} and \tilde{b} (5.2 eV and 6.1 eV energy loss, respectively) triplet states rise between 10° and 80° scattering angles. The relative values for all these states rise by a factor of two, roughly from 25 eV to 35 eV, apart from the \tilde{a} -state DCS which rises by a factor of four (Trajmar et al. 1970).

Lassettre et al. (1968) reported energy loss spectra at 40 eV and 50 eV collision energy, 0° and 10° scattering angles. Energy loss spectra in the region of valence electronic excitations were examined also by Dance and Walker (1973), van Veen and Plantenga (1976), Wilden et al. (1977, 1980), Hammond et al. (1987), Dressler and Allan (1987) and de Souza and de Souza (1988); in the region of the core-excitation by Tronc et al. (1979). Assignments of observed states were done by Dressler and Allan (1987). Forward-angle electron scattering was studied by Cooper et al. (1988, 1995) for the 6–180 eV energy

loss. Photoabsorption CS below the ionization threshold reaches a maximum value of $0.33\cdot 10^{-22}$ m² at 9.5 eV (Cooper *et al.* 1988).

Optical emission from dissociated fragments (40–200 nm wavelength range) was measured by Pang et al. (1987) at 20–400 eV collision energy. The atomic hydrogen Lyman- α emission intensity at 100 eV in C₂H₂ is 30% lower than in CH₄. Emission spectra in the 415–440 nm range due to the $A^{2}\Delta \rightarrow X^{2}\Pi$ de-excitation of the CH radical at 17–100 eV impact energies were studied by Tokeshi et al. (1996).

Ionization. An excellent agreement exists between the early (Tate and Smith 1932, Gaudin and Hagemann 1967) and recent (Zheng and Srivastava 1996, Tian and Vidal 1998, Kurepa et al. 1991, Djurić et al. 1996) ionization CS measurements. The semiempirical model (Hwang et al. 1996) reproduces well the experimental data. In fig. 1 we report the data of Tate and Smith (1932) and of Zheng and Srivastava (1996) only.

The ionization CS in C_2H_2 at 100 eV is by as much as 70% higher than the ionization CS for the isolectronic molecules CO and N_2 (Tate and Smith 1932). This difference still amounts to 40% at 750 eV. Dissociative ionization in C_2H_2 is rather small at 100 eV: 67% of the ionization yield is constituted by parent $C_2H_2^+$ ions. The C_2H^+ , H^+ , C_2H^+ , and C_2H^+ and C_2H^+ in the ionization of the ionization of the ionization of the ionization of the ionization yield is constituted by parent $C_2H_2^+$ ions. The $C_2H_2^+$, $C_2H_2^+$ and $C_2H_2^+$ in the ionization in C_2H_2 is rather small at 100 eV: 67% of the ionization yield is 2.5%, 5.5%, 4.0%, 1.3%, respectively (Zheng and Srivastava 1996, Tate et al. 1935).

Dissociative attachment has been studied in several works (see Dressler and Allan 1987 and references therein). A large maximum for the C_2H^- yield occurs at 2.9 eV (Dressler and Allan 1987). At 8 eV another broad C_2H^- band and several ovarlapping C_2^- bands with vibrational structures are visible (Dressler and Allan 1987). Rutkowsky et al. (1980) reported $6-7 \cdot 10^{-24}$ m² peak values both at 2.5 and 7.5–8 eV energies. The position of the first peak in the paper of Rutkowsky et al. seems to be shifted by -0.3 eV, compared to the value of Dressler and Allan (1987). As these latter authors performed a detailed calibration of the energy scale using the O^- peak formation from CO at 9.62 eV, in fig. 1 we have shifted the spectrum of Rutkowsky et al. (1980) by +0.3 eV.

Resonances. A shape resonance of the ${}^2\Pi_{\rm g}$ (π^*) type at 2.6 eV has been observed in the trapped electron measurements of Dance and Walker (1973), van Veen and Plantenga (1976), in the dissociative attachment CS (Dressler and Allan 1987) and in the vibrational excitation (Kochem *et al.* 1985, Andrić and Hall 1988). In TCS this resonance manifests itself as a structureless peak, similar to the one observed in CO, see fig. 25 in part I.

At 6 eV a core excited-resonance, analogous to that at 8 eV in N_2 , of the Σ_g symmetry and due to the $s\sigma$ partial-wave scattering has been seen in the vibrational CS (Andrić and Hall 1988). Dressler and Allan (1987) observed four additional Feshbach resonances in the dissociative attachment CS between 7.5 eV and 9.5 eV. These states were also observed in the transmission spectra (van Veen and Plantenga 1976) and in threshold electrons spectra (Hammond *et al.* 1987). The configurations of these states have been discussed by Andrić and Hall (1988).

Sum check indicates a rather good consistency between the elastic cross-sections of Khakoo et al. (1993) up to 100 eV, the ionization data of Tate and Smith (1932) and the total values of Sueoka and Mori (1989), see Table I. The difference (20%) between the total and summed values in the 15 eV-50 eV energy range is probably due to the electronic excitation part. At 400 eV the sum of Fink's et al. elastic and Tate and Smith's ionization CS is 10% higher than the TCS of Xing et al. (1995); at 1000 eV this sum is 40% lower. This confirms the possibility of errors in the normalization procedure of elastic CS of Fink et al. (1975).

Table I. – Integral cross-sections for electron scattering on acetylene (in 10^{-20} m² units).

Energy	Elastic	Vibrational	Ionization	Summed	Total
5.0	20.3 K	2.1 K		22.4	23.7 S
10.0	20.3 K	1.2 K		21.5	22.1 S
15.0	15.1 K		0.75 T	15.9	18.9 S
20	12.8 K		1.8 T	14.6	17.7 S
30	8.5 K		3.4 T	11.9	14.9 S
50	5.6 K		4.7 T	10.3	12.5 S
100	3.7 K		5.0 T	8.7	9.3 S
	11.3 F				
200	6.1 F		4.0 T	10.1	6.9 S
400	3.4 F		2.6 T	6.0	4.9 S
					5.38 X
600	1.0 F		2.0 T	3.0	4.06 X
1000	0.68 F		1.18 H	1.86	2.61 X

- ${\sf F}$ Fink et~al.~(1975) presently integrated (±15% error)
- H Hwang et al. (1996) semiempirical values
- K Khakoo et al. (1993)
- S Sueoka and Mori (1989)
- T Tate and Smith (1932)
- X Xing et al. (1995)

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2[.]2. Benzene (C_6H_6). – Benzene molecule has been the subject of intensive studies by electron transmission spectroscopy. However, only few CS measurements were reported. Integral cross-sections for C_6H_6 are shown in fig. 2.

Total cross-sections were measured by Holst and Holtsmark (1931) at collision energies between 0.5 eV and 25 eV using a low-energy Ramsauer technique. Sueoka (1988) has determined normalized TCS between 1 eV and 400 eV in a linear-transmission apparatus with longitudinal magnetic field. Możejko et al. (1996) reported absolute TCS in a linear apparatus with cylindrical spectrometer (Gdansk laboratory) at 0.6–250 eV and in a modified Ramsauer set-up (Trento laboratory) at 90–3500 eV. Very low-energy absolute TCS between 35 meV and 2 eV in C_6H_6 , C_6H_5D and C_6D_6 were studied by Gulley et al. (1998); using a strong axial magnetic field they also measured the "backward" scattering CS down to 10 meV. Gulley et al. used two apparatuses, both of them based on a synchrotron-radiation electron source, at Daresbury (UK) and Århus (Denmark). As stated by Gulley et al. (1998) only negligible differences appeared between the spectra from the two laboratories.

TCS from the two low-energy experiments (Sueoka 1988, Możejko *et al.* 1996) agree well in shape. The data of Sueoka are somewhat lower, probably due to the normalization

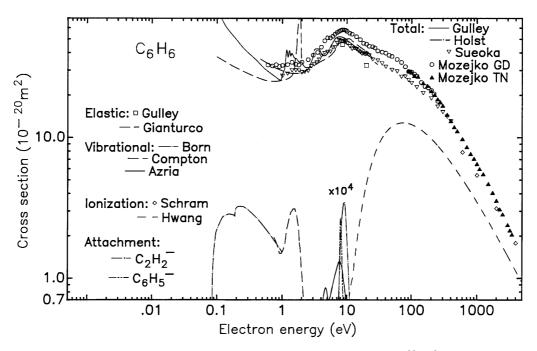


Fig. 2. – Integral cross-sections for electron scattering on C_6H_6 (in 10^{-20} m² units). Total: Możejko GD, low-energy data (Gdansk laboratory) from Możejko et al. (1996); Możejko TN, intermediate-energy data (Trento laboratory) from Możejko et al. (1996), absolute; Sueoka (1988), normalized; Holst and Holtsmark (1931), absolute; Gulley et al. (1998), "total scattering", absolute. Elastic: Gulley and Buckman (1999a), absolute; Gianturco and Lucchese (1998), calculation. Vibrational: Born approximation (eq. (15) in part I); Compton et al. (1968) and Azria and Schulz (1975) are shown only for illustration purposes, see text. Ionization: Schram et al. (1966); Hwang et al. (1999), semiempirical. Dissociative attachment: Fenzlaff and Illenberger (1984), $C_2H_2^-$ and $C_6H_5^-$ yields in arbitrary units, see text.

procedure. The TCS in C_6H_6 exhibits a broad maximum, as in alkanes, centered near 8.5 eV and reaching a value of $58 \cdot 10^{-20}$ m² (Możejko et al. 1996). On the low-energy side of this maximum a shoulder structure at about 4–5 eV was seen both by Sueoka (1988) and Możejko et al. (1996). The data of Holst and Holtsmark do not show this shoulder but exhibit a well-distinct maximum centered between 3 and 4 eV. Electron transmission experiments (Sanche and Schulz 1973, Mathur and Hasted 1976, Burrow et al. 1987) indicated the presence of a ${}^2B_{2g}$ resonant state at 4.82 eV (Burrow et al. 1987).

At 1.4–1.5 eV another, weak maximum was seen in TCS by Możejko *et al.* (1996) and by Sueoka (1998). Gulley *et al.* (1998) observed, instead, a weakly modulated, vibrational-like structure in TCS, starting at 1.17 eV and with 0.115 eV spacing. Numerous transmission-current measurements (see below) localized an ${}^{2}E_{2u}$ resonance in this energy range.

Swarm experiments. Christophorou et al. (1966) measured drift velocities relative to C_2H_4 at thermal energies. Approximating the momentum transfer CS by an $E^{-1/2}$ functional dependence they derived a momentum transfer CS a factor of nine higher in C_6H_6 than in C_3H_8 . High values of CS (50 · 10⁻²⁰ m² at 0.1 eV) were confirmed also by Mechlińska-Drewko (1980) in her D_T / μ measurements.

Studies of Rydberg atoms quenching in C_6H_6 were performed by Uematsu *et al.* (1984) and Popple *et al.* (1993). In experiments with potassium atoms prepared in intermediate ($n \cong 30$, where n is the principal quantum number) Rydberg states very low ($\cong 3 \cdot 10^{-8} \text{ cm}^3 \text{ s}^{-1}$) rate constants were observed, for collisions leading to state-changing of potassium (Popple *et al.* 1993). Instead, high rates were observed for formation of associated $KC_6H_6^+$ ions at low collisional velocities. This indicates an effective transfer of collisional energy to internal energy of the $KC_6H_6^+$ complex. Such a complex may be subsequently stabilized by energy transfer to the Rydberg electron.

Elastic cross-sections at 8.5, 10 and 20 eV have been reported by Gulley and Buckman (1999a,b). The integral elastic CS at 8.5 eV is 50% higher than at 20 eV. The DCS at energy of the TCS maximum (8.5 eV) shows a rather unusual angular shape, with an almost flat minimum between $60^{\circ}-120^{\circ}$ and a quick rise below 60° (see fig. 6 subsect. 2.3). At 20 eV the DCS shows a more visible d-wave like pattern. Experimental DCS at 8–20 eV (Gulley and Buckman 1999a,b) agree in shape with the calculations of Gianturco and Lucchese (1998), which indicate the trapping of an incoming electron by the l=6 partial-wave potential barrier. Both the experiment (Allan 1989) and the theory (Gianturco and Lucchese, 1998) pointed out the existence of a broad shape resonance in the region of elastic (and total) CS maximum (8.5 eV), due to the capture of an incoming electron to the $^{2}(e_{1u}\sigma^{*})$ molecular orbital.

Calculated integral elastic CS (Gianturco and Lucchese 1998) follow very well the shape of TCS as measured by Sueoka (1988). However, the peak calculated at 1.82 eV is much narrower than the weak maximum seen in TCS measurements (Sueoka 1988, Możejko et al. 1996). This is probably due to neglecting the nuclear motion in the calculations. (The same effect is visible, for example, in fixed-nuclei calculations for CS₂, see fig. 21 in part II.) Both the 1.82 eV peak and the broad maximum calculated at about 10 eV (Gianturco and Lucchese 1998) are shifted to slightly higher energies with respect to the features observed in experimental TCS (Sueoka 1988, Możejko et al. 1996).

Vibrational excitation. Benzene possesses 30 normal vibrational modes but only 20 fundamental frequencies. Herzberg's labelling for the modes, used for example by Wong and Schulz (1975), Cheung and Bishop (1982), differs from Wilson's notation, currently in use (see, for example, Iachello and Oss 1993). We use below Herzberg's labelling, giving reference to the symmetry group. Among the possible vibrational modes only few

are efficiently excited in electron collisions (Larkin and Hasted 1972, Wong and Schulz 1975). At 1.18 eV collision energy, *i.e.* in the region of the first shape resonance, the modes involving C-C bond are excited (Wong and Schulz 1975): the ν_2 (C-C symmetric stretch, A_{1g} symmetry), the ν_{16} (C-C in-plane bending, E_{2g}) and the ν_{20} (C-C out-of-plane bending, E_{2u}) modes and their harmonics. The DCS measured at 25°–100° for the ν_2 mode in this energy region show an angular distribution characteristic for the β 0 partial-wave scattering, with a clear minimum near 40° and the second minimum at 90° somewhat washed out (Wong and Schulz 1975).

DCS for the infrared active mode ν_4 (C-H out-of-plane bending, $A_{2\mathrm{u}}$) descend with angle and show a strong peak in the forward direction; the excitation decreases monotonically with energy without structures in the resonance regions (Wong and Schulz 1975). In the region of the 4.8 eV resonance, only the excitations of ν_7 (C-H out-of-plane bending, $B_{2\mathrm{g}}$ symmetry group) and ν_1 (C-H symmetric stretch, $A_{1\mathrm{g}}$) modes are enhanced (Wong and Schulz 1975). The energy dependence of the DCS at 60° for the ν_7 mode show another large maximum centered at 8.0 eV (Azria and Schulz 1975).

We are not aware of absolute measurements of integral CS for vibrational excitation in benzene. In order to give a very approximate picture of the vibrational excitation CS, in fig. 2 we have reported: the vibrational excitation for the infrared active (ν_4 , ν_{12} , ν_{13} , ν_{14}) modes as evaluated with the Born approximation (eq. (15) in part I) with transient dipole moments from Bishop and Cheung (1982); the threshold-electron spectrum of Compton et al. (1968) normalized at 1 eV to Born evaluation; the ν_1 DCS at 60° of Azria and Schulz (1975) normalized to the Compton et al.'s data at 3.5 eV.

Electronic excitation. We are not aware of CS for electronic excitation in C_6H_6 . Energy loss spectra were collected at 300 eV collision energy by Skerbele and Lassettre (1965), at 40–100 eV by Lassettre et al. (1968), at 13.6 and 20 eV by Doering (1969), at 12.7–40 eV by Doering (1977), at 25–75 eV by Frueholz et al. (1977, 1979), at 10 and 100 eV by Wilden and Comer (1980). In particular, the first three triplet states ${}^3B_{1u}$, ${}^3B_{2u}$ with energy loss in the 3.53–5.79 eV were studied by Wilden and Comer (1980) with energy resolution allowing to separate vibronic modes. The oscillator strength for the 1 ${}^1B_{2u} \rightarrow \tilde{X}^1A_{1g}$ transition (4.79 eV excitation threshold, see Frueholz et al., 1979) at 400 and 500 eV collision energy was measured by Klump and Lassettre (1977).

Spectra of trapped electrons in the electronic excitation energy region were collected by the SF₆ scavenger technique (Compton *et al.* 1968, Hubin-Franskin and Collin 1970) and the retarding-field method (Brongersma *et al.* 1969, van Veen and Plantenga 1975). Brongersma *et al.* (1969) at 1.6 eV above the excitation thresholds, reported CS values of $1.5 \cdot 10^{-20}$ m² and $3.2 \cdot 10^{-20}$ m² for the 4.7 eV ($^{1}B_{2u} + ^{3}E_{1u}$ excitations) and 6.9 eV ($^{1}E_{1u}$ state) energy loss processes, respectively. Electron momentum spectroscopy for valence shells was performed by Samardzic *et al.* (1993).

Optical emission. Low-resolution (10 nm) fluorescence excitation functions in the 250–400 nm wavelength range the metastables yield and the optical emission spectra in the 220–520 nm wavelength range were reported for 2–300 eV collision energies by Smyth et al. (1974a, b, c, respectively). The fluorescence spectrum for the symmetry forbidden vibronic levels of the $^{1}B_{2u}$ state was studied by Kawazumi et al. (1984). Emission in the 185–900 nm wavelength region from H, C and CH fragments was studied by Beenakker and de Heer (1974). Cross-sections at 100 eV collision energy for the emission from dissociated fragments (Balmer H_{β} line and the CH radical $A^{2}\Delta \rightarrow X^{2}\Pi$ transition) amount to 0.68 \cdot 10⁻²² m² and 0.34 \cdot 10⁻²² m², respectively (Beenakker and de Heer 1974). The CS for the emission from the $C_{6}H_{6}$ transition $^{1}B_{2u} \rightarrow X^{1}A_{1g}$ exhibits a

threshold peak (Beenakker *et al.* 1974b) and amounts to $0.21 \cdot 10^{-22}$ m² at 100 eV (Beenakker *et al.* 1974a).

Ionization. We are only aware of measurements of Schram et al. (1966) in the 0.6–12 keV energy range. These data are significantly higher than the results of a semiempirical model of Hwang et al. (1996). According to the two sets of data, the ionization CS at 1000 eV would contribute for 54% (Hwang et al. 1996) or for 83% (Schram et al. 1966) to the TCS. As a comparison we recall that the ionization CS in methane (Nishimura and Tawara 1994) is 63% of the TCS (Zecca et al. 1992) at 1000 eV.

Dissociative attachment. There are few measurements in C_6H_6 . Azria and Schulz (1975) reported a large dissociative attachment peak of 2.2 eV FWHM centered at 8.0 eV, with a value of about $2.6 \cdot 10^{-24}$ m². This peak resembles the dissociative attachment in CH_4 , see fig. 1 in part II. Fenzlaff and Illenberger (1984) measured relative yields for the formation of the $C_2H_2^-$, centered at 8 eV and for the $C_6H_5^-$, centered at about 9 eV. In fig. 2 we have normalized the $C_2H_2^-$ peak of Fenzlaff and Illenberger to the value of Azria and Schulz at 8.0 eV. The attachment of low-energy (< 1 eV) electrons in C_6H_6 mixtures with N_2 and Ar was studied by Christophorou and Goans (1974).

Resonances. The C₆H₆ molecule exhibits an interesting pattern of resonant states. Allan (1989) observed three resonant enhancements in the excitation functions for the $2\nu_1$ and $3\nu_2$ vibrational modes: at about 1.2, 4.8 and 8 eV (see also Ben Arfa and Tronc 1990). The lowest, $C_6H_6^-$ (${}^2E_{2u}$) state was seen in electron transmission spectra (Boness et al. 1967, Larkin and Hasted 1972, Sanche and Schulz 1973, Nenner and Schulz 1975, Mathur and Hasted 1976, Burrow et al. 1976, Fesenko and Iogansen 1977, Frazier et al. 1978, Burrow et al. 1987) and in threshold electrons spectra (Compton et al. 1968). The vibrational structure observed in transmitted current, with an offset at 1.15 eV and the first spacing of 123 meV, resembles that for excitation of the totally symmetric breathing vibrational mode of the C₆ ring in C₆H₆ molecule (Sanche and Schulz 1973, Mathur and Hasted 1976). A similar vibrational structure has also been observed in the recent TCS measurements performed with 8 meV energy resolution (Gulley et al. 1998). Calculations (Gallup 1986, Gianturco and Lucchese 1998) indicate that the 1.1 eV resonance occurs via capture of one electron (l=3 partial wave in the entrance and l=0at the exit channel) into the doubly degenerate $c_{2u}(\pi^*)$ antibonding orbital located over the C-C bonds.

The next, $C_6H_6^-$ ($^2B_{2g}$) short-lived, temporary negative ion state manifests itself in electron transmission measurements as a structureless minimum (Sanche and Schulz 1973, Nenner and Schulz 1975, Mathur and Hasted 1976, Burrow et al. 1976), centered at 4.8–4.9 eV. This state was also observed in threshold electrons spectra (Compton et al. 1868, van Veen and Plantenga 1975). Trapping of one electron (the l=4 incoming partial wave) to the π^* -orbital has been proposed as a mechanism for the formation of this temporary negative-ion state (Gallup 1986, Gianturco and Lucchese 1998). As stated by Gallup (1986), since the l=4 partial wave peaks far from the center of the molecule, the C-H (ν_7) rather than C-C (ν_1) deformations are responsible for the vibrational spectrum (Wong and Schulz 1975).

A short-lived resonance was observed in transmission spectra (Mathur and Hasted 1976) at about the energy of the TCS maximum. This resonance gives rise to an unusual shape of DCS at 8.5 eV (Gulley and Buckman 1999a) compared to other hydrocarbons. It was discussed theoretically by (Gianturco and Lucchese 1998). A broad enhancement of the ν_1 vibrational mode was observed close to this energy (Azria and Schulz 1975).

A number of additional resonant states was observed in transmission spectra on solid benzene (Sanche 1979); only a few of these structures were confirmed in thresholdexcitation experiments (Compton et al. 1968, van Veen and Plantenga 1975). K-shell resonances in C_6H_6 were observed by inner-shell energy loss spectra (Horsley et al. 1985).

Sum check. A reasonable agreement exists between the elastic CS of Gulley and Buckman (1999a) and total CS of Możejko et al. (1996). At 8.5 eV the TCS is higher than the elastic integral CS by $12 \cdot 10^{-20}$ m². Brongersma et al. (1969) obtained an electron excitation CS for three states of about $5 \cdot 10^{-20}$ m² at 8.5 eV. A part of the remaining difference between TCS and (elastic + electronic) excitation CS comes from the vibrational excitation, enhanced in this energy region. At 20 eV the sum of elastic (Gulley and Buckman 1999a) and ionization (Hwang et al. 1996) CS is $37 \cdot 10^{-20}$ m² vs. TCS determinations of $45.9 \cdot 10^{-20}$ m² by Możejko et al. (1996) and $39.3 \cdot 10^{-20}$ m² by Sueoka (1988).

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2.3. Linear hydrocarbons (C₂H₄, C₂H₆, C₃H₈). – In many cases cross-sections for several hydrocarbons were presented in the same paper; therefore also in the present work we will group them together. Table II contains a list of significant experiments in which integral CS for electron scattering on hydrocarbons (excluding those performed only for CH₄, C₂H₂ or C₆H₆) have been measured. Integral CS for ethene C₂H₄, ethane C₂H₆, propane C₃H₈ are shown in figs. 3, 4 and 5, respectively. As far as possible, we will discuss also the results on other hydrocarbons in this chapter. A set of total and partial CS for C₂H₆ was given, among others, by Shishikura *et al.* (1997) and for several hydrocarbons by Hayashi (1992).

Total cross-sections. TCS were measured by Brüche in 1929-1930 in a Ramsauer-like apparatus covering the 0.2–50 eV energy range; by Sueoka and Mori (1986) both for electrons and positrons in a linear configuration using a longitudinal magnetic field; by Floeder et al. (1985) in a linear set-up with a weak longitudinal magnetic field; by Szmytkowski and Krzysztofowicz (1995) in an electrostatic spectrometer; by Nishimura and Tawara (1991) in a linear electrostatic set-up; by Lunt et al. (1994) in two experiments using synchrotron radiation as electron source. TCS of Nishimura and Tawara (1991) are about 5% systematically higher than the measurements of Floeder et al. (1985),

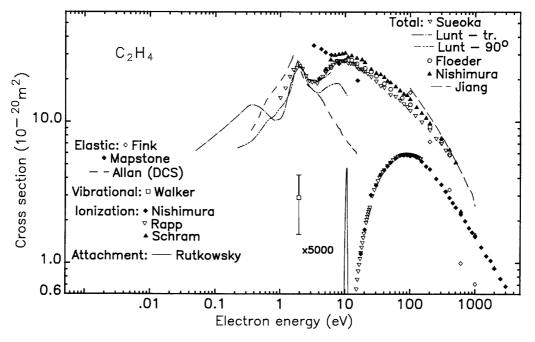


Fig. 3. – Integral cross-sections for electron scattering on ethene (C_2H_4). Total: Floeder et al. (1985) absolute; Nishimura and Tawara (1991) absolute; Sueoka and Mori (1986) normalized; Lunt et al. (1994), relative scattering into $90^{\circ} \pm 30^{\circ}$ and the transmission experiment, presently normalized to Sueoka and Mori at 2.0 eV; Jiang et al. (1995), optical model, shown only above 100 eV for clarity. Elastic: Mapstone and Newell (1992), integrated in the present work; Fink et al. (1975) normalized to independent-atoms theory, integrated in the present work; Allan (1994), 90° DCS multiplied by 4π . Vibrational: absolute of Walker et al. (1978) at 1.95 eV. Ionization: Nishimura and Tawara (1994); Rapp and Englander-Golden (1965); Schram et al. (1966). Dissociative attachment: Rutkowsky et al. (1980).

Table II. – Important measurements of integral cross-sections for electron scattering on hydrocarbons (works on only $\mathrm{CH_4},\,\mathrm{C_2H_2}$ and $\mathrm{C_6H_6}$ are not included). Note that only the first authors are reported.

e (1929) e (1930a) e (1930b) r (1985) nura (1991) kowski (1995) i (1986)	C ₂ H ₂ , C ₂ H ₄ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , n-C ₄ H ₁₀ n-C ₄ H ₁₀ , iso-C ₄ H ₁₀ CH ₄ , C ₂ H ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ , cyclo-C ₃ H ₆ , n-C ₄ H ₁₀ , iso-C ₄ H ₁₀ , 1-C ₄ H ₈ C ₂ H ₄ , C ₂ H ₆ , C ₃ H ₆ , cyclo-C ₃ H ₆ , C ₃ H ₈ C ₂ H ₆	0.8–50 0.8–50 0.2–50 5–400 4–500
kowski (1995) ı (1986)	iso- C_4H_{10} , 1- C_4H_8 C_2H_4 , C_2H_6 , C_3H_6 , cyclo- C_3H_6 , C_3H_8	
1998) a (1999)	CH ₄ , C ₂ H ₄ , C ₂ H ₆ CH ₄ , C ₂ H ₄ , C ₂ H ₆ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ C ₃ H ₈ C ₃ H ₆ , C ₅ H ₁₀ , C ₆ H ₁₂ , C ₈ H ₁₆	0.6–250 1–400 0.05–11 0.01–0.18 0.8–600 0.7–600
	C_2H_2, C_2H_4, C_2H_6 C_2H_6	$100 – 1000 \\ 2 – 100$
n (1994), a (1999) (1985) one (1992) (1998)	$\begin{array}{c} C_{3}H_{8} \\ CH_{4},\ C_{2}H_{6} \\ CH_{4},\ C_{2}H_{4},\ C_{2}H_{6} \\ C_{2}H_{6} \\ CH_{4},\ C_{2}H_{4},\ C_{2}H_{6} \end{array}$	$\begin{array}{c} 2100 \\ 7.520 \\ 315 \\ 0.410 \\ 3001300 \end{array}$
n (1972) n (1974) kle (1978)	CH ₄ , C ₂ H ₂ , C ₂ H ₄ , C ₂ H ₆ , C ₃ H ₆ , 1-C ₄ H ₈ C ₂ H ₂ , C ₂ H ₄ C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ , cyclo-C ₃ H ₆ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₄ H ₁₀ , n-C ₅ H ₁₂ C ₃ H ₆ , cyclo-C ₃ H ₆ , C ₃ H ₈ ,	0.01-0.06 0.01-1 0.01-1 0.02-0.3
no (1986) dt (1992)	$n\text{-}C_4H_{10}$, iso- C_4H_{10} cyclo- C_6H_{12} , cis- $C_{10}H_{18}$, trans- $C_{10}H_{18}$ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , $n\text{-}C_4H_{10}$, $n\text{-}C_5H_{12}$, iso- C_4H_{10} , neo- C_5H_{12} , $n\text{-}C_8H_{18}$, $n\text{-}C_{10}H_{22}$ CH ₄ , C ₂ H ₄ , C ₂ H ₆ , C ₃ H ₈ , cyclo- C_3H_6 , C ₃ H ₆	0.01–0.3 0.01–0.5
(1985) one (2000)	C_2H_6 CH_4 , C_2H_6 C_2H_4 , C_2H_6 C_3H_8	3-20 7.5-20 3.2-15.4 7.5
rdt (1965) tre (1968a) tre (1968b) on (1979)	CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₂ H ₄ , cyclo-C ₆ H ₁₂ CH ₄ , C ₃ H ₈ , n-C ₇ H ₁₆ C ₂ H ₂ , C ₂ H ₄ , C ₂ H ₆ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₄ H ₁₀ CH ₄ , C ₂ H ₆ , n-C ₅ H ₁₂ C ₂ H ₂ , C ₂ H ₄ , C ₃ H ₃ ,	70 33–100 50–180 30–250
	C_4H_5 (1-butyn, 1,3-butadien), C_6H_6 CH_4 , C_2H_6 , C_3H_8 , C_4H_{10} , C_5H_{12} ,	20–90 tr–200*
	a (1986) (1994) (1998) a (1999) (1975) a (1988) en (1994), a (1999) (1985) one (1992) (1998) 1998) an (1967) an (1972) an (1974) rkle (1978) 1998) dt (1978) 1998) dt (1992) kura (1997) en (1990) (1985) one (2000) en (1994) tre (1964) rdt (1965) tre (1968a) tre (1968b) on (1979) ermann (1979)	1994) CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ CH ₄ , C ₂ H ₆ , C ₃ H ₈ , C ₃ H ₆ C ₃ H ₈ C ₃ H ₆ , C ₅ H ₁₀ , C ₆ H ₁₂ , C ₈ H ₁₆ C ₂ H ₆ C ₂ H ₄ , C ₂ H ₆ C ₂ H ₄ , C ₂ H ₆ C ₂ H ₆ C ₃ H ₈ C ₂ H ₆ C ₄ H ₆

Table II. – Continued.

Process	Authors	Gases	Energies (eV)	
Optical Beenakker (1975) emission Donohue (1977) Möhlmann (1977) Tokeshi (1996)		$\begin{array}{c} CH_4,\ C_2H_2,\ C_2H_4,\ C_2H_6\\ CH_4,\ C_2H_4,\ CH_3OH\\ CH_4,\ \dots,\ C_8H_{18}\\ C_2H_2,\ C_2H_4,\ C_2H_6 \end{array}$	100-5000 0-2000 0-2000 17-100	
Ionization	Schram (1966)	$\begin{array}{c} CH_4,\ C_2H_4,\ C_2H_6,\ C_3H_6,\\ cyclo-C_3H_6,\ C_3H_8,\ 1,3\text{-}C_4H_6,\ 1\text{-}C_4H_8,\\ trans-2\text{-}C_4H_8,\ cis-2\text{-}C_4H_8,\ n\text{-}C_4H_{10},\\ iso-C_4H_{10},\ n\text{-}C_5H_{12},\ neo\text{-}C_5H_{12},\\ iso-C_5H_{12},\ n\text{-}C_6H_{14},\ C_6H_6 \end{array}$	600–12000	
	Rapp (1965) Nishimura (1994)	C ₂ H ₄ CH ₄ , C ₂ H ₄ , C ₂ H ₆ , C ₃ H ₆ , cyclo-C ₃ H ₆ , C ₃ H ₈	tr-145 tr-3000	
	Chatham (1984) Duric (1991) Grill (1993a) Grill (1993b)	CH ₄ , C ₂ H ₆ , SiH ₄ , Si ₂ H ₆ CH ₄ , C ₂ H ₃ , C ₃ H ₈ C ₂ H ₆ C ₃ H ₈	15–400 tr–240 tr–950 tr–950	
Attachment	Rutkowsky (1980)	$CH_4, C_2H_6, C_3H_8, C_4H_{10}, C_2H_4, \\ C_3H_6, iso-, trans-, 1-C_4H_8, \\ 1,3-butadien, C_2H_2, C_3H_4, 1-C_4H_6$		

^{*} Energy loss tr - Threshold

for all the hydrocarbons studied in common.

For C_2H_4 and C_2H_6 the TCS of Sueoka and Mori (1986) are 5% lower than the data of Floeder et~al. in the whole energy range of overlap (4–400 eV). For C_2H_6 the results of Szmytkowski and Krzysztofowicz (1995) are placed between the data of Nishimura and Tawara (1991) and of Floeder et~al. (1985). Note that the measurements of Nishimura and Tawara (1991) were performed with the best (extrapolated to zero) angular resolution. For C_3H_8 Sueoka and collaborators (Tanaka et~al. 1999) have corrected the measured TCS for the angular resolution error. The data of Tanaka et~al. agree very well in shape with other experiments (Nishimura and Tawara 1991, Floeder et~al. 1985) but are somewhat higher (10% at 8 eV). This difference is opposite in sign with respect to that for C_2H_4 and C_2H_6 for which no correction was done (Sueoka and Mori 1986).

TCS for C_2H_4 , C_2H_6 (and CH_4) at very low energies are made by Lunt et~al.~(1994) who performed measurements of scattering with $90^{\circ} \pm 30^{\circ}$ angles (Daresbury laboratory) and of the transmitted current (Orsay laboratory). To get absolute values of TCS, we have normalized the Orsay data to the measurements of Sueoka and Mori (1986) at 1 eV for C_2H_6 and at 2 eV for C_2H_4 . According to Lunt et~al., the TCS in C_2H_6 exhibits a Ramsauer minimum at about 0.6 eV and rises monotonically up to 6 eV (for clarity only a low-energy part of data is presented in fig. 4). For C_2H_4 no such deep minimum was seen but an additional maximum at 2.0 eV attributed to the $^2B_{2g}$ resonant state was observed (Lunt et~al.~1994, Sueoka and Mori 1986). A structure is also visible at about 0.4 eV in the transmission experiment (Orsay laboratory) but it is probably an experimental artifact, as it has not been observed in the measurements performed in the Daresbury laboratory (Lunt et~al.~1994). In fig. 3 we present both the Orsay and

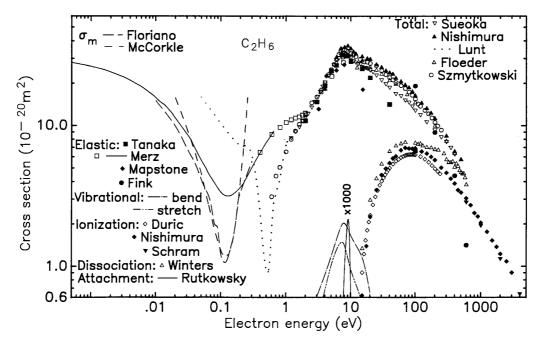


Fig. 4. – Integral cross-sections for electron scattering on ethane (C₂H₆). Total: Floeder et al. (1985) absolute; Nishimura and Tawara (1991) absolute; Sueoka and Mori (1986) normalized; Lunt et al. (1994) relative from transmission experiment, presently normalized; Szmytkowski and Krzysztofowicz (1995) absolute. Elastic: Tanaka et al. (1988) absolute; Mapstone and Newell (1992) normalized to theory and presently integrated; Merz and Linder (1998), measurements and semiempirical extrapolation; Fink et al. (1975) normalized to independent-atoms theory and presently integrated. Momentum transfer: Floriano et al. (1986), McCorkle et al. (1978). Vibrational: Boesten et al. (1990) absolute. Ionization: Duric et al. (1991); Nishimura and Tawara (1994); Schram et al. (1966); results of Grill et al. (1993b) practically coincide with those of Duric et al. (1991). Dissociation: Winters (1979). Dissociative attachment: Rutkowsky et al. (1980).

Daresbury results, for comparison. The more recent measurements of backscattering from the Orsay laboratory (Lunt *et al.* 1998) for C_2H_6 differ from previous measurements and are not reported in fig. 4.

As already noticed by Brüche (1930a), alkanes exhibit a maximum in the TCS at about the same energy, 7–8 eV. The absolute values of these maxima scale gradually, by a factor of two between CH_4 (Sueoka and Mori 1986) and C_3H_8 (Tanaka et al. 1999). Additionally, some shoulder structure (Nishimura and Tawara 1991, Floeder et al. 1985) is visible at 20–30 eV in light alkanes, C_2H_6 , C_3H_8 . For alkenes (C_2H_4 , propene C_3H_6) and heavier alkanes (cyclo-propane C_3H_6 , n-buthane C_4H_{10} , iso-butane C_4H_{10} , 1-butene C_4H_8) this shoulder is less visible (Floeder et al. 1985).

At energies above 20 eV the TCS for scattering on hydrocarbons rises with increasing the number of electrons. This dependence in the 100-400 eV energy (E) range has been parameterized by Floeder *et al.* (1985) as

(1)
$$\sigma^{\pm}(E) = aN_e E^{-1/2} [1 \pm b \exp[-cE]],$$

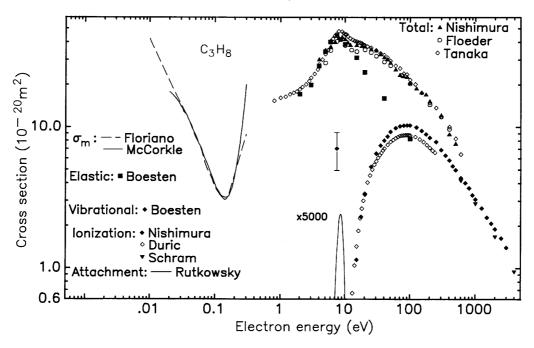


Fig. 5. – Integral cross-sections for electron scattering on propene (C₃H₈). *Total:* Floeder et al. (1985) absolute; Nishimura and Tawara (1991) absolute; Tanaka et al. (1999), absolute corrected for forward scattering. Elastic: absolute of Boesten et al. (1994). Momentum transfer: Floriano et al. (1986); McCorkle et al. (1978). Vibrational: absolute of Boesten et al. (1994) at 7.5 eV. Ionization: Duric et al. (1991); Nishimura and Tawara (1994); Schram et al. (1966); results of Grill et al. (1993b) practically coincide with those of Duric et al. (1991). Dissociative attachment: Rutkowsky et al. (1980).

where N_e is the total number of electrons in the target, signs + and - refer to positrons and electrons, respectively, and the parameters are $a=7.2\cdot 10^{-20}~\rm m^2~\rm eV^{-1/2},~b=0.17,~c=(333~\rm eV)^{-1}.$ On the other hand, Szmytkowski (1989) and Nishimura and Tawara (1991) argued that the TCS values in this energy range depend on the molecular polarizability. More measurements and in a larger energy range are needed to verify these models. Theoretical TCS for C_2H_4 were reported at 5–20 eV by Rescigno and Schneider (1992). The additivity rule was used by Jiang et~al.~(1995) at 10–1000 eV and Joshipura and Vinodkumar (1999) at 50–5000 eV to calculate molecular cross-sections starting from the atomic ones.

Elastic. Measurements on C₂H₄ were performed only by few groups (see table II for references). Mapstone and Newell (1992) have normalized their DCS at 8 eV and 15 eV to the theory of Winstead et al. (1991). The integral CS, obtained by integration of the DCS of Mapstone and Newell as performed in this work, agrees with the TCS (Floeder et al. 1985) at 8 eV. Below 8 eV, the integrated DCS of Mapstone and Newell increases with decreasing energy, in contrast to the TCS that decreases up to 2 eV. To get an agreement with the TCS values (Sueoka and Mori 1986) at lower energies, the relative data of Mapstone and Newell (1992) at 6.0, 5.0, 4.3 and 3.3 eV should be multiplied by 0.82, 0.72, 0.61 and 0.55, respectively. DCS in C₂H₄ at 3.3 eV and 8.0 eV (Mapstone and Newell 1992) are shown in fig. 6. Preliminary data of Suzuki et al. (1999) are quite

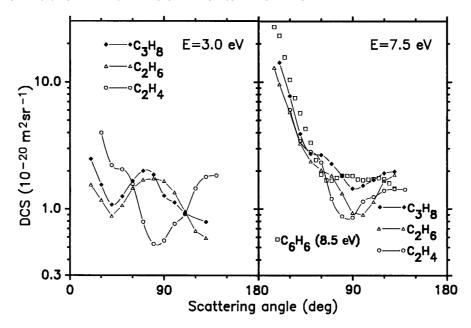


Fig. 6. – Differential cross-sections for elastic scattering of electrons on linear hydrocarbons and benzene. C_2H_4 (3.3 and 8.0 eV): Mapstone and Newell (1992), normalized to theory (Winstead et al. 1991) at 8 eV, 3.3 eV DCS renormalized (present work) to TCS (Sueoka and Mori 1986) by a factor 0.55. C_2H_6 : Tanaka et al. (1988) absolute. C_3H_8 : Boesten et al. (1994) absolute. C_6H_6 : Gulley and Buckman (1999) at 8.5 eV.

different compared to the measurements of Mapstone and Newell (1992) below 5 eV but agree with them at 8 eV. Suzuki *et al.* reported DCS at 2 eV almost uniform in angle. DCS at 90° at 0.4--16 eV were also reported by Allan (1994). These data, differently from the "TCS" measurements of Lunt *et al.* (1994), do not exhibit a minimum at 1 eV but only a shoulder structure, similarly to in C_2H_6 (see figs. 3 and 4).

For C_2H_6 the most recent differential and integral CS in the 0.3–10 eV range were given by Merz and Linder (1998). These authors also extrapolated their integral CS to lower energies using two semiempirical methods: the phase shift analysis (see eq. (7) in part I) and the modified effective range theory (eq. (8) in part I). DCS in the limit of zero-energy change quickly with energy. Already at 0.15 eV the DCS exhibits a minimum at 90° , indicating a dominant p-wave scattering (Merz and Linder 1998). The d-wave scattering, somehow masked by other partial wave scattering, becomes distinguishable at impact energy of 2 eV (Tanaka $et\ al.\ 1988$, Merz and Linder 1998).

The Ramsauer minimum in C_2H_6 integral elastic CS, as obtained in the semiempirical analysis of elastic DCS measurements (Merz and Linder, 1998), in the transmission (Orsay) and in the $90^{\circ} \pm 30^{\circ}$ scattering measurements of Lunt *et al.* (1994) and in the swarm experiments (McCorkle *et al.* 1978, Floriano *et al.* 1986), differ rather significantly, see fig. 4. However, these discrepancies can be somewhat apparent, as the physical quantities measured in these experiments are different and quick changes in DCS angular dependencies influence the integral cross-sections in different ways. Note that the integral elastic CS value at the Ramsauer minimum amounts to $3.0 \cdot 10^{-20}$ m² at 0.17 eV, three times more than the Ramsauer-Townsend minimum in momentum transfer CS,

 $1.0\cdot 10^{-20}$ m² at 0.1 eV, as evaluated by Merz and Linder (1998) and in agreement with McCorkle *et al.* (1978).

The C_2H_6 absolute elastic CS has also been measured at 2–100 eV, 15° –130° by Tanaka et al. (1988). Relative CS have been measured at 7.5–20 eV, 32° –142° by Curry et al. (1985) and at 3.3–15.5 eV, 30° –140° by Mapstone and Newell (1992). All these data agree reasonably well in the overlapping energy ranges. The integral elastic CS given by Tanaka et al. (1988) and those obtained in the present work by integration of the elastic data of Mapstone and Newell (1992) indicate a maximum of integral elastic CS at about 7–8 eV. The angular distribution at this energy is dominated by the f-partial wave scattering with some contribution from the p-wave (Tanaka et al. 1988). At 15 eV the normalized DCS of Mapstone and Newell are somewhat lower than the values of Tanaka et al. (1988).

C₂H₄, C₂H₆ (and CH₄) intermediate energy DCS have been measured by Fink et al. (1975) at 100–1000 eV, 0°–120° scattering angles and recently by Maji et al. (1998) at 300–1300 eV, 30°–120° scattering angles. Maji et al. (1998) have not compared directly their results to the earlier data. In C₂H₄ we have found, interpolating the data, that the DCS of Maji et al. at 300 eV agrees well with the measurements of Fink et al. (1975) at 130° but is underestimated by a factor of two at 30°. On the other hand, at 1000 eV the two data agree at 30° but at 130° the DCS of Maji et al. are lower by a factor of two. We note similar discrepancies for CH₄ (see subsect. 2.1 in part II), comparing also to the most recent measurements (Iga et al. 2000). In C₂H₆ at 100 eV the DCS of Fink et al. (1975) normalized to the independent-atoms model are a factor 2–3 higher than the data of Tanaka et al. (1988). Consequently, also the integral CS obtained in the present work using the data of Fink et al. is higher both than the integral CS of Tanaka et al. (1988) and than the TCS of Floeder et al. (1985), see fig. 4. At high energies the integrated CS of Finch et al. fall quickly with energy; some error due to the normalization procedure of DCS (Finch et al. 1975) is not to be excluded.

C₃H₈ integral elastic CS (Boesten *et al.* 1994, Tanaka *et al.* 1999) exhibit a maximum at the same energy (7.5 eV) as in CH₄ and C₂H₆. In fig. 6 we compare the DCS for C₂H₆ and C₃H₈ at 3.0 and 7.5 eV (Tanaka *et al.* 1988, Boesten *et al.* 1994) and at 3.3 and 8.0 eV for C₂H₄ (Mapstone and Newell 1992). In the region of the total and elastic CS maxima all the three gases exhibit similar DCS angular distributions. However, these distributions are quite different from that in CH₄ at the same energy (compare fig. 3 in part II). At 3 eV the DCS for C₂H₆ (Tanaka *et al.* 1988) and C₃H₈ (Boesten *et al.* 1994) are almost equal for all angles and resemble the *d*-wave pattern. At about the same energy the C₂H₄ angular distribution (Mapstone and Newell 1992, normalized in the present work to the TCS, see before) is more complex than in C₂H₆, see fig. 6. This indicates the vicinity of a resonant state, visible in the TCS at 2 eV (Sueoka and Mori 1986).

Calculations of elastic scattering on hydrocarbons at low energies have been performed, among others, by Sun *et al.* (1992) and McCurdy (1992) for C₂H₆, by Winstead *et al.* (1991) for C₂H₄, C₂H₆, C₃H₈, Si₂H₆, by Brescansin *et al.* (1998) for C₂H₄.

Swarm measurements at 0.02–0.06 eV for several hydrocarbons (Bowman and Gordon 1967) indicate a rise of the momentum transfer CS in the limit of zero energy; however, the absolute values given by Bowman and Gordon have not been confirmed by more recent analyses. In C_2H_4 the analysis of Duncan and Walker (1972), not shown in fig. 3, indicates no Ramsauer-Townsend minimum down to 0.01 eV; their momentum transfer CS is almost constant $(5 \cdot 10^{-20} \text{ m}^2)$ between 0.01 and 0.1 eV. Between 0.4 and 3.0 eV the momentum transfer CS of Duncan and Walker practically coincides with presently

normalized 90° elastic DCS of Allan (1994). For C_2H_6 and C_3H_8 the momentum transfer CS of McCorkle *et al.* (1978) and Floriano *et al.* (1986) practically coincide, while the data of Duncan and Walker (1974) are somewhat shifted towards lower energies. The momentum transfer CS in C_2H_6 obtained by the modified effective range theory (see eq. (8) in part I) from elastic DCS measurements (Merz and Linder 1998), not shown in fig. 4, fall between the results of McCorkle *et al.* (1978) and Duncan and Walker (1974).

The recent measurements (Schmidt and Roncossek 1992) of drift velocities, longitudinal and transverse diffusion coefficients in light hydrocarbons at 0.02– $14 \cdot 10^{-21}$ Vm² agree with previous data (McCorkle *et al.* 1978, Gee and Freeman 1980, 1983, Floriano *et al.* 1986) within $\pm 5\%$, apart from the drift velocities in C_2H_6 where the discrepancy amounts to -15% for low reduced fields. Drift velocities for low ($< 2 \cdot 10^{-21}$ Vm²) reduced fields were measured in 13 hydrocarbons from C_2H_4 to $C_{10}H_{22}$ by Christophorou *et al.* (1966), while characteristic energies for fields up to $50 \cdot 10^{-21}$ Vm² in n- C_5H_{12} to n- C_8H_{18} by Mechlińska-Drewko (1980). Christophorou *et al.* (1966) compared the low-energy hydrocarbon cross-sections to that in C_2H_4 , approximating them with an $E^{-1/2}$ energy dependence. Mechlińska-Drewko derived CS in the 0.05–0.4 eV range: these CS agree well at the position of the Ramsauer-Townsend minimum with those of Floriano *et al.* (1986) but are underestimated at 0.4 eV. Recently, Shishikura *et al.* (1997) have measured the drift velocity and the longitudinal diffusion in C_2H_6 and its mixtures with Ar at 0.03– $300 \cdot 10^{-21}$ Vm².

As noticed by Floriano et~al.~(1986), the Ramsauer-Townsend minimum occurs at the same energy $(0.12\pm0.01~{\rm eV})$ for all the chain alkanes from ${\rm C_2H_6}$ to $n\text{-}{\rm C_{10}H_{22}}$. The amplitude of the cross-section minima rises from about $1.2\cdot10^{-20}~{\rm m^2}$ and $3.0\cdot10^{-20}~{\rm m^2}$ for ethane and propane (Floriano et~al.~1986, McCorkle et~al.~1978), through $4.6\cdot10^{-20}~{\rm m^2}$ and $5.8\cdot10^{-20}~{\rm m^2}$ for butane and neopentane (McCorkle et~al.) to about $16\cdot10^{-20}~{\rm m^2}$ and $22\cdot10^{-20}~{\rm m^2}$ for n-octane and n-decane, respectively (Floriano et~al.~1986). The position of the minimum in the momentum transfer CS shifts to higher energies with the increase of the sphericity of the molecule: it occurs at $0.17~{\rm eV}$ for i-butane, $0.22~{\rm eV}$ for neo-pentane and $0.25~{\rm eV}$ for methane (Floriano et~al.~1986). The influence of the chain length on the electron mobility was studied by Gee et~al.~(1988). Electron scattering parameters in 14 hydrocarbons have been reviewed by Freeman and Armstrong (1985).

Scattering length. Large discrepancies exist among the different evaluations of the scattering length in hydrocarbons. Measurements of the pressure shift in perturbated molecular photoionization (Meyer et al. 1991) gave scattering lengths of $-2.34a_0$, $-3.33a_0$ and $-4.31a_0$ for CH₄, C₂H₆ and C₃H₈, respectively. Merz and Linder (1998) extrapolating their DCS measurements with the use of a modified effective range theory (see eq. (8) in part I) obtained a scattering length of $-3.00a_0$ in C₂H₆. Lunt et al. (1994) in C₂H₆ reported a scattering length of $-4.47a_0$ from their transmission data, of $-4.09a_0$ from the 90° \pm 30° experiment and of $-4.21a_0$ from the swarm data of McCorkle et al. (1978). In C₂H₄ Lunt et al. obtained a scattering length of $-11.7a_0$ extrapolating their data and of $-20.9a_0$ analyzing the momentum transfer CS of Bowman and Gordon (1969).

Vibrational excitation. The C₂H₄ vibrational excitation exhibits two peaks at about 1.8 eV and 7.5 eV (Walker et al. 1978). In the first region the dominating excited modes are totally symmetric ν_2 (C=C stretching) and ν_3 (C=C stretching with H-C-H bending). Both modes exhibit a d-wave-like angular distribution. At 1.95 eV the vibrational excitation amounts to about 10% of the TCS, with ν_2 , ν_3 and 2 ν_3 CS equal to 1.4 · 10⁻²⁰ m², 1.1 · 10⁻²⁰ m² and 0.3 · 10⁻²⁰ m² (±40%), respectively (Walker et al. 1978). At 7.5 eV, apart from ν_2 and ν_3 , the dominant mode is ν_1 (symmetric C-H stretching) and the DCS are isotropic. Allan (1994) reported the ν_2 absolute DCS at

 $1.95~{\rm eV}$: his data practically coincide with Walker et~al.'s measurements. In fig. 3 the integral vibrational excitation of Walker et~al. at $1.95~{\rm eV}$ is shown.

In C₂H₆ (Boesten *et al.* 1990) and in C₃H₈ (Boesten *et al.* 1994) the vibrational excitation exhibits an enhancement around the energy corresponding to the maximum in the total and/or elastic CS, indicating a resonant process. The contribution from the vibrational channel to the TCS at the maximum position is higher for these two heavier hydrocarbons than for CH₄ (Szmytkowski and Krzysztofowicz 1995, Tanaka *et al.* 1999). It amounts to 10% and 15% for C₂H₆ and C₃H₈, respectively. For both C₂H₆ and C₃H₈, unresolved bending modes are excited with slightly (approximately 3:2) higher probability than stretching modes. In C₃H₈ a rise of the vibrational excitation of both bending and stretching modes has been observed at large (90° and 120°) scattering angles close to 2 eV (Boesten *et al.* 1994). This could be an indication of a low-energy resonant state.

Direct (non-resonant) vibrational excitation in C_2H_4 and C_2H_6 is rather weak. In particular, only little signal was observed at 1 eV from the most intense infrared active, ν_7 (out-of-plane bending) mode in C_2H_4 (Walker et al. 1978); in C_2H_6 only a slight rise was seen in 30° DCS towards 1 eV (Boesten et al. 1990). In C_2H_6 , according to the Born approximation (eq. (15) in part I) with dipole transition moments from Bishop and Cheung (1982), the integral elastic CS is expected to reach a maximum of $0.5 \cdot 10^{-20}$ m² at 0.55 eV so it could be hardly visible in TCS. Low values of vibrational CS in the near-to-threshold region in C_2H_4 has also been observed in the transmission experiment of Lunt et al. (1994). Vibrational excitation DCS in C_2H_6 in the range 3.2-15.4 eV and in C_2H_4 at 8 eV and 15.5 eV have recently been reported by Mapstone et al. (2000).

Electronic excitation. C₂H₄ is the simplest system with C-C π bonding. For this reason electronic excitation in ethene was extensively studied in energy loss (Lassettre and Francis 1964, Ross and Lassettre 1966, Lassettre et al. 1968a, Trajmar et al. 1970, Kuppermann et al. 1979, Wilden and Comer 1980, Allan et al. 1994), trapped-electrons (Bowman and Miller 1965, Brongersma et al. 1969, Hubin-Franskin and Collin 1970, Verhaart and Brongersma 1980) and threshold electrons (Dance and Walker 1973, van Veen 1976) experiments. Energy loss spectra indicate the presence of the triplet at \tilde{a} ³B_{1u} state, observable as a broad maximum extending from 3.2 eV to about 6 eV. This structure is well separated from the next prominent feature, the ¹B_{1u} (π , π *) valence excited state centered around 7.5 eV. This last state overlaps with a number of sharp Rydberg transitions (Wilden and Comer 1980) and with a broad one (Vergaart and Brongersma 1980), presumably a valence transition in the 8–12 eV energy-loss range (Trajmar et al. 1970, Allan et al. 1994).

The \tilde{a} $^3B_{1\mathrm{u}}$ state of $\mathrm{C_2H_4}$ has been extensively studied by Allan (1994). The rovibrationally integrated DCS at 90° reaches a maximum of $0.04 \cdot 10^{-20}$ m²/sr at 7.0 eV. This value is somewhat lower than the theory of Rescigno and Schneider (1992). The DCS for this state exhibit an unusual, backward-peaked angular dependence, especially at higher collision energies (7.2 eV and 14.2 eV, Allan, 1994; 25 eV, Trajmar *et al.*, 1970). Measurements of Allan (1994) did not confirm the existence of a resonance in the at \tilde{a} $^3B_{1\mathrm{u}}$ excitation, as observed in the trapped-electron measurements (van Veen 1976) and as predicted theoretically at 5 eV (Rescigno and Schneider 1992).

Excitations in C_2H_4 to the Rydberg, 7.11 eV energy loss (for $\nu'_2 = 0$), to the \underline{E} (8.93 eV for $\nu'_2 = 0$) and to the \underline{C} (8.27 eV for $\nu'_2 = 0$) states were studied by Trajmar et al. (1970). DCS for these states at 25 eV (collision energy) are forward centered and scale with respect to each other by a factor of five each. Assignment of Rydberg levels was done in electron energy loss studies (Wilden and Comer 1980) as well as in high-resolution

photoabsorption experiments (Wu et al. 1991 and Chen and Wu 1999). However, the interpretation of the electronic spectrum in C_2H_4 is still controversial (compare Mulliken 1977, Ryu and Hudson 1995).

Forward electron scattering experiments in C_2H_4 were performed with a low-energy resolution (1 eV FWHM) by Ibuki et al. (1989) and with a high one (0.05 eV) by Cooper et al. (1995). Due to the narrowness of the energy loss lines, the values of maxima in the derived photoabsorption CS depend on the energy resolution. For this reason we quote only energy-integrated photoabsorption CS rather than absolute values in the maxima. The photoabsorption oscillator strength integrated over energy amounts to 0.423, 0.124 and 0.333 in the 6.3–8.79 eV, 8.79–9.45 eV and 9.45–10.99 eV bands, respectively (Cooper et al. 1995). Inner shell transitions in C_2H_4 (and C_2H_2) were studied by Tronc et al. (1979).

In C_2H_6 , energy loss spectra were studied by Lassettre and Francis (1964) at 390 eV collision energy; by Ross and Lassettre (1966) at 150 eV; by Lassettre *et al.* (1968b) at 50–180 eV; by Johnson *et al.* (1979) at 30–250 eV; by Dillon *et al.* (1987) at 200 eV. No singlet-triplet transitions were observed for energy losses down to 4.5 eV (Lassettre *et al.* 1968b).

High-resolution forward-scattering energy loss spectra in C_2H_6 show three vibronic progressions centered at 9.4, 10.7 and 11.5 eV which are due to transitions from the $1e_g$ orbital to different Rydberg orbitals (Au et al. 1993). The first electronic transition $1e_g \rightarrow 3s$ with the onset at 7.9 eV (Johnson et al. 1979) is very weak. The photoionization CS for the first Rydberg band $1e_g \rightarrow 3p$ reaches its maximum of $0.3 \cdot 10^{-20}$ m² at 9.4 eV (Au et al. 1993).

For C_3H_8 we are aware of an extensive study of Ehrhardt et al. (1965) at 70 eV collision energy, 0–30 eV energy-loss and 0°–145° angular range. The energy loss spectrum at 6° shows a maximum at about 8–9 eV that amounts to 4/5 of the peak at 16 eV. Energy-loss spectra were also reported by Lassettre et al. (1968a) and Dillon et al. (1987). Electron excitation in forward scattering in C_3H_8 was studied by Au et al. (1993). The first peak in the photoabsorption CS, at 8.9 eV, amounts to $0.3 \cdot 10^{-20}$ m². Compared to C_2H_6 , the CS peak in the photoionization range rises significantly, from $0.75 \cdot 10^{-20}$ m² to $1.2 \cdot 10^{-20}$ m² in C_2H_6 and C_3H_8 , respectively (Au et al. 1993). Electron momentum spectroscopy in (e, 2e) experiment was done by Pang et al. (1999) and Jia et al. (1999).

Optical emission from dissociated fragments in hydrocarbons was measured by Beenakker and de Heer (1975), Donohue et al. (1977), Möhlmann and de Heer (1977), Tokeshi et al. (1996). Emission CS for the Balmer H_{α} line (amounting to $3.1 \cdot 10^{-22}$ m² in CH₄) diminishes with the increasing of the number of carbon atoms in the molecule, from $2.5 \cdot 10^{-22}$ m² in C₂H₆ to $0.62 \cdot 10^{-22}$ m² in n-C₆H₁₄ (Möhlmann and de Heer 1977). Emission from the $A^2\Delta \to X^2\Pi$ transition in the CH radical falls from $0.61 \cdot 10^{-22}$ m² to $0.2 \cdot 10^{-22}$ m² for the same series of molecules. Unsaturated hydrocarbons (C₂H₂, C₂H₄) show higher emission CS than C₂H₆; the difference is especially significant for the CH ($A \to X$) emission (Möhlmann and de Heer 1977). Vibrational states of the dissociated CH fragment were studied by Tokeshi et al. (1996).

Dissociation CS was measured in C_2H_6 at 15–600 eV by Winters (1979). The dissociation CS is somewhat higher than the total ionization CS. This is because the measurements of Winters include dissociation both due to electronic excitation and ionization. The sum check indicates that Winter's data are in a good agreement with the other CS.

Ionization. The recent measurements of ionization in hydrocarbons (Chatham et al. 1984, Duric et al. 1991, Grill et al. 1993a, b, Nishimura and Tawara 1994) agree quite well with earlier data (Rapp and Englander-Golden 1965, Schram et al. 1966).

Generally, the measurements of Nishimura and Tawara (1994) are higher than the other sets of data, but the difference does not exceed 15%. The semiempirical values of C₂H₆ ionization CS from the binary-encounter Bethe model by Hwang et al. (1996) agree within the experimental error bar with the measurements of Chatham et al. (1984), Duric et al. (1991) and Grill et al. (1993a) normalized to this latter at 100 eV, and are somewhat lower than those of Schram et al. (1966). In C₃H₈ the semiempirical model falls between the data of Duric et al. (1991) and Grill et al. (1993b) in the threshold region and coincides with these two sets and the results of Schram et al. (1996) above 100 eV. Only in C₂H₄ the model of Hwang et al. is lower than the experiment (Rapp and Englander-Golden 1965, Nishimura and Tawara 1994), by 20% in the maximum of the CS. The single-point measurements at 75 eV (Lampe et al. 1957) and at 70 eV (van Houte et al. 1992) disagree rather seriously with other experiments.

Both for C_2H_6 (Chatham *et al.* 1984) as well as for C_3H_8 (Grill *et al.* 1993b) the dissociative ionization dominates: parent ions $C_2H_6^+$ and $C_3H_8^+$ constitute 13% and 6.5%, respectively, of the total ionization CS at 100 eV. The $C_2H_4^+$ ion (50% of the yield) dominates in C_2H_6 ionization and the $C_2H_5^+$ ion (25%) in C_3H_8 .

Dissociative attachment. Broad and weak peaks for the H⁻ production, centered at about 9.8, 9.5, 9.0 eV, were observed in C_2H_6 , C_3H_8 and n- C_4H_{10} , respectively (Dorman 1966). More recently, Rutkowsky et al. (1980) reported dissociative attachment CS in the 0–12 eV energy range for as many as 14 hydrocarbons with one-to-four carbon atoms. CS for heavier hydrocarbons of this series are lower than that for CH_4 , being of the order of 10^{-24} m². C_2H_4 and saturated hydrocarbons CH_4 , C_2H_6 , C_3H_8 , C_4H_{10} show only one peak, shifting gradually from 10.5 eV to 8.5 eV. Alkenes and alkadienes develop also some low-energy peaks, depending not only on the overall chemical formula, but also on the isomeric structure (Rutkowsky et al. 1980).

Resonances. The peak in TCS for C_2H_4 at 1.8 eV (Sueoka and Mori 1986) corresponds to the $^2B_{2g}$ shape resonance due to the capture of one electron to the π^* orbital (Walker et al. 1978, Gallup 1986). This resonance was seen as a structureless dip in the transmission spectra (Sanche and Schulz 1973, Burrow and Jordan 1975), in the vibrational excitation channel (Walker et al. 1978, Ben Arfa and Tronc 1990, Allan 1994). A peak at 7.5 eV in the vibrational DCS has been assigned (Allan 1994) to the $^2(\pi, \pi^{*2})$ $^2B_{3u}$ core excited resonance, with a hole in the π -orbital and a double occupancy of the π^* orbital. This resonance was observed also in trapped-electrons experiments (Dance and Walker 1973, van Veen 1976) and in electron transmission spectra in the 6–8 eV energy range (Sanche and Schulz 1973). We are not aware of any confirmation of the weak structure seen at about 0.3 eV in transmission current by Boness et al. (1967) and in the Orsay transmission experiment (Lunt et al. 1994). Theoretical analysis of possible symmetries of the resonant states in C_2H_4 was performed by Gallup (1986) and Ben Arfa and Tronc (1990).

For the alkanes, CH₄, C₂H₆, C₃H₈ a broad maximum was observed in integral elastic CS (Boesten *et al.* 1994) and TCS at about 7.5 eV. In C₂H₆, on the basis of elastic angular distributions (Tanaka *et al.* 1988) and the vibrational excitation (Boesten *et al.* 1990) the existence of a short-lived resonant state has been postulated at this energy. Two possible symmetries are: $^2A_{2u}$ corresponding to a temporary capture of the electron to the π^* valence orbital of the CH₃ group and 2E_u with the capture of the electron to the σ^* -orbital of the C-C valence bond (Boesten *et al.* 1990, Merz and Linder 1998). Shape resonances in C₂H₂, C₂H₄ and C₂H₆, as seen in carbon atom 1s photoionization, have been discussed recently by Kempgens *et al.* (1997). Temporary anion states of numerous polyatomic hydrocarbons were also reviewed by Jordan and Burrow (1987).

Sum check for C₂H₄ is not possible, at the moment. At low energies elastic CS

Table III. – Integral cross-sections for electron scattering on ethane (C_2H_6) (in 10^{-20} m² units).

Energy	Elastic	Vibrational	Ionization	Dissociation	Summed	Total
3.0	14.7 T	0.97 B			15.7	16.0 S
5.0	24.3 T	1.20 B			25.5	25.1 S 24.6 NT
7.5	31.7 T	3.49 B			35.2	34.7 S 30.6 F
10	28.8 T	2.60 B			31.4	32.5 S 35.4 NT
15	25.3 T	1.71 B	0.62 N 0.22 C	1.3 W	27.6 28.5	28.9 S 31.0 NT
20	21.8 T	0.88 B	2.24 N 0.44 C	3.0 W	24.9 26.1	27.6 S 29.7 NT
40	14.1 T		5.41 N 0.69 C	6.6 W	19.4 21.4	21.9 S 23.8 NT
100	6.6 T		6.89 N 0.80 C	7.6 W	13.5 15.0	15.4 S 16.4 NT 14.7 F

- B Boesten *et al.* (1990)
- C Chatham et al. (1984), non-dissociative part of ionization CS
- F Floeder et al. (1985), values in selected points are given only
- N Nishimura and Tawara (1994)
- NT Nishimura and Tawara (1991), read from figure
- C Szmytkowski and Krzysztofowicz (1995)
- T Tanaka et al. (1988)
- W Winters (1979), dissociation (neutral + ionization), read from figure

(Mapstone and Newell 1992, Allan 1994, Suzuki et al. 1999) and TCS (Sueoka and Mori 1986, Lunt et al. 1994) measurements and the theory (Winstead et al. 1992) are in serious disagreement. Ambiguities regard even the shape of the cross-section, see fig. 3. Serious discrepancies regard also high-energy CS: elastic (Fink et al. 1975, Maji et al. 1998) on one side and total (Sueoka and Mori 1986, Floeder et al. 1985, Nishimura and Tawara 1991) on the other.

For C_2H_6 the sum check, see Table III, would indicate a perfect (within 3%) agreement between summed elastic and vibrational cross-sections of Tanaka and co-workers and the total values of Szmytkowski and Krzysztofowicz (1995) up to the ionization threshold. This could be someway fortuitous, taking into account the experimental error bars: 30% for the integral vibrational CS (Boesten et al. 1990), 15–22% for the elastic CS (Tanaka et al. 1988) and 5% for the TCS. Between 15 eV and 100 eV the summed (with the ionization data of Nishimura and Tawara 1994) cross-sections are less than 10% smaller than the TCS; this amount could correspond to the electronic excitation. Electronic excitation CS can also be evaluated from the difference between the dissociation CS (Winters 1979) and the dissociative ionization CS (Chatham et al. 1984): such a difference amounts to

Table IV. –	Integral	cross-sections	for	electron	scattering	on	propane	(C_3H_8)	(in	10^{-20}	m^2
units).											

Energy	Elastic	Vibrational	Ionization	Summed	Total
7.5	44.5 B	7.0 B		51.5	45.7 T 41.5 NT
10	38.4 B	5.2*		43.6	45.1 T 42.0 NT 37.7 F
15	30.9 B	3.5*	1.14 N	35.5	40.2 T 38.0 NT 34.2 F
20	24.3 B	1.6*	3.31 N	29.2	37.8 T 37.1 NT
40	15.9 B		8.0 N	23.9	32.3 T 31.1 NT
100	8.21 B		10.2 N	18.4	21.5 T 20.5 NT

^{*} Rough extrapolation (present work) from excitation functions of Boesten et al. (1994)

 $1.8 \cdot 10^{-20}$ m² at 30 eV and $2.3 \cdot 10^{-20}$ m² at 100 eV. At 100 eV the sum of elastic (Tanaka et al. 1988), dissociation (Winters 1979) and non-dissociative ionization (Chatham et al. 1984) CS is in good agreement with the TCS value (Szmytkowski and Krzysztofowicz 1995).

Also for C_3H_8 , see Table IV, the partial (Boesten et al. 1994, Nishimura and Tawara 1994) cross-sections sum pretty well to the total (forward-scattering corrected) CS (Tanaka et al. 1999). Some discrepancy exist at 7.5 eV where probably the integral elastic CS of Boesten et al. (1994) is overestimated. On the other hand the TCS of Floeder et al. at 10 eV and 15 eV are lower than other TCS results (Tanaka et al. 1999, Nishimura and Tawara 1991) and the summed value. At 100 eV the difference between the total (Tanaka et al. 1999) and elastic (Boesten et al. 1994) plus ionization (Nishimura and Tawara 1994) CS gives a rough estimate, $2.1 \cdot 10^{-20}$ m² of the electronic excitation CS. This values agrees with our estimate of the electronic excitation in C_2H_6 at the same energy (see above). Both for C_2H_4 (Chatham et al. 1984) and C_3H_8 (Grill et al. 1993b) ionization constitutes 40% of the TCS at 100 eV (Szmytkowski and Krzysztofowicz 1995 and Tanaka et al. 1999, respectively). Above 100 eV, it is not possible to perform the sum check in C_2H_6 and C_3H_8 due to the lack of reliable data.

B - Boesten et al. (1994), normalized to elastic CS

F - Floeder *et al.* (1985)

N - Nishimura and Tawara (1994)

NT - Nishimura and Tawara (1991), read from figure

T - Tanaka et al. (1999)

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 $2^{\circ}4$. Fullerene (C₆₀). – Fullerene discovery (Kroto et al. 1985, Krätschmer et al. 1990) triggered numerous experimental and theoretical investigations. In many aspects electron scattering on fullerene is atypical. This target attaches electrons with a very high CS in a few eV collision energy range. Lifetimes of the negatively charged complexes are very long. For this reason, almost no resonant vibrational excitation was observed for fullerenes in the gas phase (Elhamidi et al. 1997). We are not aware of TCS measurements in fullerene.

Elastic DCS in C_{60} in the $30^{\circ}-90^{\circ}$ angular and in the 1-12 eV energy ranges have been measured by Tanaka et al. (1994). There are big differences in the energy dependences of the DCS at specific angles: for 30° a single maximum of DCS is observed at 5 eV, for 50° , a minimum at 5 eV, a broad maximum at 8 eV and a rise towards 1 eV. At 9 eV three minima are observable in the $30^{\circ}-90^{\circ}$ angular range. A rough evaluation of the absolute DCS value of $410 \cdot 10^{-20}$ m²/sr at 7 eV and 30° scattering angle could be subject to an error by a factor of 2 or 3 (Tanaka et al. 1994).

Elastic CS in C_{60} has been calculated by Gianturco *et al.* (1999) at 0–30 eV scattering energies. A number of resonant structures is visible in the integral CS. In the whole energy range examined the integral CS exceeds the value of $100 \cdot 10^{-20}$ m², showing a sharp peak of $180 \cdot 10^{-20}$ m² at 3 eV and a broad, structured maximum of about $160 \cdot 10^{-20}$ m² around 10 eV.

Vibrational excitation. Vibrational infrared frequencies in gas phase were found to be (Frum et al. 1991) 65.7, 71.1, 146 and 175 meV. The lowest Raman transition was observed at 34 meV (Bethune et al. 1991). Electron energy loss spectra obtained with 30 meV resolution show at 7 eV (30° , 50°) two peaks: a higher one in the 150 meV–200 meV region and a lower one in the 60 meV–110 meV (Tanaka et al. 1994).

Electronic excitation. The ground state of C_{60} is 1A_g . Energy loss spectra for condensed phase were obtained, among others, by Gensterblum et al. (1991), Weaver et al. (1991), Lucas et al. (1992). Electron energy loss spectra in gas phase at 5° scattering angle, 50 and 100 eV collision energy were reported by Tanaka et al. (1994). They observed peaks at 2.2, 3.7, 4.8, 5.5. and 5.8 eV corresponding to one-electron $\pi \to \pi^*$ transition (Gensterblum et al. 1991). Tanaka et al. reported also some features at 5.8, 6.1, 6.3 and 6.5 eV superimposed on the 1 eV-wide π -electron plasmon (collective excitation) peak and a weak maximum at the 28.3 eV energy loss, attributed to the $\sigma + \pi$ plasmon excitation.

Bulliard et al. (1993) studied electron energy loss spectra with 60 meV resolution for the 1–10 eV energy loss and residual energies from 0.08 to 20 eV. They localized three distinct bands at 2.24, 3.77 and 4.88 eV and a large one at 6.1 eV. At low residual energies, where forbidden transitions are enhanced, they observed a rise of the 2.2 eV band, a sharp peak at 2.98 eV and some signal around 3.4 and 4.4. eV. A band peaking at 2.26 eV, being a superposition of triplet states and optically forbidden single states, was also observed in spectra with low residual energies by Abouaf et al. (1993). These authors, similarly to Bulliard et al., observed a triplet character peak at 2.96 eV. In none of the energy-loss spectra for the C_{60} gas phase (Abouaf et al. 1993, Bulliard et al. 1993, Jaffke et al. 1994, Tanaka et al. 1994) the lowest $^3T_{2g}$ triplet state reported at 1.55 eV in solid film (Lezius et al. 1992) was found.

At 10 and 70 eV collision energy the DCS for all different energy loss processes between 2.3–7.3 eV are forward centered (Abouaf *et al.* 1993). At 70 eV, 4° scattering angle, the 1 $^{1}T_{\rm u}$, 3 $^{1}T_{\rm u}$, 6 $^{1}T_{\rm u}$ energy loss peaks (3.03, 3.78, 4.84 eV) rise in amplitude; a possible peak for the 8 $^{1}T_{\rm u}$, excitation at 5.87 eV is hidden in the plasmon maximum.

Ionization. The total ionization CS (Düsner et al. 1995, Matt et al. 1996) rises quickly

with energy (Tarnovsky et al. 1998) up to $33 \cdot 10^{-20}$ m² at 50 eV. The fall at high energies is unusually slow: $16 \cdot 10^{-20}$ m² at 1000 eV (Matt et al. 1996). The C_{60}^+ ion amounts to 75% of total ionization at 100 eV and almost 90% at 1000 eV. Fragment ions, with even number of atoms, down to C_{44}^+ were observed with diminishing intensities (Matt et al. 1996). In a more recent experiment, fragment ions down to C_4^+ were also observed (Muigg et al. 1998).

The high rate for multiple ionization in C_{60} is rather anomalous compared to other molecular targets: for example, the C_{60}^{2+} ion yield amounts to 30% of the C_{60}^{+} one (Matt et al. 1996). In this respect, fullerene resembles noble gases. Also the small share of the dissociative ionization is anomalous (C_{58}^{+} being 30 times less than the C_{60}^{+} yield). The semiempirical model of Deutsch et al. (1996) disagrees with the experiment, giving a maximum value of ionization CS of almost $100 \cdot 10^{-20}$ m² and overestimating the high-energy (0.4–5.0 keV) experimental CS (Itoh et al. 1999) by a factor of two. The first determination of the ionization CS in C_{60} , $54 \cdot 10^{-20}$ m² at 38 eV (Sai Baba et al. 1992a), was higher than the recent measurements (Matt et al. 1996). Ionization of C_{70} was studied by Sai Baba et al. (1992b) and Matt et al. (1997).

Dissociative attachment in C_{60} is subject to intense investigations. It was measured by flowing afterglow method (Smith et al. 1993, Smith and Španěl 1996), Rydberg atoms quenching (Huang et al. 1995, Fink et al. 1995, Weber et al. 1996) and beam (Elhamidi et al. 1997) methods. Smith et al. reported a very small rate coefficient at 300 K rising to $3 \cdot 10^{-7}$ cm³ s⁻¹ at 4500 K. From these measurements an activation energy of 0.26 eV was deduced. Smith and Španěl (1996) argued that, in spite of the high attachment rate, similar to that in SF₆, the attachment mechanism in these two targets is essentially different: an s-wave capture in SF₆ and a p-wave capture (with a centrifugal barrier) in C_{60} . Jaffke et al. (1994) in a beam experiment obtained a similar value of the activation energy. They also showed quite complicated energy dependence of attachment CS between the threshold and 12 eV, with several maxima reaching about $100 \cdot 10^{-20}$ m². In another work of this group (Lezius et al. 1993) the zero-energy peak has been attributed to a nuclear-excited Feshbach resonance mechanism.

Quite different results, indicating a high electron attachment CS in C_{60} at zero energy, have been obtained in Rydberg-atoms quenching experiments (Huang *et al.* 1995, Finch *et al.* 1995, Weber *et al.* 1996) and in a high-resolution beam experiment (Elhamidi *et al.* 1997). These experiments indicate a rate coefficient of $2 \cdot 10^{-8}$ cm³ s⁻¹ (Weber *et al.* 1996) for the formation of the C_{60}^- anion at near-to-zero energies. Furthermore, this rate is independent of the quantum number of the Rydberg atom, indicating an *s*-wave character of the electron capture.

Elhamidi et al. (1997) located the maxima in attachment CS at 0.75, 1.20, 2.08, 2.74, 3.5, 4.27 and 5.35 eV and attributed them to long-lived resonant states; the first two of these states are shape resonances, the other states are understood as Feshbach resonances. The lifetimes of the C_{60}^- temporary anion falls exponentially with the incident electron energy, from 500 μ s at 7 eV to 40 μ s at 11 eV (Elhamidi et al. 1997). Studies of electron attachment to bigger fullerenes C_n , n=70, 76, 84 (Abouaf et al. 1999), show that at the same electron energy the lifetime of metastable C_n^- anions rises exponentially with the number of carbon atoms in the molecule.

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3. - Halides

Many types of halides, mainly fluorides and chlorides (and among them mixed halides containing H, F and Cl atoms) were studied by electron scattering. In this section we follow the presentation order proposed for hydrides (and hydrocarbons). We start from five-atom molecules: chlorofluorocarbons (CF_4 , CCl_4 and mixed), then methyl halides (CH_3X) and silicon tetrahalides (SiX_4). In subsect. 3.5 we discuss NF_3 and some similar molecules, in subsect. 3.6 fluorine-containing analogs of heavier hydrocarbons (C_2F_6 , C_3F_8 , C_6F_6). At the end, in subsects. 3.8 and 3.9 we review electron scattering cross-sections in SF_6 and its analogs.

Tetrahedral fluorides (CF₄, SiF₄) exhibit features in TCS similar to those observed in tetrahedral hydrides. One particularity of both groups of molecules is that a Ramsauer-Townsend minimum in the elastic and/or total CS is well visible. However, the presence of peripheral fluorine atoms in CF_4 and SiF_4 influences the resonant scattering differently. More marked differences regard also vibrational excitation, enhanced by substituting hydrogen atoms by fluorine ones.

3.1. Tetrafluoromethane (CF₄). – The importance of CF₄ relies in the fact that this gas is the most frequently used for plasma etching and semiconductor passivation (Kastenmaier et al. 1996). For this reason, besides measurements of total and partial CS, very particular phenomena of electron scattering, like distribution of kinetic energies of dissociated fragments, have been studied in detail (Harland and Franklin 1974, Le Coat et al. 1994, Bruce et al. 1994). As a result of this recent interest, a set of quite congruent data have been established for CF₄. Cross-sections summaries, mostly oriented towards plasma modeling, were given by Hayashi (1987), Morgan (1992), Bordage et al. (1996); rather rough sets of momentum transfer CS have been published by Mašek et al. (1987) and Stefanov et al. (1988). Numerous works regard plasma modeling (Jauberteau et al. 1991, Gogolides et al. 1994).

Extensive reviews on CF₄ electron impact CS have recently been published by Bonham (1994) and Christophorou *et al.* (1996). We refer the reader to these works for detailed figures and tabulated presentation of partial CS and swarm parameters in CF₄ and its mixtures with other gases. In fig. 7 we report integral CS for CF₄; in fig. 13 (subsect. 3.2) we compare DCS for elastic scattering in CF₄, CF₃Cl and CF₂Cl₂.

Total cross-sections. Total absolute CS have been measured from 0.4 eV (Jones 1986, Szmytkowski et al. 1992) up to 4000 eV (Zecca et al. 1992). Relative measurements between 0.07 and 0.4 meV were obtained by Field et al. (1984). The agreement between the data of Jones and Szmytkowski et al. is very good at low energies. Between 20 and 120 eV the data of Szmytkowski et al. seem to be the most reliable. The data of Jones at 50 eV are underestimated by 7% (Jones 1986) due to scattering on exit orifices of the gas cell. In fig. 7 we report them corrected by an energy-dependent factor, varying smoothly from 1.01 at 10 eV to 1.07 at 50 eV. The data of Zecca below 120 eV can be overestimated due to beam instabilities. The remeasured data by Sueoka et al. (1994) are in better agreement with other experiments at low energies than the previous measurements from the same apparatus (Mori et al. 1985). However, above 30 eV the data of Sueoka et al. are somewhat lower than other results and seem to be slightly energy shifted. Such a shift is plausible in a time-of-flight apparatus, using a longitudinal guiding magnetic field.

In the experiment of Field et al. (1984) at Orsay laboratory, a synchrotron radiation electron source and a transmission geometry were used; the energy resolution of the

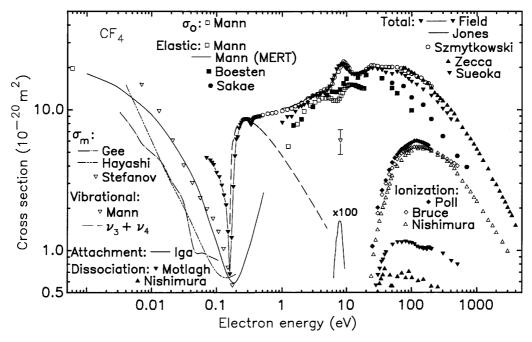


Fig. 7. – Integral cross-sections for electron scattering on CF_4 (in 10^{-20} m² units). Total: Field et al. (1984), normalized (this work) to Szmytkowski et al. (1992) at 0.4 eV; Szmytkowski et al. (1992); Jones (1986); Zecca et al. (1992); Sueoka et al. (1994). Elastic: Mann and Linder (1992a); Boesten et al. (1992); Sakae et al. (1989); MERT analysis of Mann and Linder (1992a) at low energies. Momentum transfer: Hayashi (1987); Stefanov et al. (1988); Gee and Freeman (1991). Vibrational excitation: Born approximation (eq. (15), part I), in general agreement with Mann and Linder (1992b) for ν_3 and ν_4 modes; at 8 eV, presently integrated DCS of Mann Ionization: summed partial cross-sections of Poll et al. (1992); Bruce and Linder (1992b). and Bonham (1993); Nishimura et al. (1999); semiempirical model of Kim and Rudd (1999) in Nishimura et al. (1999) and data from coincidence measurements (Bruce et al. 1994) not shown for clarity. Dissociation into neutrals: absolute of Motlagh and Moore (1998); Nishimura et al. (1999), semiempirical analysis; absolute of Mi and Bonham (1998) and of Nakano and Sugai (1992) not shown for clarity; total (ionization + neutrals) dissociation CS of Winters and Inokuti (1982), not shown for clarity, coincide above 50 eV with ionization CS (Bruce and Bonham 1993). Dissociative attachment: Iga et al. (1992), sum of CF₃⁻ and F⁻ signals.

apparatus was 40 meV. The relative measurements of Field $et\ al.$ have been normalized (in this work) at 0.45 eV to the data of Szmytkowski $et\ al.$ These data suggest the existence of a very narrow Ramsauer-Townsend minimum in TCS at 0.17 eV followed by a small bump at 0.27 eV. Note that this narrowness, similarly to the case of C_6H_6 , could be an instrumental artifact.

Swarm experiments. Due to the above-mentioned practical importance of CF₄, the swarm parameters (electron mobility, characteristic energies) have recently been subjected to numerous measurements (Curtis et al. 1988, Hunter et al. 1988, Schmidt and Polenz 1988, Gee and Freeman 1991, Lisovskiy and Yegorenkov 1999) and semiempirical analysis (Mašek et al. 1987, Vasenkov 1999). The analysis of swarm results is difficult in the case of CF₄ due to the overlap between the maximum in the vibrational CS and the minimum in the elastic channel (see fig. 7).

In particular, Gee and Freeman (1991) and Hayashi (1987) reported a momentum transfer CS in the Ramsauer-Townsend minimum of about $1 \cdot 10^{-20}$ m², while Curtis *et al.* (not shown in the figure) one of $0.01 \cdot 10^{-20}$ m². Similarly, some discrepancies exist at very low energies (*e.g.*, 0.07 eV), where different data (of Gee and Freeman, Curtis *et al.*, Hayashi, Stefanov *et al.*) span from about $6 \cdot 10^{-20}$ m² to $15 \cdot 10^{-20}$ m², respectively.

Mann and Linder (1992a) used a modified effective range formalism (see part I, eq. (8)) to derive elastic and momentum transfer CS at low energies from their elastic differential CS measurements above 0.5 eV. Their momentum transfer CS at 0.1 eV is bounded by the results of Hayashi from the upper and Curtis $et\ al$. from the lower side. We note also that the early approximation of the momentum transfer CS at thermal energies from cyclotron resonance measurements (Tice and Kivelson 1967) by the 8/E function (in $10^{-20}\ {\rm m}^2/{\rm meV}$ units) was quite reliable, agreeing reasonably well both with Hayashi's and Mann and Linder's analysis.

Scattering length. Mann and Linder (1992a) by a modified effective range analysis derived a zero-energy cross-section σ_0 of $19.5 \cdot 10^{-20}$ m². This value is close to that in CH₄. We note also a good agreement in the determination of the Ramsauer-Townsend minimum between Mann and Linder's (1992a) analysis and the normalized TCS measurements of Field *et al.* (1984), see fig. 7. However, according to Hayashi (1987) and Morgan (1999), the zero-energy CS for CF₄ could be much bigger, exceeding the value of $60 \cdot 10^{-20}$ m².

Elastic cross-sections of Boesten et al. (1992) between 1.5 and 100 eV overlap very well (within 10%) in their low-energy limit with the measurements of Mann and Linder (1992a), normalized to Jones (1986). Both sets of data indicate a rapid fall of integral CS below 3 eV, pointing to the existence of a Ramsauer-Townsend minimum, predicted also theoretically (see, for example, Gianturco et al. 1996). Mann and Linder (1992a) extrapolated their experimental elastic CS towards zero energy using a modified effective range theory (part I, eq. (8)). According to this evaluation the Ramsauer-Townsend minimum amounts to $0.6 \cdot 10^{-20}$ m² at 0.185 eV.

At low energies we note a resonant dip in the integral elastic CS at 7 eV, just below the maximum in TCS. This dip, followed by a maximum, is visible both in Boesten's et al. and Mann and Linder's data. All recent theories, Huo (1988), Winstead et al. (1993), Gianturco et al. (1996), Natalense et al. (1995), Isaacs et al. (1998) reproduce sharp peaks in elastic integral CS but not such a dip structure. Note that none of these theories include vibrational excitation effects. Differential CS exhibit rapid changes in shape across the resonance, becoming forward peaked just at 10 eV (Mann and Linder 1992a).

At 100 eV the integral CS of Boesten et al. (1992) are 20% lower than the intermediate-energy measurements of Sakae et al. (1989). On the other hand, the discrepancy in differential CS for these two sets of data is, on the average, less than 10%. Therefore, the difference in the integral values derives probably from extrapolation and/or integration procedures. As Sakae et al. extended the DCS angular measurements down to 5°, their integrated values seem to be more reliable at 100 eV than the result of Boesten et al.

Tanaka et al. (1997) compared DCS at $15^{\circ}-135^{\circ}$ and 1.5, 30 and 100 eV in CF₄, CH₄ (and fluoromethanes CH₃F, CH₂F₂, CHF₃). At 1.5 eV, the CF₄ and CH₄ differential cross-sections are similar in shape; the absolute values in the two gases at 90° differ by a factor of two (compare also fig. 3 in part II and fig. 13 for 2 eV energy). At 100 eV the absolute DCS in CF₄ and CH₄ differ by a factor of four; DCS in CF₄ shows a more complex, interference-like pattern. Well-pronounced structures, due to interference (Raj 1991) and absorption (Khare et al. 1994) effects, are also visible in the intermediate en-

ergy DCS (Sakae $et\ al.\ 1989$). Meier $et\ al.\ (1994)$ evidenced effects of multiple scattering in small-angle elastic DCS at 1 keV.

Vibrational excitation, differently than in CH₄, influences highly the TCS below 0.4 eV. Mann and Linder (1992b) and also Curtis et al. (1988) indicated the importance of the ν_3 (asymmetric stretch), optically allowed vibrational mode. Mann and Linder proved also a good agreement between excitation functions for infrared-active modes at different angles and the Born approximation, up to a few eV energy. Therefore, we applied the Born approximation (part I, eq. (15)) to the ν_3 excitation at low energies: the ν_3 cross-section amounts to about 70% of the TCS at 0.5 eV. We note however that vibrational-excitation CS for ν_3 and ν_4 (asymmetric deformation) modes as derived from swarm data by Curtis et al. would exceed Born values and exhibit near-to-threshold structures, similar to those observed experimentally, for example, in HCl (see subsect. 3.5 in part II).

The ν_3 excitation function at 90° shows a large, structureless maximum centered at about 10 eV (Boesten et al. 1992). In the region of the maximum of TCS at 9 eV, the ν_3 vibrational excitation is the dominant channel for the resonant decay, with cross-section amounting to about $5 \cdot 10^{-20}$ m² (Mann and Linder 1992b, Boesten et al. 1992). Only the $2\nu_3$ and barely $3\nu_3$ overtones were observed at 8 eV, indicating a short lifetime of the resonant state (Boesten et al.). The overall vibrational excitation, as estimated roughly in the present work by integrating the angular spectra of Boesten et al. (in agreement with Mann and Linder 1992b), contributes to about 1/3 of the TCS at 8 eV, see a point with error bar in fig. 7.

Compared to CH₄, the vibrational excitation in CF₄ is more pronounced, both for direct (near-to-threshold) and resonant scattering, see fig. 7. This difference reflects the molecular structure, *i.e.* higher values of transient dipole moments for CF₄ than for CH₄ (Bishop and Cheung 1982).

Electronic excitation. Energy-loss spectra have been reported for valence excitations by Harshbarger and Lassettre (1973) and King and McConkey (1978) at 500 eV collision energy, by Kuroki et al. (1992) at 200 eV, by Verhaart et al. (1978) and Curtis and Walker (1989) at 11–17 eV in a trapped-electron experiment; for inner shell excitations by Tronc et al. (1979) at 1.5 keV collision energy and by Letardi et al. (1987) at a constant final energy of 1.5 keV. Absolute oscillator strength for valence excitations have been given by King and McConkey (1978) and Zhang et al. (1989) and for inner shells excitations by Zhang et al. (1992). Photoabsorption CS below the ionization threshold reaches a maximum of $0.24 \cdot 10^{-20}$ m² at 13.5 eV (Zhang et al. 1989). Detailed momentum spectroscopy for valence electrons was performed by Leung and Brion (1984). We are not aware of measurements of electronic excitation CS; an indirect evaluation can be done using dissociation CS (see below).

Optical emission in a form of a large band in the 200–500 nm wavelength region, attributed to different dissociated fragments in excited states, has been observed in numerous works (van Sprang et al. 1978, Aarts 1985, Blanks and Becker 1987, Müller et al. 1992 and reference therein). The absolute CS at 100 eV for the overall emission feature between 200 and 560 nm was evaluated as $0.45 \cdot 10^{-20}$ m² (Müller et al. 1992).

Absolute values for emission from fluorine atoms produced in collisional dissociation at 200 eV electron impact energy have been given by Becker and collaborators (Roque et al. 1991, Blanks et al. 1987): $0.8 \cdot 10^{-20}$ m² for emission at 95.5 and 97.5 nm due to transitions into the ground state $(2p^43s)^{2,4}P \rightarrow (2p^5)^2P^0$ and about $4.0 \cdot 10^{-20}$ m² for emission in the spectral range 620–780 nm, due to transitions from the 3p excited manifold $(2p^43p) \rightarrow (2p^43s)^{2,4}P^2$. In the CS given for this first transition the cascad-

ing contribution has been subtracted (Roque et al. 1991). Emission due to inner-shell excitation was studied by Glans et al. (1994).

Ionization cross-sections of Bruce and Bonham (1993) superseding those of Ma et al. (1992) and Bruce et al. (1992) agree within a few percent with the re-measured (Poll et al. 1992) cross-sections from the Innsbruck group (Stephan et al. 1985). The semiempirical results of Kim and collaborators from a binary-encounter-Bethe model (Nishimura et al. 1999, Kim and Rudd 1999) coincide with the measurements of Poll et al. (1992) and Bruce et al. (1992), if multiple ionization is included in the model. Some discrepancies regard energies above 100 eV, see fig. 7.

The newest measurements (Nishimura et al. 1999) are about 20% lower in the maximum than those of Poll et al. The binary-encounter model agrees perfectly with Nishimura et al.'s measurements, if no multiple ionization is allowed. Nishimura et al. argued that the difference between the binary-encounter-Bethe models with and without multiple ionization accounts for the dissociation into neutrals.

Similarly to SiH₄, the ionization in CF₄ has a dissociative character: no parent CF₄⁺ stable ion has been observed in electron scattering (Mi *et al.* 1996) neither in photoionization (Colding *et al.* 1991). Two excited states of CF₄⁺ with lifetimes of 16 and 180 ns were observed in the optical emission experiment of van Sprang *et al.* (1978). CF₃⁺ is the prevailing stable ion, amounting to 2/3 of the total ionization. Yields for production of other single charged ions (CF₂⁺, CF⁺, F⁺, C⁺) are one order of magnitude smaller (Bruce and Bonham 1993, Poll *et al.* 1992).

From optical emission measurements at 100 eV (Müller et al. 1992, Blanks and Becker 1987, van Sprang 1978) it follows that more than 15% of CF₃⁺ ions are created in excited states. Also other debris fragments, like F atoms, are produced in excited states (van der Burgt 1991, Blanks et al. 1987) or with high recoil energies. About 6% of the ionization events leads to simultaneous production of two ions (Bruce et al. 1992, 1994).

Detailed sets of partial ionization, dissociation into neutrals and dissociative attachment CS were given by Bonham (1994).

Dissociation CS (dissociative ionization + neutrals) was measured by Winters and Inokuti (1982) up to 500 eV. Dissociation into neutrals can be evaluated from the difference between these measurements and the ionization CS. As numerous ionization CS determinations (Ma et al. 1991, Poll et al. 1992, Bruce and Bonham 1993) diverge above 200 eV, some indetermination regards also the dissociation CS. Dissociation into neutrals is an important scattering channel at energies below 100 eV. The difference between the data of Winters and Inokuti (1982) and those of Poll et al. (1992) amounts to $0.5 \cdot 10^{-20}$ m² at 30 eV.

Recently, several direct measurements of dissociation into neutrals have been performed. Nakano and Sugai (1992) and Sugai et~al.~(1995) studied selective ionization of radicals; Mi and Bonham (1998) measured DCS for scattered electrons in coincidence with ions; Motlagh and Moore (1998) used adsorption of radicals at tellurium surface. Bonham and Bruce (1992) in a semiempirical analysis reported $0.98 \cdot 10^{-20}~\text{m}^2$ at 80 eV. In direct measurements Mi and Bonham (1998) obtained $0.74 \cdot 10^{-20}~\text{m}^2$ at 35 eV. These values agree well with the determination of Motlagh and Moore and the recent analysis of Nishimura et~al.~(1999), see fig. 7. Bigger discrepancies exist above 100 eV.

The measurements of Sugai et~al.~(1995) indicate at 120 eV a dissociation-into-neutrals CS as low as $0.11 \cdot 10^{-20}$ m². According to these data the CF₃ radical is the dominating neutral fragment, exceeding both CF₂ and CF yields by almost two folds. Sugai et~al. normalized their data using the ionization CS for radicals of Tarnovsky and Becker (1993). Nakano and Sugai (1992) using a different normalization, to the N₂ dissociation,

reported higher CS, $0.73 \cdot 10^{-20}$ m² at 120 eV. This value is in good agreement with measurements of Motlagh and Moore (1998). On the other hand, the data of Nakano and Sugai seem to be underestimated below 100 eV, being much lower than the determination done by Winters and Inokuti (1982).

The recent analysis of Nishimura *et al.* (1999) using their ionization CS and measurements of Winters and Inokuti (1982) agrees very well in shape with the data of Motlagh and Moore (1998) but is lower by a factor of two, exactly. Dissociation into neutrals (Mi and Bonham 1998) amounts to 3.5% of the TCS (Szmytkowski *et al.* 1992) at 35 eV.

Metastables relative yields for collision energies up to 400 eV have been studied in a time-of-flight experiment by van der Burgt and McConkey (1991). Thresholds for several fragmentation processes have been established; the majority of them have been attributed to formation of F atoms in different excited states. No cross-sections have been reported.

Electron attachment. A broad peak for negative-ions formation is centered at about 7.5 eV (Harland and Franklin 1974, Spyrou et al. 1983, Curtis and Walker 1989, Weik and Illenberger 1995). Different measurements agree well. The F^- , CF_3^- and F_2^- yields amount, respectively, to 78%, 20% and 2% of the total dissociative attachment CS (Iga et al. 1992). Note, however, a different partitioning (50% for CF_3^- and F^- , each) reported by Spyrou et al. (1983). The positions of different negative ion peaks move from 6.8 eV for F_2^- to 7.8 eV for CF_3^- (Iga et al. 1992). In its maximum the dissociation attachment CS (Iga et al. 1992) constitutes only about $8 \cdot 10^{-4}$ of TCS.

Analysis of the angular distributions for each ion species (Le Coat et~al.~1994) and their kinetic energies (Le Coat et~al.~1997) indicates that more than one resonance state can be involved in the dissociative attachment between 6 and 12 eV. The parent CF_4^- anion is metastable (see the calculation of Gultsev and Adamowicz 1995) and was observed in electron scattering on CF_4 clusters (Lotter et~al.~1989).

Resonances. The attribution of the resonant structures observed in vibrational and dissociation CS is not clear as the agreement with theory (Tossel and Davenport 1984, Huo 1988, Winstead et al. 1993, Natalense et al. 1995, Gianturco et al. 1996) is rather poor. The 8 eV broad structure visible in all studied channels has been attributed (Boesten et al. 1982) to the T_2 (l=1) shape resonance. However, Le Coat et al. (1994) in their analysis of the angular distribution for negative-ions production indicate two overlapping resonances, a shape one with 2T_2 symmetry and a Feshbach one, with 2T_1 symmetry, placed, respectively, at about 6.5 and 7.7 eV. This attribution was confirmed by successive measurements (Le Coat et al. 1997) on kinetic energies of F^- fragments.

At about 20–25 eV another, broad maximum is visible in total (Szmytkowski et al. 1992), elastic and vibrational (Boesten et al. 1982) CS. It has been attributed to the T_2 (l=2) resonance (Boesten et al. 1982).

There is some uncertainty on the existence of a weak resonance at about 0.3 eV. A local maximum was visible in the relative TCS measurements of Field *et al.* (1984). Similarly, the ν_3 vibrational excitations derived from swarm data by Curtis *et al.* (1988) exhibit a peak, above the Born's value (formula (15), part I). This resonance was tentatively attributed to a Rydberg state by Verhaart *et al.* (1978) on the basis of the vibrational structure visible in negative-ions yield.

A sum check performed at selected energies is presented in table V. At very low energies elastic data of Mann and Linder (1992a, b) are summed with ν_3 and ν_4 excitation evaluated from Born approximation (formula (15), part I, with dipole transition moments after Bishop and Cheung 1982). At 1.5 eV and above, the elastic CS of Boesten et al. (1992) were used. At 0.16 eV and between 0.4 and 5 eV the summed CS exceed

Table V. - Integral cross-sections for electron scattering on CF₄.

Energy (eV)	Elastic	Vibrational	Dissociation	Ionization	Sum	Total
0.16	0.62 M	0.44 B			1.06	0.675 F
0.20	0.60 M	5.1 B			5.7	6.1 F
0.40	1.58 M	8.6 B			10.2	8.8 Sz
1.5	7.74 Bo	4.6 B			12.3	10.1 Sz
3.0	10.5 Bo	2.9 B			13.4	12.6 Sz
5.0	12.7 Bo	2.0 B			14.7	13.5 Sz
8.0	13.9 Bo	6.0 M			19.9	20.0 Sz
15	16.9 Bo	0.8**			17.7	17.9 Sz
20	17.63* Bo	0.7**	1.0** W	(0.31 P)	19.3	19.3 Sz
30	17.0 Bo	0.5**	2.5 W	(2.01 P)	20.0	20.4 Sz
50	15.5 S		4.3 W	(4.22 P)	19.8	20.1 Sz
100	12.2 S		5.55 W (1.12 MM)	(5.82 P)	17.8	18.0 Z
150	10.8 S		5.51 W	(5.82 P)	16.3	16.3 Z
200	8.56 S		5.32 W	(5.17 Br)	13.9	14.4 Z
300	6.46 S		4.72 W	(4.66 Br)	11.2	11.6 Z
500	4.68 S		3.78 W	(3.98 Br)	8.46	8.47 Z

^{*} Interpolated data

the total one. This discrepancy can be due to two reasons. The first one is a possible overestimation of the vibrational CS by the Born approximation, as was indicated by measurements of Mann and Linder (1992b) at large angles. The second reason is that, at very low energies, different factors can introduce systematic errors to TCS measurements. At the resonance position (8 eV) the elastic CS of Boesten et al. (1992) sums up very

^{**} Extrapolated data

^() Not included in summation

 $[\]dot{\mathsf{B}}$ - Born approximation (present work) for ν_3 and ν_4 , with dipole transition moments from Bishop and Cheung (1982)

 $[\]mbox{\sc Bo}$ - Bosten $\it et \,\it al. \,(1982)$

Br - Bruce and Bonham (1993)

F - Field et al. (1984), normalized (this work) to Szmytkowski et al. (1992) at 0.4 eV

M - Summed vibrational excitations (Mann and Linder 1992b), integrated in this work

MM - Motlagh and Moore (1998), dissociation into neutrals

P - Poll et al. (1992)

S - Sakae et al. (1989)

Sz - Szmytkowski et al. (1992)

W - Winters and Inokuti (1982), total dissociation

Z - Zecca et al. (1992)

well with the vibrational excitation (Mann and Linder 1992b, integrated in the present work) to the TCS value (Szmytkowski et al. 1992).

At high energies we summed the dissociation data of Winters and Inokuti (1982) with the elastic CS of Boesten et al. (1992) at 20 and 30 eV and Sakae et al. (1989) above 50 eV. Agreement of the summed CS with the measured absolute TCS is excellent. Therefore, we suppose that the ionization CS of Poll et al. (1992) and Bruce and Bonham (1993) are probably slightly overestimated. According to Motlagh and Moore (1998), the dissociation into neutrals, occurring via the electronic excitation would constitute 5% of TCS (Szmytkowski et al. 1992) at 100 eV. This is a similar amount as the electronic excitation CS contribution in CH₄ (see subsect. 2.1 in part II).

Comparison with CH_4 . The CF_4 molecule, due to its symmetry and zero dipole moment, for some aspects resembles the CH_4 molecule. The elastic CS of the two molecules are similar, with a well-pronounced Ramsauer-Townsend minimum. Differently to CH_4 , however, the CF_4 total cross-section is highly influenced by vibrational excitations of infrared-active modes both at low energies and in the region of the 8 eV resonance. This difference reflects the molecular structure, *i.e.* high values of transient dipole moments for the CF_4 molecule.

Another difference regards the partitioning between ionization and total CS; it is more marked in CH₄ than in CF₄ (at 500 eV 55% for the CH₄ molecule compared to 45% for CF₄). A similar effect for other hydrocarbons and perfluoro compounds was noticed by Gee and Freeman (1990). The different thresholds for ionization (12.6 eV for CH₄ and 16.9 eV for CF₄) reflect the electronegativity of the outermost atom (0.75 eV for H and 3.4 eV for F). The high energies of dissociated fragments in CF₄ (Bruce *et al.* 1994, van Burgt and McConkey 1991) are probably due to the big difference (ΔE) between the energies of the dissociation and ionization thresholds: $\Delta E = 10.3$ eV for CF₄ (Nakano and Sugai 1992) compared with $\Delta E = 1.2$ eV for CH₄.

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3.2. Chlorofluoromethanes (CF₃Cl, CF₂Cl₂, CFCl₃) and CCl₄. – Chlorofluoromethanes are widely used in the semiconductor industry for plasma etching and passivation. The CCl₄ and CFCl₃ molecules attach electrons at very low energies and are used in electrical switches. All four gases possess strong absorption bands in the infrared and therefore are active as greenhouse gases. Furthermore, they are very stable and this leads to long permanence times in the atmosphere. The role of chlorine-containing gases in the stratospheric ozone destruction is also generally acknowledged (Eckman et al. 1987).

The cross-sections and transport coefficients for CF₂Cl₂ and CCl₄ have been reviewed by Novak and Fréchette (1985) and Hayashi (1987). For CF₂Cl₂ an extensive review of total, partial cross-sections and swarm coefficients, containing recommended values, was performed by Christophorou *et al.* (1997).

Total cross-sections for all chlorofluoromethanes and CCl₄ have been measured between 0.5 eV and 50 eV by Jones (1986) using a time-of-flight apparatus and between 80 eV and 4000 eV by Zecca et al. (1992) using a Ramsauer-type transmission setup. A linear transmission method with an electrostatic energy selector has been used by Szmytkowski et al. (1992) to measure CCl₄ total cross-sections between 0.5 eV and 200 eV. A trochoidal spectrometer has been used between 0.2 and 12 eV by Underwood-Lemons et al. (1994) for CF₃Cl and CF₂Cl₂ and by Wan et al. (1991) for CCl₄. Cross-sections for scattering into a backward cone for CF₃Cl, CF₂Cl₂, CFCl₃ and CCl₄ have been measured by Randell et al. (1993). TCS for CF₃Cl, CF₂Cl₂, CFCl₃ and CCl₄ are presented in figs. 8–11, respectively. TCS values of Jones (1986) for all four gases and CF₄ up to 20 eV are compared in fig. 12.

As stated by Jones (1986), his measurements at energies above 20 eV suffered from a systematic error due to scattering on the exit apertures; this effect tends to lower the measured cross-sections. For this reason the data reported in figs. 8–12 have been corrected by an energy-dependent factor, according to the error limits given by Jones (1986). The applied correction was +7% at 50 eV descending smoothly to 0% at 4 eV. The corrected data merge well with the results of Zecca et al. (1992) at higher energies and almost coincide, in the case of CCl₄, with the measurements of Szmytkowski et al. (1992). Note that the recent, remeasured data of Hamada and Sueoka (1995) for CCl₄ are still slightly lower than Jones's and Szmytkowski et al.'s sets, although the difference (10% on the average) is smaller than in their previous experiment (Mori et al. 1985). The early measurements of Holst and Holtsmark (1931) for CCl₄ agree generally in shape but are lower than the recent results (Jones 1986, Szmytkowski et al. 1992). Holst and Holtsmark observed two maxima in TCS, at about 1 and 7 eV; for the second maximum the amplitude reported $(58 \cdot 10^{-20} \text{ m}^2)$ is close to the recent determination $72.4 \cdot 10^{-20}$ m² (Szmytkowski et al. 1992). The theoretical work in CCl₄ of Natalense et al. (1995) reproduces the general shape of TCS but all calculated structures are shifted to higher energies.

At low energies the data of Wan et al. (1991) for CCl₄ agree well with the measurement of Jones (1986) and Szmytkowski et al. (1992). In their high-energy limit the data of Wan et al. are slightly energy shifted with respect to the other recent measurements. Underwood-Lemons et al. (1994) measured TCS for CF₃Cl, CF₂Cl₂ (and CF₃Br, CF₃I, CF₂Br₂). Their data agree very well in shape with the measurements of Jones (1986) and also in absolute values, if corrected by multiplicative factors of 1.25 for CF₃Cl and 1.33 for CF₂Cl₂. This discrepancy in Underwood-Lemons et al.'s measurements should be attributed to difficulties in determining the effective, energy-dependent scattering length in their trochoidal spectrometer (Wan et al. 1991).

Randell et al. (1993) using a magnetically confined beam obtained cross-sections for

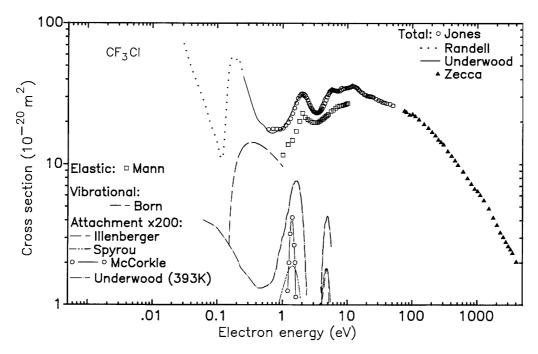


Fig. 8. – Integral cross-sections for electron scattering on CF₃Cl. *Total*: Jones (1986), absolute, corrected (this work: see text) for angular resolution; Underwood-Lemons *et al.* (1994) normalized (this work) by a factor of 1.25; Randell *et al.* (1993), backward scattering, normalized (this work) to Underwood-Lemons *et al.* (1994); Zecca *et al.* (1992), absolute. *Elastic*: Mann and Linder (1992a). *Vibrational*: Born approximation (eq. (15), part I), present work. *Electron attachment*: Spyrou and Christophorou (1985), beam, absolute; Illenberger (1979), beam normalized (this work) to Spyrou and Christophorou (1985) at 4.8 eV; McCorkle *et al.* (1980), swarm; Underwood-Lemons *et al.* (1995), beam, at 393 K.

"backward scattering" in the three chlorofluoromethanes and CCl₄ at very low energies. They normalized the relative data in CF₃Cl and CF₂Cl₂ at 10 meV to the rovibrational excitation CS obtained in the Born approximation (eqs. (14) and (15), part I) and the data in CFCl₃ and CCl₄ to the theoretical electron attachment CS obtained assuming an s-wave scattering (Klots 1976). In figs. 8–11 we present the results of Randell normalized to the lowest-energy points of Underwood-Lemons et al. (1994) for CF₃Cl and CF₂Cl₂, Jones (1986) for CFCl₃ and Wan et al. (1991) for CCl₄. Only in the case of CF₃Cl the normalization done by Randell et al. has to be corrected by as much as a factor of ten.

Normalized data of Randell *et al.* match pretty well in shape with the measurements by Wan and collaborators (Wan *et al.* 1991, Underwood-Lemons *et al.* 1994) and indicate a sharp rise of TCS below 0.01 eV collision energy. For CF₃Cl and CF₂Cl₂ weak maxima in "backward scattering" CS at about 0.15 eV have been observed, in some correspondence of the maxima in vibrational CS, as evaluated from the Born approximation (see figs. 8, 9).

All four considered gases exhibit several maxima between 1 eV and 15 eV, see fig. 12 in which Jones' data are reported. The amplitude of the first of these peaks rises and shifts to lower energies going from CF₃Cl to CFCl₃ and CCl₄. Only for CF₂Cl₂ the first peak is located at an energy lower than that for CCl₄. The three chlorofluoromethanes

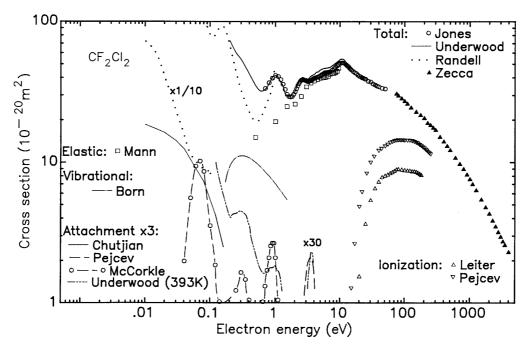


Fig. 9. – Integral cross-sections for electron scattering on CF₂Cl₂. Total: Jones (1986), absolute, corrected (this work: see text) for angular resolution; Underwood-Lemons et al. (1994), normalized (this work) by a factor of 1.33; Randell et al. (1993), backward scattering, normalized (this work) to Underwood-Lemons et al. at 0.2 eV; Zecca et al. (1992), absolute. Elastic: Mann and Linder (1992b). Vibrational: Born approximation (eq. (15), part I), present work. Electron attachment: Chutjian and Alajajian (1987), analytical formula from photoionization; McCorkle et al. (1980), swarm; Pejčev (1979), beam, absolute, low-energy data not shown due to a poor energy resolution; Underwood-Lemons et al. (1995), beam, at 393K. Ionization: Leiter et al. (1989), Pejčev et al. (1979).

CF₃Cl, CF₂Cl₂ and CFCl₃ exhibit another weak structure between 3.5–5.5 eV and a broad maximum at about 10 eV. Above 15 eV the TCS for CF₃Cl, CF₂Cl₂, CFCl₃ and CCl₄ follow, within the experimental error, the same functional dependence on energy (Zecca *et al.* 1992).

Elastic cross-sections for CF₃Cl, CF₂Cl₂ have been measured by Mann and Linder (1992a, b, respectively); the absolute CS scale has been defined by normalization to TCS values of Jones et al. (1986) at 4 eV and checked by relative-flow technique. Both for CF₂Cl₂ (Mann and Linder 1992b) and CF₃Cl (Mann and Linder 1992a) the elastic integral CS rises between 1 and 10 eV. For CF₃Cl a maximum has been evidenced at about 2 eV, in correspondence to the first maximum in the TCS. The DCS energy dependencies indicate the presence of a Ramsauer-Townsend minimum below 1 eV for both CF₃Cl and CF₂Cl₂ (Mann and Linder 1992a, b). The DCS for elastic scattering in CF₃Cl and CF₂Cl₂ at 2 eV are forward peaked, differently than in CF₄ (see fig. 13); this has to be attributed to the permanent dipole moments of CF₂Cl₂ and CF₃Cl. At 8 eV the DCS (fig. 13) for all three gases are forward peaked. One also notes a more complex angular dependence of DCS for CF₂Cl₂ than for CF₃Cl and CF₄, both at 2 eV as well as at 8 eV. We recall that CF₂Cl₂ has the lowest symmetry compared to the other

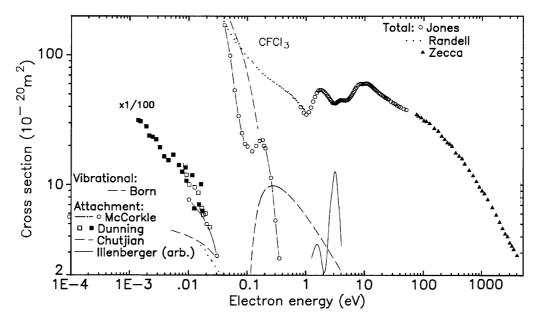


Fig. 10. – Integral cross-sections for electron scattering on CFCl₃. Total: Jones (1986), absolute, corrected (this work: see text) for angular resolution; Randell et al. (1993), backward scattering, normalized (present work) to Jones et al. at 0.8 eV; Zecca et al. (1992), absolute. Vibrational: Born approximation (eq. (15), part I), present work. Electron attachment: McCorkle et al. (1980, 1982), swarm; Zollars et al. (1984), open squares and Dunning (1987), filled squares, Rydberg atoms quenching; Chutjian et al. (1984), photoionization technique, analytical formula; Illenberger (1979), beam, arbitrary units.

chlorofluoromethanes. For this reason, CF₂Cl₂ could support more possible resonant states (Mann and Linder 1992b).

Elastic DCS of Daimon et al. (1983) for CCl₄ at 70 eV, 200 eV and 400 eV were normalized to the independent-atom-model calculation. In the present work the DCS of Daimon et al. have been extrapolated to 0° and then integrated. The comparison of these integral CS with the total (Szmytkowski et al. 1992, Zecca et al. 1992) and ionization (Leiter et al. 1984, 1989) CS suggests that the data of Daimon et al. are overestimated by more than a factor of two at 70 eV and by at least a factor of 1.5 at 200 eV. In fig. 11 we have reported these points multiplied by a 0.5 factor. The effects of multiple intermolecular scattering in CF₃Cl elastic DCS at 1 keV were studied by Meier et al. (1994).

Vibrational cross-sections have been measured by Mann and Linder (1992a) and (1992b) for CF_3Cl and CF_2Cl_2 , respectively.

In CF₃Cl below 1 eV the vibrational excitation functions of the infrared active mode $\nu_1(\text{CF}_3 \text{ symmetric stretch})$ unresolved from the ν_4 mode (CF₃ asymmetric stretch) and of the ν_2 mode (CF₃ symmetric deformation) follow roughly the Born approximation (see Mann and Linder 1992a).

The resonant vibrational excitation evaluated as the difference between Jones' (1986) TCS and the elastic CS (Mann and Linder 1992a) shows three maxima: at about 1.7 eV, 5.5 eV and 8.5 eV. At 1.7 eV the symmetric C-Cl stretching mode ν_3 dominates over the ν_2 . At 5.5 eV overtones up to the 6-th level of the $\nu_{1,4}$ unresolved mode are excited. At

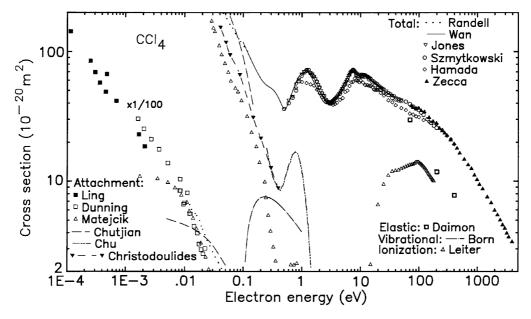


Fig. 11. – Integral cross-sections for electron scattering on CCl₄. Total: Randell et al. (1993), backward scattering normalized to attachment; Jones (1986), absolute, corrected (this work: see text) for angular resolution error; Wan et al. (1991), absolute; Hamada and Sueoka (1995), absolute; Szmytkowski et al. (1992), absolute; Zecca et al. (1992), absolute. Elastic: Daimon et al. (1983) rescaled (this work) by a factor of 0.5. Vibrational: Born approximation (eq. (15) part I), present work. Ionization: Leiter et al. (1984), normalized (this work, see text). Electron attachment: Ling et al. (1992) filled squares, Dunning (1987) and Zolters et al. (1984), open squares, Rydberg atoms quenching; Christodoulides and Christophorou (1971), swarm unfold; Chu and Burrow (1990), beam, normalized (present work) to swarm; Chutjian and Alajajian (1985), threshold photoionization, analytical formula; Matejcik et al. (1995), beam.

8.5 eV a very weak peak in the ν_2 excitation function is observed.

In CF₂Cl₂ (Mann and Linder 1992b) at energies below 1 eV, the ν_1 (CF₂ symmetric stretch) unresolved from the ν_6 (CF₂ asymmetric stretch) and the ν_8 (CCl₂ asymmetric stretch) infrared active modes are excited with similar intensities. At higher energies the difference between the total (Jones 1986) and the elastic (Mann and Linder 1992b) cross-section exceeds the values of the vibrational excitation evaluated through the Born approximation in the 1–10 eV energy range (Mann and Linder 1992b). In particular, this difference evidences indirect (resonant) vibrational excitation peaks centered at about 1.0 eV and 2.5 eV, and weak large structures around 4 eV and 6 eV. At 1.0 eV the ν_3 (CCl₂ symmetric stretch) and at 4.0 eV the $\nu_{1,6}$ modes prevail. At the 1 eV resonance the vibrational excitation contributes about 40% to the TCS.

For the three chlorofluoromethanes and CCl₄ we have reported in figs. 8–11 the vibrational excitation for infrared active modes as calculated by the Born approximation formula (eq. (15), part I), using transient dipole moments of Bishop and Cheung (1982).

Electronic excitation spectra for chlorofluoromethanes were reported by Verhaart et al. (1978) and for CCl₄ by Lee et al. (1975). King and McConkey (1978) gave absolute oscillator strength spectra in the region of valence and Rydberg transitions for CFCl₃, CF₂Cl₂ and CF₃Cl. Absolute oscillator strengths for excitations in the 7.5–200 eV energy

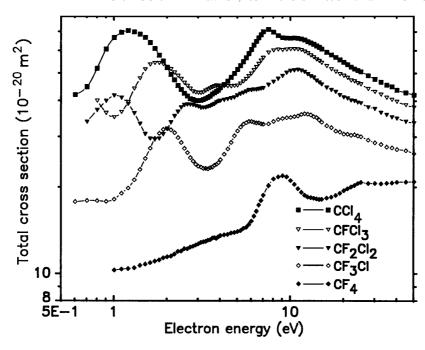


Fig. 12. – Comparison of total cross-sections at low energies for CF₄, CF₃Cl, CF₂Cl₂, CFCl₃ and CCl₄ (Jones 1986).

loss range have been given for CF_3Cl , CF_2Cl and $CFCl_3$ by Zhang et~al. (1991a, 1991b, 1991c, respectively); for inner-shell excitations in all the five gases, CF_4 - CCl_4 , by Zhang et~al. (1992); for valence-shell by Au et~al. (1997). Oscillator strength for inner- and valence-shell excitation in CCl_4 has also been measured by Burton et~al. (1994).

Optical emission from chlorofluoromethanes is dominated by the de-excitation of several molecular and atomic fragments. Absolute emission cross-sections in the 185–850 nm band for CF₃Cl, CF₂Cl₂, CFCl₃ were measured by van Sprang et al. (1978) and for CF₂Cl₂ by Jabbour and Becker (1989) and Martínez et al. (1992). At 100 eV the summed emission CS in the 670–780 nm region from excited F atoms diminishes with diminishing the number of fluorine atoms in the molecule: 3.3, 2.8, 2.2, 1.7 \cdot 10⁻²² m², for CF₄, CF₃Cl, CF₂Cl₂ and CFCl₃, respectively (van Sprang et al. 1978). In a complementary way the summed emission CS from chlorine atoms in various excited states rise: 3.7, 6.2, 9.2 \cdot 10⁻²² m² for CF₃Cl, CF₂Cl₂ and CFCl₃, respectively (van Sprang et al. 1978).

Cross-sections for vacuum ultraviolet emission from fluorine atoms transitions $(2p^43s)^{2,4}P \rightarrow (2p^5)^2P$ (95.5 and 97.5 nm wavelength) in CF₂Cl₂, SF₆, CF₄ and NF₃ were studied by Roque *et al.* (1991). The emission CS in CF₂Cl₂ compared to other fluorine-containing gases exhibits a more complex energy dependence. This indicates different molecular fragmentation patterns in CF₂Cl₂ than in the other gases.

The emission CS at 100 eV from CCl radicals amounts to $0.29 \cdot 10^{-22}$ m² (Jabbour and Becker 1989) for CF₂Cl₂ compared to $0.34 \cdot 10^{-22}$ m² for CFCl₃ and $0.86 \cdot 10^{-22}$ m² for CCl₄ (Kasukabe *et al.* 1993). Relative emission CS at 50–200 eV and lifetimes of excited molecular radicals in CF₂Cl₂ have been given by Martínez *et al.* (1992).

Ionization cross-sections have been measured by Märk and collaborators (Leiter et

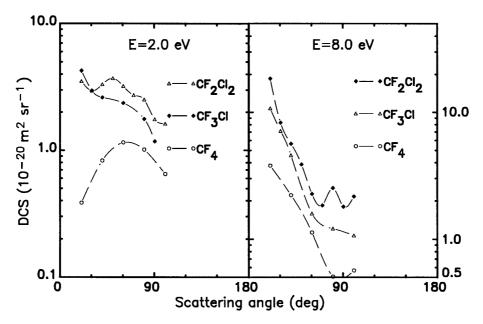


Fig. 13. – Differential cross-sections for elastic scattering on CF₄, CF₃Cl, CF₂Cl₂ at 2.0 and 8.0 eV. Experimental data from the measurements of Mann and Linder (see relevant subsections for references).

al. 1984 and 1989) for CCl_4 and CF_2Cl_2 , respectively. As noticed recently (Leiter et al. 1989), the early measurements from this group suffered from an underestimation error due to the non-complete collection of ions. Therefore, we have corrected the early measurements for CCl_4 by an energy-dependent factor. The correction has been based on the comparison between old (Stephan et al. 1985) and new (Poll et al. (1992) measurements for CF_4 . The corrected values of the ionization cross-section in CCl_4 are in good agreement with the results of Gryzinski's classical binary encounter approximation (see Leiter et al. 1984). The measurements of Pejčev et al. (1979) for CF_2Cl_2 are 40% higher than the renormalized results of Leiter et al. (1989).

If we adopt the corrected measurements of Leiter *et al.* (1984) different chloro- and fluoromethanes show a similar partitioning scheme: at 175 eV the ionization for CF_4 , CF_2Cl_2 and CCl_4 constitutes about 35% of TCS (Zecca *et al.* 1992). This is significantly less than, for example, in methane (about 45%).

A number of different ions were observed for CF_2Cl_2 with CF_2Cl^+ and Cl^+ amounting to 52% and 12%, respectively (Leiter *et al.* 1989) at 100 eV. For CCl_4 the CCl_3^+ (73%) and CCl_2^+ (12%) ions dominate (Leiter *et al.* 1984).

Dissociation processes in CF_2Cl_2 and $CFCl_3$ have been studied by detection of metastables by Allcock and McConkey (1978) but no absolute cross-sections were given.

Dissociative attachment. Chlorofluoromethanes, and in particular CCl₄ and CFCl₃, due to their ability to attach electrons at zero collision energy, were subject to numerous investigations. In detail, they were studied by swarm techniques (Christodoulides and Christophorou 1971, Christophorou et al. 1971, McCorkle et al. 1980, Spyrou and Christophorou 1985, Orient et al. 1989, Wang et al. 1998), crossed beams (Spence and Schulz 1973, Verhaart et al. 1978, Pejčev et al. 1979, Illenberger et al. 1979, Olthoff

et al. 1986, Chu and Burrow 1990, Wan et al. 1991, Kiendler et al. 1996, Hahndorf et al. 1994, Underwood-Lemons et al. 1995, Matejcik et al. 1995, Hahndorf and Illenberger 1997), rare-gas photoionization technique (Chutjian 1981, Chutjian et al. 1984, Chutjian and Alajajian 1985, Alajajian and Chutjian 1987, Chutjian and Alajajian 1987), flowing plasma afterglow (Smith et al. 1984, Smith and Španěl et al. 1994, Španěl et al. 1995), microwave radiolysis (Shimamori et al. 1992), and Rydberg atoms quenching (Foltz et al. 1977, Zollars et al. 1986, Harth et al. 1989, Ling et al. 1992, Popple et al. 1996, Finch et al. 1997). These techniques cover a wide energy range in a complementary manner: Rydberg atoms quenching the subthermal region (0.01 meV-1 meV, depending on the kind of Rydberg atoms chosen for detection), the photoionization up to 100 meV, swarms up to 1 eV.

Different methods yield, in general, congruent results for attachment coefficients, i.e. when the cross-section is averaged over an electron velocity distribution. However, detailed dependencies —both the absolute values and the positions of maxima in the attachment CS measured by different methods— show some discrepancies. A recent comparison for CF₂Cl₂, for example, was done by Wang et al. (1998). For this gas the photoionization method (Chutjian and Alajajian 1987) and a recent high-resolution beam experiment (Kiendler et al. 1996, Deniff et al. 1999) indicate a monotonic rise of the dissociative attachment CS in the zero-energy limit, while some beam and swarm experiments (Illenberger et al. 1979, McCorkle et al. 1980) show a near-to-zero-energy peak. Reviews of experimental methods have also been performed by Chutjian et al. (1996), for the Rydberg atoms quenching method by Dunning (1995), for beam techniques by Illenberger (1992), for flowing plasma afterglow by Smith and Španěl (1994).

Dissociative attachment coefficients (of about $3-4 \cdot 10^{-7}$ cm³ s⁻¹) in CCl₄ and CFCl₃ at room temperature are two orders of magnitude higher than for CF₂Cl₂ (Smith *et al.* 1984, and references therein). These high values reflect the rise of the attachment CS towards zero collision energy. For example, at 40 meV collision energy the attachment CS amounts to 260, 220 and $3.3 \cdot 10^{-20}$ m² in CCl₄, CFCl₃ and CF₂Cl₂, respectively (Alajajian and Chutjian 1987 and references therein).

A slight decrease of the dissociative attachment coefficients with rising temperature was observed for CCl₄ (Smith *et al.* 1984, Orient *et al.* 1989) and a moderate rise for CFCl₃ (Spence and Schulz 1973, Smith *et al.* 1984). A sharp rise (more than 10 times between 300 and 500 K) was observed for CF₂Cl₂ (Smith *et al.* 1984, Wang *et al.* 1998) and a rise by about a factor of two for CF₃Cl (Spyrou and Christophorou 1985, Hahndorf *et al.* 1994).

For CCl₄, the measurements by the high-state Rydberg potassium atoms quenching (Dunning 1987, Ling et al. 1992) indicate, within the experimental error, the applicability of Wannier's (Vogt and Wannier 1954) threshold law $\sigma_{\rm att}(E) \propto E^{-1/2}$ in the 0.01–1 meV energy range and Klots's (1976) simplified formula up to 10 meV. Dissociative attachment CS as high as $5 \cdot 10^{-16}$ m² have been reported at 0.01 meV (Ling et al. 1992). The lifetime of an intermediate CCl₄^{-*} excited ion has been evaluated to be 7.5 ps (Popple et al. 1996)

The data of Dunning and collaborators in CCl₄ (Foltz *et al.* 1977, Zollars *et al.* 1985, Dunning *et al.* 1987) merge well with the dissociative attachment CS at thermal energies obtained from high-resolution beam (Matejcik *et al.* 1995, 1997), threshold ionization (Orient *et al.* 1989, Chutjian and Alajajian 1985) and swarm unfolded data (Christodoulides and Christophorou 1971). A broad peak of about $3 \cdot 10^{-20}$ m² (Wan *et al.* 1991) for Cl⁻ ions was seen at 0.8 eV in beam (Olthoff *et al.* 1986, Chu and Burrow 1990, Oster *et al.* 1989) and swarm (Christodoulides and Christophorou 1971) experiments. A weak peak for the CCl₃⁻ formation was observed at 1.3 eV, while two peaks,

at 1.7 and 6.0 eV were observed for the CCl_2^- ion (Oster *et al.* 1989).

In CFCl₃, similarly as in CCl₄, a sharp rise of the negative ion yield in the zero-energy limit was also observed (McCorkle *et al.* 1980, Illenberger *et al.* 1979, Alajajian and Chutjian 1987). The absolute values of dissociative attachment CS are close to those in CCl₄ (Zollars *et al.* 1984, Shimamori *et al.* 1992). At energies of a few meV the attachment CS approaches the theoretical limit for s-wave scattering (Chutjian 1981, Orient *et al.* 1989). Measurements of velocity and angular distribution of negative ions produced through electron transfer in collisions with velocity-selected potassium Rydberg atoms were done by Finch *et al.* (1997). The lifetime of the intermediate excited CFCl₃^{*} was established as about 3.5 ps.

A peak in CFCl₃ attachment CS of about $22 \cdot 10^{-20}$ m² was observed at 0.18 eV in a swarm experiment (McCorkle *et al.* 1980). Additional peaks, about three orders of magnitude weaker, were observed at 1.6 eV and 3.3 eV with Cl_2^- and F^- ions dominating, respectively (Illenberger *et al.* 1979, Oster *et al.* 1989).

In CF₂Cl₂ Pejčev et al. (1979) observed only a large structure at low energies, probably due to a limited energy resolution (200 meV) of their beam. McCorkle et al. (1980) in a swarm experiment evidenced three maxima: at 0.07, 0.30 and 0.95 eV. The latter two peaks agree roughly with the swarm-derived dissociative attachment CS of Christophorou et al. (1974), Petrović et al. (1989) and the beam experiment of Underwood-Lemons (1995) who noticed peaks at 0.2 eV and 1.1 eV. The near-to-zero peak of McCorkle et al. was observed in other experiments as a continuous rise of the CS with lowering energies. Kiendler et al. (1996) argued that the dissociative attachment CS at near-to-zero energies changes as 1/E, indicating the s-wave character of electron capture. The rise in CF₂Cl₂ is less pronounced than in CCl₄ and CF₃Cl (Chutjian and Alajajian 1987, Christophorou et al. 1974). At high temperatures the zero-energy attachment CS rises, at the expense of the 0.2 eV peak (Underwood-Lemons et al. 1995, Kiendler et al. 1996, Hahndorf and Illenberger 1997). In fig. 9 we present measurements of Underwood-Lemons et al. at 393 K only.

At the 1 eV maximum the CF_2Cl_2 attachment CS amounts to about $0.9 \cdot 10^{-20}$ m² (see Illenberger *et al.* 1979 and references therein). A maximum, an order of magnitude weaker, was observed also at 3.0–3.5 eV (Pejčev *et al.* 1979, Verhaart *et al.* 1978, Underwood-Lemons *et al.* 1995, Oster *et al.* 1989). Many ions (Cl⁻, F⁻, CFCl⁻₂, FCl⁻, Cl⁻₂) contribute to the dissociative attachment CS at this maximum.

In CF₃Cl two peaks at about 1.4 eV and 4.6 eV, with approximately the same intensity $(1 \cdot 10^{-22} \text{ m}^2)$, Spyrou and Christophorou 1985) are observed in the total negative ion yield (McCorkle *et al.* 1980, Spyrou and Christophorou 1985, Underwood-Lemons 1995, Hahndorf *et al.* 1994). At 4.8 eV the Cl⁻ yield dominate (Illenberger *et al.* 1979); FCl⁻ and F⁻ ions were observed at about 4.0 eV (Oster *et al.* 1989, Hahndorf and Illenberger 1997). At high gas temperatures (800 K) a near-to-zero-energy peak in the Cl⁻ yield appears, due to the change in the reactivity mechanism from endothermic to exothermic (Hahndorf *et al.* 1994). The rise of gas temperature enhances and shifts the 1.4 eV peak towards lower energies (compare also with the classical approach of Lehr and Miller 1996).

Relative intensities for negative fragments in chlorofluoromethanes (CCl₄, CFCl₃, CF₂Cl₂, CFCl₃, CF₄) scale as 1000, 800, 50, 2, (0) for the Cl⁻ ion and as (0), 4, 6.5, 0.4, 10 for the F⁻ ion, respectively (Oster *et al.* 1989).

Resonances. Several shape resonances are visible in TCS between 1.0 and 10 eV for the considered molecules. The number of such resonances rises with lowering the symmetry of the target: two resonances for CCl_4 , three for $CFCl_3$ and CF_3Cl and four for CF_2Cl_2 ,

see also fig. 12. We note, furthermore, that the corresponding peaks in the TCS rise and move to lower energies when the total atomic mass of the molecule increases. A common feature of these resonances is that the maxima in the dissociative attachment CS are shifted, usually by 0.4-0.5 eV, towards lower energies compared to the maxima in total and/or vibrational excitation CS (see also Mann and Linder 1992a, b).

Some disagreement exists for molecular orbitals involved in the formation of resonant states at higher energies. A general accord exists on the fact that the C-Cl σ^* orbitals are involved in the lowest-energy resonances and that they are stabilized by a fluorine atom (Burrow *et al.* 1982). But higher resonances, located above 2 eV and assigned in vibrational excitation measurements on CF₃Cl and CF₂Cl₂ (Mann and Linder 1992a, b, respectively) to C-F σ^* orbitals, could be either of the π^* type or of a mixed type (Underwood-Lemons *et al.* 1994, 1995).

In CF₃Cl the existence of a low-energy resonance (or even two of them, one of a Rydberg and one of a valence type) in the 0–2eV region was claimed by Verhaart *et al.* (1978) in their electron transmission experiment. A peak at 1.4 eV observed in the electron attachment (Verhaart *et al.* 1978, McCorkle *et al.* 1980) is shifted by -0.5 eV compared to the maximum in TCS (Jones 1986) and in the vibrational CS (Mann and Linder 1992a). The A_1 (C-Cl σ^*) character of this resonance (Underwood-Lemons 1994) leads to the excitation of the C-Cl symmetric stretching vibrations (Mann and Linder 1992a) and to the Cl⁻ ions production (Illenberger *et al.* 1979).

In CF₃Cl the attribution of the resonance at 5.5 eV, with the E symmetry (Mann and Linder 1992a) to a specific molecular orbital (C-F σ^* , Cl 3p or C-Cl π^*) is not clear (see Underwood- Lemons et al. 1994 and 1995). We recall that numerous types of ionic fragments appear between 3.9 eV and 4.8 eV (Illenberger et al. 1979). Similarly, the 8.5 eV peak in TCS, attributed to the A_1 -symmetry resonance, can involve either the C-F σ^* -orbital, as suggested by vibrational excitation studies (Mann and Linder 1992), or the C-F π^* -orbital as suggested by the calculations of Underwood-Lemons et al. (1994).

In CF₂Cl₂ the studies of electron attachment (Illenberger 1979, 1982, 1992), electron transmission spectra (Verhaart *et al.* 1978, Burrow *et al.* 1982) and vibrational excitation (Mann and Linder 1986b) allow to evidence four valence-excited resonances of the A_1 (C-Cl σ^*), B_2 (C-Cl σ^*), A_1 (C-F σ^*) and B_1 (C-F σ^*) type at energies, respectively, 1.0 eV, 2.5 eV, 4 eV and about 6 eV (Mann and Linder 1992). The B_2 resonance, visible in the vibrational excitation channel (Mann and Linder 1992b), apparently does not lead to the production of negative ions. We note that the theoretical assignment (Underwood-Lemons *et al.* 1995) of intermediate B_2 and A_1 resonances to the π^* -type orbitals differs from that of Mann and Linder (1992 a and b).

For CFCl₃ three maxima are observed in TCS, at 1.7 eV, 4 eV and 8 eV (Jones 1986). The calculations of Burrow *et al.* (1982) suggest that the peak at 1.7 eV in the TCS (and at 1.5 eV in the electron attachment CS) is associated with the occupation of the e (C-Cl σ^*) orbital while the 4.0 eV peak is associated with the a_1 (C-F σ^*) orbital. In comparison to the lighter molecule CF₃Cl the analogous resonances in CFCl₃ are shifted towards lower energies.

In CCl_4 only two maxima are observed in TCS, at 1.2 eV and 7.5 eV (Jones 1996), *i.e.* lower in energy than the corresponding structures in $CFCl_3$. The 1.2 eV maximum has a negative-ion precursor at 0.8 eV (Chu and Burrow 1990).

Sum check. Due to the lack of absolute measurements of partial CS no sum check is possible for chlorofluoromethanes. Several measurements at low energies are not absolute and require the use of normalization procedures.

The measurements of Wan et al. (1988) for CCl₄ and of Underwood-Lemon et al. (1994)

for CF_3Cl and CF_2Cl_2 indicate the rise of TCS at energies below 0.5 eV. For CF_3Cl and CF_2Cl_2 this rise could point out towards the existence of a Ramsauer-Townsend minimum, as indicated also by energy spectra in the differential DCS measurements (Mann and Linder 1992a, b). Additional peaks below 1 eV were observed for both gases in a magnetically-guided-beam experiment (Randell *et al.* 1993), see figs. 8 and 9. However, we are not able to attribute any partial process to these peaks.

For CFCl₃ and CCl₄ the sharp rise in TCS is due to the contribution of the electron attachment. In CCl₄ a good merging between the normalized TCS of Randell *et al.* (1993) and recent electron attachment CS both at 0.01 eV (Ling *et al.* 1992, Dunning 1987, Matejcik *et al.* 1995), as well as at 0.05 eV (Chutjian and Alajajian 1985) has to be noticed. These, as well as swarm experiments (Christodoulides and Christophorou 1971), indicate that the energy dependence of the electron attachment CS changes from $E^{-1/2}$ in the zero-energy limit to E^{-1} at energies above 0.1 eV. However, some differences in absolute magnitudes, probably caused by specific normalization procedures applied in the evaluation of attachment coefficients (see Spanel *et al.* 1995), still exist, see figs. 10 and 11.

For CFCl₃ the agreement between normalized TCS (Randell *et al.* 1993) and electron attachment CS (McCorkle *et al.* 1982, Zollars *et al.* 1984, Dunning 1987) is poor. A small (5 meV) energy shift in TCS measurements would explain the difference between the two data sets.

At energies of the order of 100 eV the sum check in CCl₄ indicates an incoherence between the elastic CS of Daimon *et al.* (1984) normalized to the independent-atoms model and the TCS (Szmytkowski *et al.* 1992, Zecca *et al.* 1992). Intermediate-energy absolute partial CS measurements would be of interest for all the gases considered here.

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ZOLLARS B. G., WALTER C. W., LU F., JOHNSON C. B., SMITH K. A. and DUNNING F. B., J. Chem. Phys., 84 (1986) 5589. 3.3. Methyl halides (CH₃F, CH₃Cl, CH₃Br, CH₃I) and other halomethanes. – In this subsection other derivatives of methane will be discussed. In particular, relatively more electron scattering measurements and calculations were performed for the methyl halides CH₃X, fluoromethanes CH_nF_{4-n} and chloromethanes CH_nF_{4-n}, with n=1-3 (CH₄, CF₄ and CCl₄ have been discussed before) than for other halomethanes. We will present these three groups of gases together. Some of the most significant results in other halomethanes will also be mentioned. At the end of this subsection, a few works on other methyl compounds (like CH₃OH) are quoted. For the lack of data no sum check neither separate presentation of partial CS is done for all these gases. Instead, we present in somewhat more detailed way comparative measurements of partial processes. In fig. 14 the total cross-sections for four methyl halides CH₃X (X = F, Cl, Br, I) are compared in the energy range up to 20 eV. In tables VI and VII some information on electron attachment in selected halomethanes is compiled.

The four methyl halide molecules possess high dipole moments of about 1.8 D. In the series (CH₃F, CH₃Cl, CH₃Br, CH₃I) the C-H bond length remains constant (1.1 Å), while the C-halogen atom bond length rises. Chloromethanes play an important role in atmospheric chemistry (Wallington *et al.* 1996). Electron interactions with trifluoromethane CHF₃ have recently been reviewed by Christophorou *et al.* (1997).

Total cross-sections for methyl halides have been measured by the Gdańsk group (Szmytkowski and Krzysztofowicz 1993 for CH₃I, Krzysztofowicz and Szmytkowski 1994 for CH₃Br, Krzysztofowicz and Szmytkowski 1995 for CH₃F, CH₃Cl) in the 0.5–200 eV region using a linear transmission method and by the Maryland group (Benitez et al. 1988—all the four molecules—and Wan et al. 1991 for CH₃Cl) in the 0.5–7.5 eV region using a trochoidal spectrometer. The measurements from the two groups agree generally in shape but not in the absolute values: the difference is especially serious for CH₃Cl: at 7.5 eV the data of Benitez et al. is 30% lower than that of Krzysztofowicz and Szmytkowski (1995). On the other hand, the measurements of Krzysztofowicz and Szmytkowski in their highenergy limit, 75–200 eV, are 20% lower than the results of Karwasz et al. (1999). Also the

Table VI. – Electron attachment cross-section in chloromethanes. Values of maxima in the dissociative attachment cross-section (in 10^{-20} m²) from beam measurements of Burrow and collaborators. In square parentheses the energies of the peaks (in eV).

Compound	Zero-energy *	Second maximum	Feshbach resonance
CH ₃ Cl	* *	$2.0 \cdot 10^{-5} [0.8] P$	$4.9 \cdot 10^{-4} [7.4] P$
$\overline{\mathrm{CH_2Cl_2}}$	$8.9 \cdot 10^{-3}$ C	$2.4 \cdot 10^{-2} [0.48] \text{ C}$	
CHCl ₃	4.7 C	3.7 [0.27] C	
CCl ₄	$1.3 \cdot 10^2$ C	1.9 [0.80] C	

^{*} The zero-energy peak is convoluted with the electron-beam energy-distribution of the experiment

C - Chu and Burrow (1990)

P - Pearl and Burrow (1993)

^{**} A signal observed in experiments of Chu and Burrow (1990) was due to impurities (Pearl and Burrow 1993)

TABLE VII. – Electron attachment cross-sections (a) and electron attachment rate coefficients (b, c) in some halocarbons and SF₆. (a) Near-to-zero-energy dissociative attachment cross-sections (in 10^{-20} m²) from beam experiments. (b) Attachment rate coefficients (in 10^{-7} cm³ s⁻¹) at 300 K from pulse radiolysis method. (c) Attachment rate coefficients at a mean electron energy of 10 meV (Rydberg Xe atoms with principal quantum number n = 37), in 10^{-7} cm³ s⁻¹.

Compound	a)	b)	c)
CHCl ₃	7.2 M	0.02 Sb	
CFCl ₃		1.8 Sb	5.8 F 6.5 D
CCl ₄	$1.8 \cdot 10^3 \text{ M}$ $5.0 \cdot 10^3 \text{ HO}$	4.0 Sb	4.0 F 7.0 D
CH_2Br_2		1.1 Sb	65 *L
CH ₃ I	$4.0 \cdot 10^3$ SC	1.2 Sa	4.4 H 20 *L
CF ₃ I		2.0 Sa	
$1,1,1-C_2Cl_2F_3$		1.1 Sb	6.0 D
SF ₆		2.3 Sb	4.0 W 4.0 D

^{*} The attribution of the total quenching rate to the electron attachment is not clear

two sets of data from the Maryland laboratory for CH₃Cl (Benitez *et al.* 1988 and Wan *et al.* 1991) differ by about 15%. Probably, these discrepancies are related to experimental problems in the trochoidal apparatus.

The TCS of all the four methyl halides exhibit a maximum at about 6.5–7 eV, a minimum at about 2–4 eV and a rise in the low-energy limit. The absolute values of the TCS rise passing from CH₃F to CH₃I, see fig. 14.

The fall of TCS in methyl halides between 0.5 and 1 eV (Szmytkowski and Krzysztofowicz 1993, Krzysztofowicz and Szmytkowski 1994, 1995) has an energy dependence closer to $E^{-1/2}$ than to E^{-1} so that it cannot be attributed directly to scattering on a dipole moment (see Itikawa 1978, Norcross and Collins 1982). Also electron attachment does not seem to be the dominant process below 1 eV. For CH₃I, different evaluations (Alajajian *et al.* 1988, Schramm *et al.* 1999) give attachment CS values below $10 \cdot 10^{-20}$ m² at 0.1 - 0.5 eV, while the TCS measurement quoted a value of $96 \cdot 10^{-20}$ m² at 0.45 eV (Szmytkowski and Krzysztofowicz 1993).

D - Dunning (1987), several types of Rydberg atoms with $n \sim 100$

F - Foltz *et al.* (1978), n = 37

 $[\]mathsf{H}$ - Hildebrandt et al. (1978), n=37

 $[\]mathsf{HO}$ - Hotop $et~al.~(1995),~\mathrm{CS}$ at $1~\mathrm{meV}$

M - Matejcik et al. (1997), CS at 5 meV

L - Ling et al. (1993) potassium Rydberg atoms with n = 400 (about 80 μeV collision energy)

Sa - Shimamori et al. (1992a)

Sb- Shimamori et al. (1992b)

SC - Schramm et al. (1999), CS at 0.5 meV

W - West *et al.* (1976), n = 38

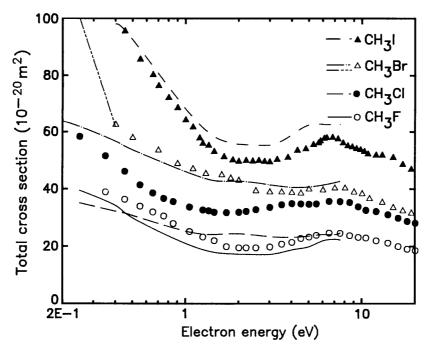


Fig. 14. – Total cross-sections for methyl halides: linear transmission technique (points), Szmytkowski and Krzysztofowicz (1993), CH₃I; Krzysztofowicz and Szmytkowski (1994); CH₃Br Krzysztofowicz and Szmytkowski (1995); CH₃Cl, CH₃F; trochoidal spectrometer (lines), Benitez et al. (1988); not shown measurements of Wan et al. (1991) for CH₃Cl from the same laboratory. Momentum transfer cross-section (analytical formula) of Petrović and Crompton (1988) for CH₃Br, dash-dot-dot line.

TCS in trifluoromethane (CHF₃) were measured by Sueoka *et al.* (1998) in the 0.7–600 eV range and by Sanabia *et al.* (1998) in a trochoidal spectrometer up to 20 eV. The TCS in CHF₃ exhibits analogies with both CH₄ and CF₄ (Sueoka *et al.* 1998): between 5 and 15 eV it coincides with the CH₄ measurements from the same laboratory and shows another large hump between 20 eV and 30 eV, similar to that in CF₄ (see subsect. **3**·1 for references). Below 3 eV a sharp rise of the TCS has been observed, attributed to the "s-wave scattering" (Sanabia *et al.* 1998). The TCS in the 300–1000 eV energy range was measured in a linear transmission experiment by Iga *et al.* (1999).

TCS in chloromethanes CH₃Cl, CH₂Cl₂, CHCl₃ were studied by Holst and Holtsmark (1931) between 0.3 and 25 eV, by Wan et al. (1991) between 0.2 and 12 eV and by Karwasz et al. (1999) between 75 and 4000 eV. Measurements of Holst and Holtsmark indicated a shape of TCS in chloromethanes similar to that in chlorofluoromethanes, i.e. with several maxima between 1 and 10 eV (compare fig. 12). Two of these maxima in the measurements of Holst and Holtsmark are quite distinct for the whole CH₃Cl, CH₂Cl₂, CHCl₃, CCl₄ series. The positions of these maxima move to lower energies and their amplitudes rise on going from CH₃Cl to CCl₄. The first maximum moves from $13 \cdot 10^{-20}$ m² at about 3 eV to $31 \cdot 10^{-20}$ m² at 1.7 eV, and the second one from $26 \cdot 10^{-20}$ m² at 7.3 eV to $36.5 \cdot 10^{-20}$ m² at 6.2 eV, for CH₃Cl and CHCl₃, respectively. The amplitudes of the maxima in the measurements of Wan et al. are less pronounced; note that in C₆H₆

the high amplitudes of the structures seen by Holst and Holtsmark are not confirmed by more recent experiments. According to Wan et al. the TCS in CH₃Cl, CH₂Cl₂, CHCl₃ at 2–12 eV are comprised between $20 \cdot 10^{-20}$ m² and $40 \cdot 10^{-20}$ m². All these TCS exhibit a rise at low energies. Karwasz et al. (1999) in their high-energy measurements observed increasing values of TCS for the CH₄, CH₃Cl, CH₂Cl₂, CHCl₃, CCl₄ series. They indicated a scaling of TCS with the molecular polarizability at intermediate (close to 100 eV) energies and the applicability of the additivity rule for TCS in the 4000 eV energy limit.

TCS in other halomethanes (CF₃Br, CF₂Br₂, CF₃I) at 0.2–12 eV were studied by Underwood-Lemons *et al.* (1994).

Swarm experiments. The drift velocity in trifluoromethane CHF₃ at the reduced electrical field up to $3.69 \cdot 10^{-21}$ Vm² was measured by Christophorou et al. (1981) and up to $80 \cdot 10^{-21}$ Vm² by Wang et al. (1999). The drift velocity is a monotonically rising function of the reduced field amounting to $7.25 \cdot 10^6$ cm s⁻¹ at $80 \cdot 10^{-21}$ Vm². The attachment coefficient in methyl chloride CH₃Cl and in methyl bromide CH₃Br up to the reduced fields of $6.2 \cdot 10^{-21}$ Vm² were measured in a swarm experiment by Datskos et al. (1990 and 1992, respectively). Diffusion and attachment coefficients for thermal electrons in CH₃Br were determined by Petrović and Crompton (1987).

Elastic scattering. Effects of fluorination in the CH₄, CH₃F, CH₂F₂, CHF₃, CF₄ fluoromethane series were studied by Tanaka et al. (1997). At 1.5 eV, the CH₄ and CF₄ elastic DCS are similar in shape (compare fig. 3 in part II and fig. 13 in the present subsect. 3.2) but scattering on intermediate fluoromethanes CH₃F, CH₂F₂, CHF₃ is forward centered, due to their permanent dipole moment. At 30 eV all the five targets show a minimum of the DCS at 100°-110°, but differently from CH₄ all the fluoromethanes show an additional shoulder at 60°. The amplitude of this latter structure rises with the number of fluorine atoms. Such "interference" effects in DCS, due to the presence of heavier atoms in the molecule, were observed, for instance, in the CO₂, OCS, CS₂ series (see fig. 18 in part II).

DCS for CH₂F₂ at 300, 400, 500 eV were measured by Bromberg (1975). Experimental differential and integral elastic CS for CHF₃ at 300–1000 eV were announced by Iga *et al.* (1999). Theoretical DCS for CHF₃ at 6.5–30 eV were given by Diniz *et al.* (1999); differential and integral CS for CH₂F₂ at 3–30 eV by Nishimura (1998); DCS for CH₃Cl at 0.5–8 eV by Rescigno *et al.* (1997).

Elastic scattering in CH₃Cl was examined between 0.5 and 9.5 eV by Shi *et al.* (1996). At energies up to 1 eV the elastic DCS follows the Born approximation as calculated for rotational excitation; at 3.2, 5.5 and 8 eV it exceeds the Born values, but is still forward centered. The DCS for 100° scattering angle shows a broad maximum at about 3.5 eV, the DCS at 30° shows a minimum at 3 eV and a rise up to the highest (9.5 eV) energy examined. At 8.0 eV and 90° scattering angle the CH₃Cl (Shi *et al.* 1996) and CH₄ (Boesten and Tanaka 1991) DCS are equal.

DCS at 90° in CH₂Cl₂ between 1.0–3.5 eV has been reported by Abouaf and Benoit (1990). The DCS shows a broad maximum at 1.5 eV and a slow fall between 2.5 and 3.5 eV.

DCS in CF₃I at 20° – 120° angles, 1.5–60 eV were reported by Okamoto *et al.* (1999). As stated by these authors, the effects of scattering from F and C atoms are rather weak, while contribution from the I atom dominates the DCS. The DCS at all studied energies show a shallow minimum at 30° – 50° , a clear hump around 90° and a sharp dip at 120° . The elastic DCS at 100° reaches its maximum value at 4–6 eV. A maximum in the vibrational excitation DCS at 60° for a 0.14 eV energy loss occurs at 5.5 eV and is

preceded by a deep minimum at 3 eV.

Intermolecular interference effects in elastic scattering of 1–4 keV electrons on the oriented methyl halides CH_3Cl and CH_3Br were studied at Bielefeld (by Böwering et al. 1994, Meier et al. 1994 and Volkmer et al. 1997 and by Volkmer et al. 1992, 1996, respectively). Effects of halogenation on low-energy elastic scattering in CH_nF_{4-n} , CH_nCl_{4-n} , CF_nCl_{4-n} , with n=1,2,3, were calculated by Natalense et al. (1999).

Vibrational excitation in CH₃Cl was examined by Shi et al. (1992, 1996). The excitation function at 100° scattering angle for the ν_3 vibrational mode (C-Cl stretch) shows a maximum at 3.2 eV; the excitation function for the ν_4 mode (CH stretch) shows a maximum at 5.5 eV (Shi et al. 1992). At 3.5 eV collision energy, apart from the ν_3 mode and its overtones, the ν_4 and ν_5 (CH₃ d-deform) modes with an intensity of about 0.2 times the ν_3 intensity and other modes with smaller intensity are visible (Shi et al. 1996). At 5.5 eV the intensity of the ν_4 mode amounts to 1.2 of the ν_3 intensity and the ν_1 (CH s-stretch), ν_5 , ν_2 (CH₃ s-deform) and ν_6 (CH₃ rock) modes are excited with an intensity of about 0.7 of the ν_3 intensity. At 8.5 eV the intensity of the ν_2 , ν_6 and ν_4 modes exceed the ν_3 intensity by more than three times and that of ν_1 and ν_5 exceeds it by almost two times. The DCS for the ν_3 mode at 3.2 eV and for the ν_4 mode at 5.5 eV are almost uniform in angle. At 100° the ν_3 DCS is about 6% of the elastic DCS. A theoretical analysis of vibrational excitation and dissociative attachment in CH₃Cl was done by Fabrikant (1991, 1994).

Vibrational excitation in CH₂Cl₂ at 1.0–3.5 eV collision energy was studied by Abouaf and Benoit (1990). Excitation of the ν_3 (CCl₂ s-stretch) and its overtones up to $3\nu_3$ was observed both at 1.3 and 3.2 eV, with some excitation of the CH stretch modes as well. The DCS at 1.3 and 2.8 eV for the ν_3 and $2\nu_3$ modes are almost uniform at 40° – 90° scattering angles and rise below 40° . Abouaf and Benoit argued that according to selection rules these modes indicate an A_1 symmetry of the resonant states and that the B_2 resonance predicted by the theory at 2.5 eV (Burrow et al. 1982) can be responsible for the small rise in DCS observed at low angles. The enhancement of vibrational excitations is visible in the whole energy range studied by Abouaf and Benoit (1990).

Rotational excitation. Due to the high dipole moments of methyl halides, the rotational de-excitation and excitation should contribute a significant part of the TCS at low energies. We are not aware of measurements or calculations for the rotational excitation CS. In CH₃Cl the rise of the DCS at low angles has been attributed (Shi et al. 1996) to rotational excitation. Experiments using Rydberg atoms quenching (Hildebrandt et al. 1978, Ling et al. 1993) indicate that the TCS for CH₃I (and CH₂Br₂) at sub-thermal energies follow the E^{-1} dependence: this could correspond to rotational de-excitation and excitation. However, some discrepancies with the Born approximation were stressed (Ling et al. 1993). On the other hand, Schramm et al. (1999) explained the very low energy rise of TCS in CH₃I (Ling et al.) as due to electron attachment.

Electronic excitation. Forward-scattering electron excitation in CH₃F, CH₃Cl, CH₃Br, CH₃I was studied by Olney et al. (1994, 1996, 1997, 1998) in 7–250, 6–350, 6–450, 4.5–488 eV energy loss ranges, respectively. The photoabsorption CS in these gases, as measured with a low (1 eV FWHM) energy resolution, shows a single, wide peak. For example, in CH₃Cl a maximum of $0.80 \cdot 10^{-20}$ m² at 16 eV and in CH₃Br of $0.88 \cdot 10^{-20}$ m² at 15 eV were observed.

Optical emission. Emission CS in CHF₃ due to atomic and diatomic fragments as well as emission in a continuous band at 200–400 nm were given in the 100–2000 eV collision-energy range by van Sprang et al. (1978). The intensity of atomic hydrogen lines and of the CH radical de-excitation lines at 0.4–5 keV impact energy was studied in

CHF₃ by Danilevskii et al. (1986). Emission from CHF₃ in the ultraviolet (50–130 nm) range was studied up to 600 eV incident energies by Wang and McConkey (1989); in the 130–520 nm range and at energies up to 1000 eV by Aarts (1985). Emission from CHF₃ in the 200–400 nm spectral range due to electron and UV (11–35 eV) photon excitation was studied by Creasey et al. (1990). Electron-collision emission from CH₃F, CH₂F₂ and CHF₃ up to 1000 eV collision energy was measured in the spectral region from 100 to 700 nm by Keller et al. (1996, 1998); emission from CH₃F, CH₂F₂, CHF₃ and CF₄ in the 600–900 nm spectral region and up to 500 eV collision energy by Torres et al. (1999); emission from CF₃I by Martínez et al. (1998).

Emission from CCl fragments formed in electron collisions with $CXCl_3$ (X = H, F, Cl, Br) was studied up to 100 eV by Ito *et al.* (1986) and Kusakabe *et al.* (1993). Emission cross-sections rise by factors of two from CHCl₃ to CFCl₃ and to CBrCl₃.

Ionization CS in CH₃F, CH₃Cl, CH₃Br (and CH₄) were measured between threshold and 200 eV by Vallance et al. (1997). The maximum ionization CS increases for heavier targets, amounting to 3.73, 6.91, 8.02, 10.3, $12.25 \cdot 10^{-20}$ m² for CH₃F, CH₃Cl, CH₃Br, CH₃I and CHCl₃, respectively. Only the CH₄ ionization CS does not follow this rule, being higher $(4.24 \cdot 10^{-20} \text{ m}^2)$ than that for CH₃F (Vallance et al. 1997).

Ionization of CHF₃ was studied in a detailed way by Goto *et al.* (1994). The total ionization CS amounts to $6.74 \cdot 10^{-20}$ m² at 120 eV. The CF₃⁺ ion is formed with 33% probability at this energy and CF⁺ with 43%.

Ionization from space-oriented CH₃Cl molecules was studied by Kasai *et al.* (1993) and from CH₃Cl, CHCl₃ and CH₃Br by Aitken *et al.* (1995). Asymmetry effects in the ionization of CH₃Cl are more pronounced for the formation of the CH₃Cl⁺ ion than for the CH₃ ion (Aitken *et al.*). We are not aware of any theoretical or semiempirical values of ionization CS in halomethanes (except for CF₄).

Dissociation into neutrals in CHF₃ up to 600 eV was measured by Winters and Inokuti (1982). The total dissociation CS maximum in CHF₃ ($5.8 \cdot 10^{-20}$ m²) is slightly higher than the maximum CS in CF₄ ($5.5 \cdot 10^{-20}$ m²). Motlagh and Moore (1998) studied dissociation CS in CH₃F, CH₂F₂, CHF₃ (and CH₄, CF₄, C₂F₆, C₃H₈). They observed that the partial CS for dissociation are proportional to statistical weight of pathways leading to the formation of a specific radical. For example, the CS section for formation of the CF₃ radical from the CHF₃ molecule is 1 /₄ of the CS for formation of CF₃ from CF₄. These proportions can be considered as a manifestation of a kind of "additivity" rule for partial cross-sections, predicted for total cross-sections (Zecca et al. 1999, Karwasz et. al. 1999).

In the case of dissociation in fluoromethanes, Motlagh and Moore reported maxima of $1.5 \cdot 10^{-20}$ m² for the CF₃ radical formation from CF₄, $0.9 \cdot 10^{-20}$ m² from CHF₂ and $0.4 \cdot 10^{-20}$ m² from CHF₃; $0.7 \cdot 10^{-20}$ m² for CHF₂ formation from CH₂F₂ and $0.5 \cdot 10^{-20}$ m² for CH₂F formation from CH₃F. All maxima occur at about 100 eV collision energy. In CHF₃ data of Motlagh and Moore (1998) are in good accord with 20–200 eV measurements of Goto *et al.* (1994). The latter authors reported 0.49, 0.032, 0.092, 0.006 and $0.18 \cdot 10^{-20}$ m² for the formation of CF₃, CHF₂, CF₂, CHF and CF, respectively, from CHF₃ at 100 eV. However, successive measurements from this group (Sugai *et al.* 1995), similarly to those for CF₄, give values by a few folds lower.

Electron attachment in methyl halides CH₃X leads to the formation of dissociated X⁻ negative ions. The total yield of negative ions at low energies exhibits a strong, rising dependence on the gas temperature. This phenomenon was studied by Spence and Schulz (1973) for CHCl₃, CH₃Br, CH₃I, (and CFCl₃, CF₃Br); by Smith *et al.* (1984) for CHCl₃ (and CF₂Cl₂, CFCl₃, CCl₄); by Datskos *et al.* (1990), Pearl and Burrow (1993)

and Španěl *et al.* (1995) for CH₃Cl; by Petrović and Crompton (1987) and Datskos *et al.* (1992) for CH₃Br. Among different gases electron-attaching at zero-energy, only for CCl₄ and SF₆ no dependence of the CS on the gas temperature was noticed. A review of results on electron attachment at thermal energies was done by Smith and Španěl (1994).

In general, several peaks in negative ions yield are observed in halomethanes: a near-to-zero-energy rise, due to positive electron affinity of these compounds, a second peak below 1 eV and other/others at energies of Feshbach resonances (see, for example, Othoff et al. 1986 for CCl₄, CCl₃Br, CCl₂Br₂, CBr₄). However, large discrepancies regarding both the positions and the amplitudes of the peaks can be introduced by gas impurities and/or reactions in the experimental systems (compare Chu and Burrow 1990 and Pearl and Burrow 1993, 1994).

Electron attachment in CHF₃ at thermal electron energies is very weak or absent. A small electron attachment rate constant ($\sim 13 \cdot 10^{-14} \text{ cm}^3 \text{ s}^{-1}$) measured in swarm experiment can be due to impurities (Wang *et al.* 1999).

Dissociative attachment in chloromethanes CH_3Cl , CH_2Cl_2 , $CHCl_3$, CCl_4 was measured by Chu and Burrow (1990) and Pearl and Burrow (1993) in a beam experiment. Their results are reported in table VI. For methyl chloride CH_3Cl , Pearl and Burrow (1993) argued that the zero-energy CS seen in their previous experiment (Chu and Burrow 1990) was due to an impurity produced by a chemical reaction on the hot filament. A weak and broad peak of $4.9 \cdot 10^{-24}$ m² is centered at 0.8 eV (Pearl and Burrow 1993). Swarm measurements yielded a maximum attachment coefficient of $0.4 \cdot 10^{-12}$ cm³ s⁻¹ in CH_3Cl at thermal energies (Petrović et al. 1989).

In CHCl₃, Chu and Burrow (1990) reported a value of $4.7 \cdot 10^{-20}$ m² for the dissociative attachment CS at zero energy. Another peak was reported at 0.27 eV (Chu and Burrow 1990, Wan et al. 1991). Electron attachment in CHCl₃ was also measured down to 1 meV in a recent beam experiment (Matejcik et al. 1997). Down to 5 meV the CS rises following an E^{-1} dependence; for lower energies it remains constant at about $7.2 \cdot 10^{-20}$ m². As stated by Matejcik et al. (1997), for electron energies between 20 and 130 meV the electron attachment proceeds via the electron s-wave capture, whereas at higher energy (above 0.4 eV) the autodetachment plays a significant role. Note, however, that the measured CS energy dependence (Matejcik et al. 1997) is roughly two orders of magnitude below the de Broglie limit for the s-wave capture (see Wigner 1948). This conclusion is also confirmed by flowing-afterglow and microwave measurements: at 300 K the attachment rate coefficient is by a factor of 100 lower in CHCl₃ than in CCl₄ and CFCl₃ (Smith et al. 1984, Shimamori et al. 1992b).

The early beam experiment on CH₃Br reported a Br⁻ peak at about 0.35 eV (Stockdale et al. 1974). The existence of this peak, increasing in amplitude and moving to lower energies with rising gas temperature, was confirmed also by swarm experiments (Wang and Lee et al. 1988, Datskos et al. 1992). An electron attachment rate constant of $10.8 \cdot 10^{-12}$ cm³ s⁻¹ has been reported at thermal energies (Datskos et al. 1992), in overall agreement with other measurements (Alge et al. 1984, Petrović and Crompton 1987). Earlier measurements (Christodoulides and Christophorou (1971) seem to be overestimated. Weak maxima were observed at about 6.9 and 9.4 eV (Stockdale et al. 1974).

For CH₃I early measurements (Stockdale *et al.* 1974) reported a single peak for I⁻ production at 0.15 eV. Other experiments (Shimamori *et al.* 1988, Shimamori *et al.* 1992 a, b, microwave technique; Hildebrandt *et al.* 1978 and Ling *et al.* 1993, Rydberg's atoms quenching; Christophorou 1976, swarm; Alge *et al.* 1984, flowing afterglow; Alajajian *et al.* 1988, threshold photoionization) indicate a rather continuous rise at the zero-energy

limit. An electron attachment rate of the order of $300 \cdot 10^{-10}$ cm³ s⁻¹ at about 0.1 eV was reported by Shimamori and Nakatani (1988) and Shimamori *et al.* (1992a) and of $1200 \cdot 10^{-10}$ cm³ s⁻¹ at 300 K by Alge *et al.* (1994), see also table VII.

Alajajian et al. (1988) were the first to report a weak maximum in the CH₃I electron attachment CS at 60 meV, superimposed on the low-energy rise. In a recent high-resolution experiment performed down to 0.5 meV collision energy, Schramm et al. (1999) noticed than this maximum occurs at the vibrational excitation threshold of the ν_3 mode. They also observed other resonant structures in the dissociative electron attachment CS (I⁻ ion formation) at the opening of successive vibrational channels ($2\nu_3$ and ν_2 modes). In the 1–4 meV range the measurements of Schramm et al. coincide with the dissociative attachment CS as derived (Schramm et al. 1999) from Rydberg-atoms experiment (Hildebrandt et al. 1978). At 0.5 meV, the electron attachment CS amounts to $4 \cdot 10^{-17}$ m², by about a factor of 10 less than the theoretical limit for the s-wave capture. Very weak peaks (of the order of 10^{-24} m²) in CH₃I dissociative electron attachment CS were observed also at about 3.5 eV and 8.5 eV, with H⁻, CH⁻ and I⁻ ions present (Krishnakumar and Nagesha 1993).

The absolute values of the electron attachment CS in methyl halides increase on going from CH₃Cl to CH₃Br, and to CH₃I. Differently from CCl₄ or SF₆, the attachment CS in methyl halides does not contribute in a significant way to the TCS. Only in CH₃I this contribution is significant, approaching the theoretical limit for the s-wave capture (Schramm et al. 1999). However, the pulse-radiolysis microwave-cavity method (Shimamori et al. 1992a, Sunagawa and Shimamori 1999) indicates, for CH₃I at 0.04 eV, an electron attachment rate coefficient lower by a factor of four than that in CF₃I (Shimamori et al. 1992b). The molecules with the highest rate coefficients in the 0.01–0.1 eV range are SF₆, CCl₄, CFCl₃, CF₃I and 1,1,1–C₂F₃Cl₃ (Shimamori et al. 1992b) and CH₂Br₂ (Shimamori et al. 1992a).

Dissociative electron attachment CS for other halomethanes (CF₃Br, CF₂Br₂, CF₃I) at 0.2–6 eV were studied in a beam experiment by Underwood-Lemons et al. (1995); for CH₂Br₂, CHBr₃, CH₂I₂, CHI₃ at 0–4 eV by Modelli et al. (1992); for CCl₃Br, CCl₂Br₂, CBr₄ by Olthoff et al. (1986). The pulse-radiolysis microwave-cavity method was applied to CF₃Br (Shimamori et al. 1992b), CBr₄, CHBr₃, CFBr₃, CF₂Br₂, CH₂ClBr, CHClBr₂, CCl₃Br (Sunagawa and Shimamori 1997), CH₃I, CH₂I₂, CF₃I (Sunagawa and Shimamori 1999). The threshold-photoionization method was used for CH₂Br₂, CF₃Br, CF₂Br₂, CCl₃Br, CH₃I (Alajajian et al. 1988) and for flowing plasma afterglow to CF₃Br (Španěl and Smith 1993). In CCl₃Br, the high-resolution beam experiment (Španěl et al. 1997) shows a dissociative attachment CS similar in shape to that in CHCl₃, but a factor of 10 higher. For low-energy peaks in the dissociative attachment CS, Pearl and Burrow (1994) for a series of normal chloroalkanes $C_nH_{2n+1}Cl$ with n=2–9, and Aflatooni et al. (1998) for a series of mono-, di- and trichloroalkanes, showed that the amplitude of the peak σ_A drops exponentially with the attachment energy E_a . Aflatooni et al. (1998) give the following approximating formula for σ_A :

(2)
$$\sigma_{\rm A} = 1860 \cdot 10^{-2.55E_{\rm a}} \cdot 10^{-20} {\rm m}^2$$

where $E_{\rm a}$ is given in eV.

Resonances. Spence (1977) discovered a series of Feshbach resonances in CH₃Cl (the first one at 7.5 eV), CH₃Br (6.7 eV), CH₃I (5.9 eV). He indicated a linear correlation between the energies of the resonances (and the corresponding Rydberg states) with the ionization potential of the molecules. The correlation between resonant states visible in

low-energy electron transmission spectroscopy and in the inner-shell excitation spectra was studied for the four methyl halides and four chloromethanes by Benitez *et al.* (1988). Electron transmission spectra for fluoro-, bromo- and iodomethanes CH_nX_{4-n} with n = 0-4 (except CH_2F_2) up to 7 eV were reported by Modelli *et al.* (1992), for $CBrCl_3$, CBr_2Cl_2 , CBr_4 by Olthoff *et al.* (1986).

For CH₃Cl a shape resonance of 2A_1 symmetry at 3.5 eV due to the temporary occupation of the 8 a_1 (C-Cl) molecular orbital was observed in electron transmission (Burrow et al. 1982) and in vibrational excitation spectra (Shi et al. 1992). This resonance is barely visible in the TCS (Krzysztofowicz and Szmytkowski 1995). Another resonant state observed in the vibrational excitation function at 5.5 eV (Shi et al. 1992) of 2E symmetry is principally associated with C-H antibonding orbitals. Theoretical calculations for CH₃Cl were performed of Fabrikant (1991) and Gallup (1993).

A weak shape resonance near 2.4 eV in CH₃Br, similar to that at 3.5 eV in CH₃Cl, was observed by electron transmission spectroscopy (Modelli *et al.* 1992).

Electron transmission spectra for the whole chloromethanes series $CH_3Cl-CCl_4$ (and $CHFCl_2$, CF_2Cl_2) at 0.2-5 eV were reported by Burrow et al. (1982). Both in CH_2Cl_2 and $CHCl_3$ they observed two temporary negative ion states associated with C-Cl σ^* -orbitals: at 1.2 and 3.4 eV for CH_2Cl_2 (2A_1 and 2B_2 symmetry, respectively) and at 0.35 eV and 1.8 eV in $CHCl_3$ (2A_1 and 2E symmetry). The existence of the two resonant states in CH_2Cl_2 was confirmed in studies of vibrational excitation (Abouaf and Benoit 1990), but to both of them the A_1 symmetry was attributed.

Coupling between dissociative attachment and vibrational excitation CS channels has been observed at room temperature in CH_3I (Schramm *et al.* 1992) and at 700 K in CH_3Cl (Pearl and Burrow 1993).

Other methyl compounds. For methanol CH₃OH total (Szmytkowski and Krzysztofowicz 1995, Sueoka et al. 1985, 1997) and forward-angle electronic excitation (Burton et al. 1992) cross-sections were studied. Total cross-sections for CH₃SH and CH₃NH₂ (Szmytkowski et al. 1995, Szmytkowski and Krzysztofowicz 1995, respectively) were reported at 1–250 eV. Electron attachment to CH₃CN via scattering of Rydberg atoms was studied at ultra-low energies by Desfrançois et al. (1994) and Popple et al. (1995). Dissociation of CH₃CN was observed in fluorescence measurements by Darrach and McConkey (1991).

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3.4. Silicon and germanium halides (SiX₄, GeX₄). – Gaseous silicon halides are widely used in semiconductor industries for plasma growth and etching (Sendra et al. 1995). In spite of this, their interactions with charged particles are known in a fragmentary way. In this subsection we discuss SiF₄, SiCl₄, GeF₄, GeCl₄. The total cross-sections for these gases are compared in fig. 15 in a wide energy range. In this subsection we mention also results on some other targets containing silicon atoms, as SiBr₄ and SiI₄.

Total, elastic and vibrational excitation. TCS in silicon and germanium halides were studied by the Gdańsk group (SiF₄, Karwasz et al. 1998; SiCl₄, Możejko et al. 1999; GeF₄, Szmytkowski et al. 1998; GeCl₄, Szmytkowski et al. 1997) in the 0.5–250 eV energy range, by Wan et al. (1989) for the chlorosilane series (SiH₄, SiH₂Cl₂, SiHCl₃) and for tetrahalosilanes (SiF₄, SiCl₄, SiBr₄, SiI₄) in the 0.2–12 eV range, and by the Trento group (SiF₄, Karwasz et al. 1998; SiCl₄, Możejko et al. 1999) in the 75–4000 eV range. Elastic CS in the 1–40 eV range were calculated by Natalense et al. (1995) and Varella et al. (1999) but the agreement with experimental TCS is poor.

TCS for SiH₄, SiF₄, SiCl₄, SiBr₄ and SiI₄ of Wan *et al.* (1989) are shown in fig. 16. Below 0.8 eV the absolute values of the TCS for the SiH₄-SiI₄ series increase with the total atomic number of the molecule. An indication of a similar relation at intermediate energies can be drawn from experimental data at 12 eV (Wan *et al.* 1989), see fig. 16.

TCS data in SiF₄ from the Gdansk and Trento laboratory (Karwasz *et al.* 1998) merge at 80–250 eV within the combined statistical uncertainties, see fig. 15. The low-energy data of Karwasz *et al.* (1998) agree well with the measurements of Wan *et al.* (1989) as far

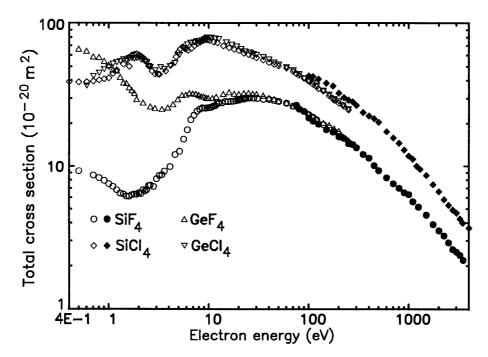


Fig. 15. – Total cross-sections for scattering on silicon and germanium halides. SiF₄ (Karwasz et al. 1998): open circles, data from Gdańsk laboratory; full circles, data from Trento laboratory. SiCl₄ (Możejko et al. 1999): open rhombuses, data from Gdańsk; full rhombuses, data from Trento. GeF₄: Szmytkowski et al. (1998). GeCl₄: Szmytkowski et al. (1997).

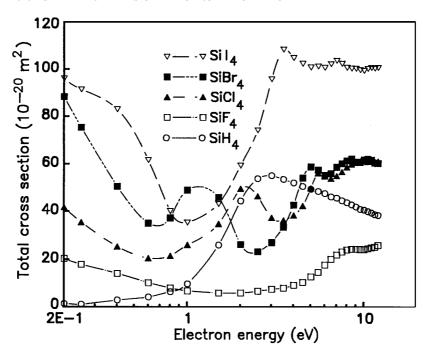


Fig. 16. – Low-energy total cross-sections for scattering on silicon tetrahalides, from Wan et al. (1989).

as the shape is concerned, but are slightly (6% on the average) higher. This discrepancy remains well inside the combined experimental error bars: in fact, the uncertainty of the pressure determination in the experiment of Wan $et\ al.$ (1989) alone amounts to 11%. The relative difference is higher at lower energies, amounting to 15% at 1 eV. It is not to be excluded that below 1 eV the TCS measured by the trochoidal spectrometer (Wan $et\ al.$ 1989) are underestimated because of a poor discrimination for the electrons forward-scattered in vibrationally inelastic collisions. TCS in SiF₄ have also been measured by Bonham and collaborators (Ma $et\ al.$ 1989) but we are not aware of tabulated data.

The TCS in SiF₄ exhibits a broad minimum at 1.5 eV and rises quickly (by a factor of three) between 2 and 7 eV. A shoulder structure is visible at 8.5 eV. A maximum value of $30 \cdot 10^{-20}$ m² is reached at 24 eV. Above 100 eV the TCS falls monotonically to the value of $2.2 \cdot 10^{-20}$ m² at 3500 eV (Karwasz *et al.* 1998). Theoretical calculations (Tossell and Davenport 1984) for the elastic CS in SiF₄ indicate the presence of a very shallow Ramsauer minimum at about 1.5 eV and a rise at low energies. However, the absolute theoretical values exceed the experimental data by a factor of four in the region of the TCS minimum.

The vibrational CS in SiF₄ has been evaluated in the Born approximation by Karwasz et al. (1998). They noticed that the low-energy rise of TCS (Wan et al. 1989, Karwasz et al. 1998) can be explained by direct (non-resonant) vibrational excitation. The difference between the TCS and the vibrational excitation CS estimated with the above procedure (i.e. the elastic CS) reaches a minimum of about $1 \cdot 10^{-20}$ m² at 0.6–0.8 eV.

The TCS in GeF₄, see fig. 15, shows an unusual, for non-polar molecules, low-energy rise, up to $66 \cdot 10^{-20}$ m² at 0.5 eV (Szmytkowski *et al.* 1998). A minimum of about

 $25 \cdot 10^{-20}$ m² at 2.5 eV and three weak maxima of similar amplitudes, $32-33 \cdot 10^{-20}$ m² at 6.4, 16 and 28 eV were observed. At 250 eV the TCS in GeF₄ amounts to $15.2 \cdot 10^{-20}$ m², almost a double value than that for GeH₄ (Karwasz 1995, Szmytkowski *et al.* 1998).

In SiCl₄ the measurements from the Gdańsk and Trento laboratory (Możejko et al. 1999) differ by about 15%, the difference being constant in the whole 100–250 eV energy range of overlap. We noted a similar discrepancy in GeH₄ — it is probably related to a higher angular-resolutions error in the Gdańsk apparatus (Możejko et al 1999). Low-energy data of Możejko et al. agree reasonably well in shape with the determination of Wan et al. (1989). Similar discrepancies in absolute values as in the case of SiF₄ are observed for SiCl₄.

The TCS for SiCl₄ (Wan *et al.* 1989, Możejko *et al.* 1999) shows three rather sharp maxima: of $61 \cdot 10^{-20}$ m² at 1.9 eV, $69 \cdot 10^{-20}$ m² at 5.5 eV, $77 \cdot 10^{-20}$ m² at 9.2 eV (Możejko *et al.* 1999). At 3500 eV the TCS amounts to $3.95 \cdot 10^{-20}$ m². Agreement with the theory (Tossell and Davenport 1984) is rather poor: the theory predicts a resonant state of T_2 symmetry around 1 eV, while the experimental position is 1.9 eV (Możejko *et al.* 1999).

The TCS in GeCl₄ (Szmytkowski *et al.* 1997) shows two rather broad maxima: of $58 \cdot 10^{-20}$ m² at 2.0 eV and of $80 \cdot 10^{-20}$ m² at about 10 eV. A shoulder structure is visible around 6 eV.

Ionization and electronic excitation. In SiF₄ only a few partial cross-section measurements exist: ionization up to 125 eV (Poll et al. 1987), dissociation into neutrals (Nakano and Sugai, 1993) and dissociative attachment (Iga et al. 1992). The available data indicate that at 60 eV the most efficient inelastic process is the dissociative ionization, leading to the formation of SiF₃⁺ ions; the CS for this process (Poll and Meichsner 1987) amounts to 25% of the TCS. The formation of parent SiF₄⁺ ion is one order of magnitude less probable at this energy. Note that in SiH₄ no parent ions were detected in electronimpact ionization (Chatham et al. 1984). We also note a relatively low probability of forming ions and radicals in excited states for SiF₄. The overall CS for these processes, as estimated from UV emission, is merely $0.13 \cdot 10^{-20}$ m² at 100 eV, four times less than for the CF₄ molecule (Aarts 1986).

The cross-section for dissociation into neutrals in SiF_4 reaches a maximum at 70 eV (Nakano and Sugai 1993), where it amounts to about 3% of the TCS (Karwasz *et al.* 1998). 75% of the neutral radicals yield is due to SiF_3 , which exceeds the SiF_2 yield by more than three folds. The SiF and Si signals at 100 eV are about 1/40 and 1/80 of the one for SiF_3 , respectively.

Detailed studies for valence and inner-shell electronic excitation in SiF₄ in the limit of forward scattering were performed by Guo *et al.* (1992). The photoabsorption CS measured with a high-resolution (0.048 eV FWHM) apparatus reaches a maximum of $0.97 \cdot 10^{-20}$ m² at 13.0 eV; a smaller maximum follows at 23 eV (Guo *et al.*).

Optical emission spectra in SiF₄ due to 200 eV electron impact were studied by Hesser and Dressler (1967). Ultraviolet emission CS were given for ionic fragments by Aarts (1986). Energy-loss spectra at 10–17 eV energy loss in trapped electrons experiment were studied by Curtis and Walker (1989). The optical emission spectrum around 564 (400) nm from SiF₄⁺ (GeF₄⁺) ions formed in electron impact ionization of SiF₄(GeF₄) was studied by Mason and Tuckett (1987a, b, respectively). Electron momentum distributions of SiF₄ valence-orbitals were measured by Fantoni *et al.* (1986).

In SiCl₄ semiempirical values of ionization CS at 30–200 eV have been reported by Deutsch *et al.* (1997). The ionization CS according to this model reaches its maximum at 60 eV where it constitutes 30% of the TCS, while at 200 eV this share rises to 35% of

the TCS (Trento data from Możejko et al. 1999).

Inner-shell energy loss spectra in chlorosilanes (SiHCl₃, SiH₂Cl₂, SiCl₄) were reported by Winkler et~al.~(1994). Optical emission in the 350–750 nm range from ionized fragments formed in electron collisions with SiCl₄ and GeCl₄ was reported by Lambert et~al.~(1988). Vacuum UV fluorescence in GeF₄ excited by 10–25 eV synchrotron radiation was studied by Boyle et~al.~(1998).

Dissociative attachment CS in SiF₄ (Wang et al. 1973, Iga et al. 1992, Sauers et al. 1993), for SiF₃ (64%), F⁻ (34%) and some F₂ production, reaches a maximum of $1.5 \cdot 10^{-22}$ m² at 11.4 eV (Iga et al. 1992), i.e. at a somewhat lower energy than the second peak observed in TCS (Karwasz et al. 1998). As noticed by Mann and Linder (1992) for carbon halides CF₃Cl and CF₂Cl₂, the peaks in the electron attachment CS usually precede the maxima in the vibrational excitation. The maximum attachment CS in SiF₄ is slightly lower than the $1.8 \cdot 10^{-20}$ m² peak in CF₄ (Iga et al. 1992).

Dissociative attachment in chlorosilanes (SiH₃Cl, SiH₂Cl₂, SiHCl₃, SiCl₄) was measured by Wan *et al.* (1991). Within the sensibility of their apparatus $(2 \cdot 10^{-22} \text{ m}^2)$, only in the case of SiHCl₃ they measured a non-zero dissociative attachment CS: a value of $0.06 \cdot 10^{-20}$ m² at "zero energy" and a similar value at the peak position (1.2 eV).

In SiCl₄ Dorman (1966) reported a "zero-energy" peak and a maximum at about 7 eV with the Cl⁻ intensity about ten times higher than that of Cl₂. Pabst *et al.* (1977b) fixed the low-energy peak for Cl⁻ formation from SiCl₄ at 1.8 eV and the threshold for this process at 0.5 eV (see also the discussion for GeCl₄ below). Another peak for the Cl⁻ ion formation is centered at 7.5 eV, while peaks for SiCl₃⁻, SiCl₂⁻ and Cl₂⁻ were found at about 7.1, 8.5 and 8.8 eV (Pabst *et al.* 1977b). The determination of Cl⁻ peak positions is in good agreement with the measurements of Wang *et al.* (1974). Moylan *et al.* (1990) observed some additional signal from the SiCl₄⁻ ion at 2 eV impact energy.

Pabst et al. (1977b) reported dissociative attachment in SiCl₄ and SiBr₄. A similar pattern of ions (SiX $_3^-$, SiX $_2^-$, X $_2^-$, X $_2^-$) was observed in both SiBr₄ and SiCl₄. The relative intensities of corresponding ions for SiCl₄ are shifted by about 1.5 eV compared to SiBr₄. No negative ions were reported at near-to-zero energies.

Negative-ion yields and ion energies in $GeCl_4$ were reported by Pabst *et al.* (1977b). Extensive studies, by X-ray absorption, inner-shell electron energy loss spectroscopy, dissociative attachment spectroscopy and *ab inito* calculations, were reported by Guillot *et al.* (1996). Dissociative attachment is dominated by the formation of the $GeCl_3^-$ anion at zero impact energy and the formation of Cl^- and $GeCl_2^-$ anions around 5.6 eV. The existence of a peak for $GeCl_3^-$ at 2 eV (Pabst *et al.* 1977b) was not confirmed by Guillot *et al.* (1996).

Dissociative attachment in tin tetrahalides (SnCl₄, SnBr₄, SnI₄) was studied by Pabst et al. (1977a).

Resonances. Electron transmission spectra at 0.2–12 eV collision energy for silicon tetrahalides and hydrochlorides were obtained by Wan et al. (1989). All examined molecules (SiH₄, SiH₂Cl₂, SiHCl₃, SiCl₄, SiBr₄ and SiI₄) but SiF₄ show a resonant state at energies below 3 eV. The energy of these resonances, visible both in the TCS and in the dissociative attachment channels, moves from 2.9 eV in SiH₄ to 2.2 eV in SiCl₄, 1.2 eV in SiBr₄ and 0.4 eV in SiI₄.

Wan et al. (1989) noticed that the sum of the resonance energy and the "term value", i.e. the ionization energy of a corresponding (i.e. occurring to the same molecular orbital) inner-shell excited state, remains constant (= 6.1 eV) for all the tetrahedral (SiF₄, SiCl₄, SiBr₄, SiI₄) molecules considered. Based on this observation and the inner-shell excitation data of Fredrich et al. (1980), they predicted a resonant energy of 2.9 eV for the

 t_2 state of SiF₄. According to the continuum multiple-scattering parametric-exchange molecular-orbitals calculations of Tossell and Davenport (1984) only a broad shoulder in the t_2 resonant channel is visible at 3 eV. However, neither resonant maximum nor a shoulder structure is visible in the experimental TCS (Wan *et al.* 1989, Karwasz *et al.* 1998).

In SiCl₄, in analogy to CCl₄, the 2.2 eV peak and the broad feature around 5 eV have been assigned to electron capture into the $\sigma^*(t_2)$ and a higher-lying t_2 orbital, respectively (Wan *et al.* 1989).

In GeCl₄, Guillot et~al.~(1996) in electron transmission spectra observed two resonant state, the first one at 1.72 eV of t_2 symmetry. A rather narrow (1.0 eV FWHM) feature was observed at 5.6 eV and tentatively assigned to the electron capture to the 6e orbital. Guillot et~al.~(1996) compared resonant states, as seen in attachment CS in CCl₄, SiCl₄ (Pabst et~al.~1977b), GeCl₄, SnCl₄ (Pabst et~al.~1977a), and discussed the "term-values" rule of Wan et~al.~(1989).

Comparison between different molecules. Clear differences in TCS can be observed between hydrides and halides of silicon and germanium, compare fig. 2 for SiH₄ with fig. 15 for SiF₄. TCS in SiH₄ exceeds the SiF₄ one in the whole 1–25 eV energy range, being lower by a factor of two above 100 eV. One notes that the TCS shape in this energy range for tetrahedral molecules CX₄, SiX₄, GeX₄ (X = H, F, Cl) depends rather on external atoms than on the central one. The hydrides exhibit a single maximum of TCS with a value that rises from $24 \cdot 10^{-20}$ m² at 8 eV in CH₄ (Zecca et al. 1991) to about $55 \cdot 10^{-20}$ m² at 3–4 eV in SiH₄ (Szmytkowski et al. 1997) and GeH₄ (Możejko et al. 1996). The fluorides (CF₄, SiF₄, GeF₄) exhibit a broad maximum at 20–30 eV and another, narrower maximum (a shoulder in the case of SiF₄) at energies slightly below 10 eV. As noticed by Karwasz et al. (1998) a maximum of a similar width occurs in the total cross-sections in SF₆. The position of this low-energy maximum moves from 8.5 eV in CF₄ to 7 eV in SF₆ and GeF₄, while the amplitude increases from $21 \cdot 10^{-20}$ m² in CF₄ (Szmytkowski et al. 1992) to $33 \cdot 10^{-20}$ m² in SF₆ (Kasperski et al. 1997).

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 $3^{\circ}5$. Trihalides (NF₃ and others). – In this subsection we discuss a few other halides, which do not belong to a particular chemical group. More data, in particular elastic CS measurements, exist for nitrogen trifluoride (NF₃) than for other compounds discussed in this chapter (BF₃, BCl₃, PF₃, PF₅). We are not aware of TCS for any of the gases presented in this chapter.

Nitrogen trifluoride (NF₃)

Elastic and vibrational cross-sections for NF₃ were measured in an absolute experiment by Boesten et al. (1996) between 1.5 and 100 eV. In spite of the fact that the molecular geometry, electronic configuration and vibrational modes are in NF₃ the same as in NH₃, the elastic and vibrational CS differ substantially for these two targets. The DCS for elastic scattering in NF₃ are rather uniform in angle between 1.5 and 3 eV, differently than in NH₃, compare fig. 3a in part II. At 8–100 eV the DCS in NF₃ shows only one minimum, with some "interference" structures, like those seen for OCS and CS₂ molecules, compare fig. 18 in part II. The NF₃ integral CS reaches a broad peak with a maximum of $18.4 \cdot 10^{-20}$ m² at 4 eV (Boesten et al. 1996).

The vibrational excitation functions for the ν_1 and ν_3 stretching modes show a 2 eV wide resonance centered at 3 eV. The excitation functions in NF₃ do not resemble those in NH₃ (Gulley *et al.* 1992) but the ones in PH₃ (Tronc *et al.* 1981). The DCS for the unresolved $\nu_{1,3}$ mode is in NF₃ forward centered at all energies between 0.5 and 9 eV (Boesten *et al.* 1996). At 3 eV the DCS falls by a factor of two between 20° and 60°; then it remains constant up to 130°.

Differential, momentum transfer and integral CS for elastic scattering and excitation to the ${}^{3}E$ and ${}^{1}E$ electronic states in NF₃ were calculated by Rescigno (1995).

Swarms. Early swarm measurements in NF₃ were compiled by Gallagher et al. (1983). More recently analysis of swarm parameters (energy distribution, attachment coefficient, drift velocity) in NF₃ and its mixtures with noble gases and N₂ were performed by Ushidora et al. (1990).

The optical emission CS from different dissociated fragments of NF₃ was studied by Becker and collaborators (Blanks et al. 1987, Blanks and Becker 1987, Roque et al. 1991). In particular, emission from fluorine excited atoms was studied: transitions between the lowest excited states and the ground state $(2p^43s)^{2,4}P \rightarrow (2p^5)^2P^0$ at 95.5 and 97.5 nm wavelength (Roque et al. 1991) and emission in the 620–790 nm range due to transitions between some higher and the lowest excited states $(2p^43p) \rightarrow (2p^43s)^{2,4}P$ (Blanks et al. 1987). In all spectral ranges studied the F emission from NF₃ is higher than the one from the CF₄ and SF₆ series: for the $(2p^43s)^2P \rightarrow (2p^5)^2P^0$ transition by a factor of almost 6 and 2, respectively (Roque et al. 1991).

Ionization in NF₃ was studied by Reese and Dibeler (1956). Appearance potentials for NF $_3^+$, NF $_2^+$

Electron attachment in NF₃ was studied in swarm experiment by Lakdawala and Moruzzi (1980), in a beam experiment by Harland and Franklin (1974) and in a flowing afterglow experiment by Miller et al. (1995). The electron attachment CS for the F⁻ + NF₂ decay channel exhibits a resonant peak of $0.6 \cdot 10^{-20}$ m² at 1.7 eV collision energy (Harland and Franklin). The F₂ and NF₂ are produced with intensities much lower than F⁻, i.e. $3 \cdot 10^{-23}$ m² and $3 \cdot 10^{-24}$ m², respectively. Another, smaller maximum in the F⁻ cross-section (the F⁻ + N + 2F decay channel) was reported by Thynne (1969). The electron attachment coefficient was measured by the flowing afterglow method in

the 300–550 K temperature range by Miller *et al.* (1995). At room temperature the attachment coefficient is of the order of 10^{-11} cm³/s. Negative ions in radio-frequency discharge were studied by Hayashi *et al.* (1997).

Phosphorus fluorides (PF₃, PF₅)

Electron excitation for phosphorus halides (PF₃, PF₅, PCl₃) in forward-scattering configuration was studied by Brion and co-workers (Au et al. 1997a, c, b, respectively). For the two fluorides, below the ionization threshold (12.3 and 15.5 eV) the photoabsorption CS reaches similar values at its maximum: $0.44 \cdot 10^{-20}$ m² at 8.0 eV and $0.42 \cdot 10^{-20}$ m² at 14.0 eV, for PF₃ and PF₅, respectively. The photoabsorption CS in PCl₃ rises monotonically from 4.5 eV up to 13.5 eV where it reaches $1.72 \cdot 10^{-20}$ m². 53% of this cross-section is due to photoionization (Au et al. 1997b).

At 100 eV the photoionization CS in the two fluorides, PF₃ and PF₅ is higher than in PCl₃ ($9 \cdot 10^{-22}$ m², $14 \cdot 10^{-22}$ m² and $2.8 \cdot 10^{-22}$ m², respectively) (Au *et al.* 1997a, c, b). At this energy, PF₂⁺ (55%) and PF₄⁺ (52%) ions dominate in the ionization pattern of PF₃ and PF₅, respectively.

The electron attachment rate coefficient in PF_5 at 300–550 K is three orders of magnitude lower than in SF_6 (Miller *et al.* 1995). No electron attachment to PF_3 was observed in this experiment.

Boron fluoride (BF₃)

The electron attachment CS in BF₃ shows a large peak between 10 and 12 eV (Harland and Franklin 1974). The maximum $(2 \cdot 10^{-24} \text{ m}^2)$ of the F₂ formation CS is reached at 11.2 eV. The F⁻ cross-section is an overlap of two unresolved peaks, the higher one centered at 10.4 eV $(1 \cdot 10^{-22} \text{ m}^2)$ and the weaker one at 12.1 eV. The ionization CS reaches a flat maximum of $5.5 \cdot 10^{-20}$ m² at 120–170 eV (Kurepa *et al.* 1976).

Boron chloride (BCl₃)

The electron attachment CS was studied in a swarm experiment by Petrović et al. (1990). The photoemission CS for the BCl $(A^{-1}\Pi \to X^{-1}\Sigma^{+})$ system was studied by Jabbour et al. (1988). Vacuum-UV fluorescence in the 9–22 eV energy range in BF₃ and BCl₃ was studied by Boyle et al. (1999). Radical and ion compositions in BCl₃ plasmas were studied by Kazami et al. (1997).

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3.6. Linear fluorocarbons (C_2F_6 , C_3F_8). – Numerous other halocarbons, in addition to methane-like (subsects. 3.1–3.3) and benzene-like (subsect. 3.7) were studied by electron scattering, mainly in swarm and electron-attachment experiments. Due to high breakdown voltages (Brand and Kopainsky 1979), fluorinated alkanes and alkenes are used in electrical switches. In this subsection we present total and partial CS for two linear perfluoroalkanes: hexafluoroethane (C_2F_6), see fig. 17, and perfluoropropane (C_3F_8), see fig. 18. Detailed analyses of electron scattering in these gases were performed by Christophorou and Olthoff (1988a and b, respectively). In this subsection we also quote some other works on linear fluorocarbons. These gases are of basic importance both for the semiconductor technology (see, for example, Doh et al. 1996) and for the environmental chemistry.

Total cross-sections. The TCS in C₂F₆, C₃F₈, c-C₄F₈ (and CHF₃) have been recently measured by Sanabia et al. (1998) in a trochoidal spectrometer up to 20 eV. The TCS for C₂F₆, C₃F₈, C₄F₈ (and CF₄) in the 0.8–600 eV range were presented by Sueoka et al. (1997, 1999). Indicative values of the backward-scattering CS in C₂F₆, C₃F₈ (and CF₄) at energies up to 1 eV, 6 eV (and 1 eV), respectively, were obtained using a synchrotron-radiation electron source by Lunt et al. (1998); absolute values of the backward-scattering CS were given in the 10–175 meV energy range (Lunt et al. 1998).

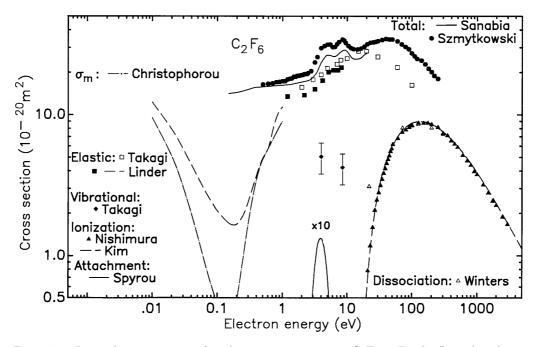


Fig. 17. – Integral cross-sections for electron scattering on C₂F₆. Total: Szmytkowski et al. (2000); Sanabia et al. (1998). Elastic: Takagi et al. (1994); Linder, experimental and modified effective-range extrapolation, data from Christophorou and Olthoff (1998a). Momentum transfer: Christophorou and Olthoff (1998a) recommended values. Vibrational excitation: Takagi et al. (1994). Electron attachment: Spyrou et al. (1983), normalized in the maximum to the recommended value by Christophorou and Olthoff (1998b). Ionization: Nishimura et al. (1999), experimental; Kim stands for the semi empirical model from Nishimura et al. (1999). Dissociation (ions + neutrals): Winters and Inokuti (1982).

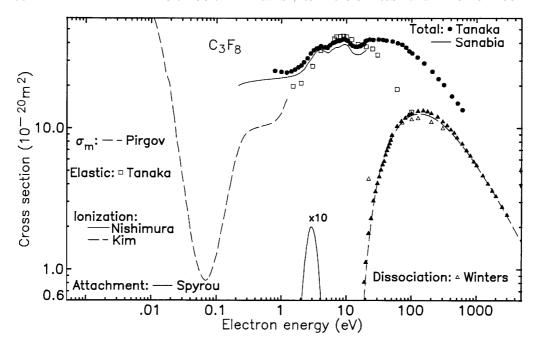


Fig. 18. – Integral cross-sections for electron scattering on C_3F_8 . Total: Tanaka et al. (1999), read from figure; Sanabia et al. (1998). Momentum transfer: Pirgov and Stefanov (1990). Elastic: Tanaka et al. (1999). Electron attachment: Spyrou et al. (1983), normalized in the maximum to the recommended value by Christophorou and Olthoff (1998b). Ionization: Nishimura et al. (1999), experimental; Kim stands for the semi-empirical model from Nishimura et al. (1999). Dissociation (ions + neutrals): Winters and Inokuti (1982).

Absolute TCS in C_2F_6 have been measured by Sanabia et~al.~(1998) up to 20 eV and by Szmytkowski et~al.~(2000) in a transmission experiment from 0.5 to 250 eV. Both experiments indicate a decrease of the TCS in low-energy limits reached by their apparatuses (see fig. 17). The backward scattering CS gives a minimum of roughly $3 \cdot 10^{-20}$ m² at 0.06–0.08 eV (Lunt et~al.~1998). The TCS exhibits resonant enhancements at 4.7 and 8.5 eV (Sanabia et~al.~1998). The TCS values by Szmytkowski et~al.~(2000) are higher than the other two data sets (Sanabia et~al.~, Sueoka et~al.~1997) and quote CS maxima of $31.3 \cdot 10^{-20}$ m² at 5.5 eV, $34.0 \cdot 10^{-20}$ m² at 8.5 eV and a broad maximum of $43.3 \cdot 10^{-20}$ m² centered around 40 eV.

The TCS in C_3F_8 has been measured by Tanaka *et al.* (1999) in the 0.8–600 eV range. The preliminary measurements of Sueoka *et al.* (1997) are not shown in fig. 17. The TCS energy dependence is to some extent similar to the one in CF_4 : the C_3F_8 total cross-section shows a similar maximum at 9 eV, the same large bump at 20–50 eV but, additionally, a weak maximum at 4 eV. The value of TCS in 9 eV maximum, about $42.5 \cdot 10^{-20}$ m², is lower in C_3F_8 than in C_3H_8 (Tanaka *et al.* 1999). The low-energy measurements (Sanabia *et al.* 1998) indicate a resonant maximum at 3.6 and a weak structure at 6.1 eV. Both the TCS in C_2F_6 and C_3F_8 fall slowly with lowering energy toward 0.1–0.2 eV (Sanabia *et al.*). Lunt *et al.* (1998) reported a minimum in the backward scattering CS of roughly $6 \cdot 10^{-20}$ m² at 0.05–0.1 eV.

Swarm experiments. The early electron swarm data in C₂F₆, C₃F₈, C₄F₁₀ were

reviewed by Gallagher *et al.* (1983). Recommended momentum transfer CS for C_2F_6 and C_3F_8 were given by Christophorou and Olthoff (1998a and b, respectively). The analysis of swarm data for C_2F_6 and C_3F_8 was also done by Pirgov and Stefanov (1990).

The drift velocity in C_2F_6 , C_3F_8 , $n\text{-}C_4F_{10}$ (and CF_4) were measured at reduced electric fields between $0.03 \cdot 10^{-21} \text{ Vm}^2$ and $500 \cdot 10^{-21} \text{ Vm}^2$ by Hunter *et al.* (1988). In perfluoroalkanes (CF_4 , C_2F_6 , C_3F_8), for reduced field values in the 10^{-20} Vm^2 range, the drift velocity is a descending function of this field, indicating the existence of a Ramsauer-Townsend minimum. The effect in C_2F_6 and C_3F_8 is less pronounced than in CF_4 ; no effect was observed in $n\text{-}C_4F_{10}$. This is consistent with the above-quoted observation in TCS measurements.

Gee and Freeman (1992) measured the electron mobility in C_2F_6 at reduced electric fields values below $2 \cdot 10^{-21}$ Vm². The momentum transfer CS falls from $20 \cdot 10^{-20}$ m² at 6 meV to about $3.2 \cdot 10^{-20}$ m² at 90 meV; then it rises sharply, due to the onset of vibrational excitation channels (Gee and Freeman).

The elastic cross-section in C_2F_6 at 2–100 eV was measured by Takagi et al. (1994). The DCS at 2 eV shows a sharp minimum at 20°, which becomes shallower with rising the energy. At 8–20 eV the DCS shows a d-wave pattern with two minima at about 40° and 120° and a maximum at 90°. The integral elastic CS shows a single, broad maximum centered at 15–20 eV.

The low-energy measurements in C_2F_6 and the modified effective-range analysis of Merz and Linder (1998) were reported by Christophorou and Olthoff (1998a). According to the analysis of Merz and Linder, the integral elastic CS shows a minimum of $1.6 \cdot 10^{-20}$ m² at about 0.2 eV.

The C_3F_8 integral elastic CS (Tanaka *et al.* 1999, note an error in the caption of their table II) shows a maximum in the integral CS coinciding both in position (9 eV) and in value $(45 \cdot 10^{-20} \text{ m}^2)$ with the one in C_3H_8 . However, outside this maximum, the integral CS in C_3F_8 is higher than that in C_3H_8 both at low and high energies.

Elastic CS for fluorinated derivatives of hydrocarbon molecules were announced by Merz and Linder (1998) in their article on C_2H_6 .

Vibrational cross-sections in C_2F_6 between 1 eV and 16 eV have been measured by Takagi et al. (1994). The excitation functions for the "stretching modes" (0.16 eV energy loss) at 60°, 90° and 120° show an enhancement at 4.5 and at 8.5 eV. At 4.5 eV the ν_1 , ν_3 , ν_2 stretching modes (all of A_{1g} symmetry) and their higher harmonics are excited; at 8.5 eV both the stretching and bending modes (E_g symmetry) ν_{10} , ν_{11} , ν_{12} are excited. The DCS for stretching modes are forward centered at all studied energies but particularly at the low ones. At 4 eV the vibrational excitation CS amounts to about 26% of the elastic CS and at 8.5 eV to about 17% (Takagi et al. 1994).

Vibrational cross-sections in C_3F_8 in the 3–15 eV energy range were reported by Tanaka et al. (1999). The energy resolution of this experiment did not allow to resolve individual vibrational modes (including 25 fundamental modes and their overtones). An enhancement of the ν_1 (C-C stretching), ν_2 (C-F₃ stretching) and ν_3 (C-F₂ stretching) mode and their overtones was observed at 3 eV and between 6 and 9 eV. At 6.5 eV and 9 eV, apart from the above modes, also the ν_4 (C-C stretching) and ν_6 (C-F₂ stretching deformation) modes are enhanced. The ν_1 mode becomes more prominent at 4 and 9 eV. DCS at 3 eV are p-wave-like for the ν_1 mode and rather d-wave-like for the ν_2 and ν_3 modes.

Electronic excitation. We are not aware of electronic excitation integral CS for C_2F_6 and C_3F_8 . Electronic excitation in six fluoroethylenes (C_2H_3F ; 1,1- $C_2H_2F_2$; cis-1,2- $C_2H_2F_2$; cis-1,1- $C_2H_2F_2$; trans-1,2- $C_2H_2F_2$, C_2HF_3 , C_2F_4) and chlorotrifluoroethylene

 (C_2F_3Cl) was studied in detail at impact energies of 20 (or 25), 40, 60 eV by Coggiola et al. (1976). Relative DCS in the 0°-80° angular range were reported. Two transitions were studied: the " $N \to T$ " ($\pi \to \pi^*$) singlet-triplet one at 4.18–4.68 energy loss and the " $N \to V$ ", optically allowed, at 6–10 eV energy loss (depending on the target). The DCS shows typical angular dependencies: a flat one for the " $N \to T$ " excitation and a forward-peaked one for the " $N \to V$ " excitation. The relative values of DCS vary by three orders of magnitude from monofluoroethylene (C_2F_3Cl). Energy loss spectra up to 16 eV energy loss were also reported by Kuppermann et al. (1979).

Ionization cross-sections, both theoretical and experimental, on C_2F_6 , C_3F_8 (and CF_4) have recently been given up to 3000 eV by Nishimura et al. (1999). For both gases the experimental data of Nishimura et al. (1999) agree well up to 100 eV with the data of Poll and Meichsner (1985) and with the data of Beran and Kevan (1969) at 20, 35 and 70 eV. The data of Kurepa (1965) are higher and those of Chantry and Chen (1989) for C_3F_8 are lower. In fig. 17 the semiempirical values reported by Nishimura et al. (1999) are labeled with "Kim".

Neither in C_2F_6 nor C_3F_8 parent ions were measured (Poll and Meichnser 1987). In both gases the CF_3^+ ion dominates. At 100 eV the branching ratios in C_2F_6 are 33%, 50%, 6% and 11% for $C_2F_5^+$, CF_3^+ , CF_2^+ and CF^+ ions, respectively; for C_3F_8 the $C_3F_7^+$, $C_2F_5^+$, $C_2F_4^+$, CF_3^+ , CF_2^+ and CF^+ ions scale as 12:5:3:67:3:10 (numerical data from Christophorou and Olthoff 1998a, b).

Dissociation cross-section. The overall CS for ionization and dissociation into neutrals of C_2F_6 and C_3F_8 were measured by Winters and Inokuti (1982) up to 300 eV with a $\pm 20\%$ uncertainty. At 70–300 eV their total (neutrals + ions) CS in C_3F_8 is double than that for CF₄, reaching the maximum value of $11.8 \cdot 10^{-20}$ m² at 100-120 eV; in C_2F_6 this maximum amounts to $8.6 \cdot 10^{-20}$ m². These values coincide, within the experimental errors, with the total ionization CS of Nishimura et al. (1999). In the near-to-threshold region (22 eV) the difference between the ionization CS (Nishimura et al. 1999) and Winter and Inokuti's CS amounts to $2.1 \cdot 10^{-20}$ m² and $2.6 \cdot 10^{-20}$ m² in $2.6 \cdot 10^{-20}$ m²

The recent measurements of Motlagh and Moore (1998) in C_2F_6 and in C_3F_8 have solely reported the overall CS for dissociation into both the ionized and neutral radicals. For C_2F_6 the overall CS for formation of CF_3 and C_2F_5 radicals was measured. This CS reaches a maximum of about $4.8 \cdot 10^{-20}$ m² at 70–100 eV. Subtracting from this value the CF_3^+ partial ionization cross-sections of Poll and Meichsner (1987), a neutral dissociation CS of about $1 \cdot 10^{-20}$ m² was deduced (Motlagh and Moore 1998). In C_3F_8 the cross-section for overall dissociation into the neutral and ionized CF_3 and C_2F_5 radicals amounts to $6 \cdot 10^{-20}$ m² at 100 eV (Motlagh and Moore 1998).

Electron attachment in fluorocarbons was studied with different methods. In swarm experiments, among others, the following targets were studied: C_2F_6 (Spyrou and Christophorou 1985a); 2- C_4F_8 (Christophorou et al. 1974); c- C_4F_6 (Datskos et al. 1993). Very low-energy experiments using electrons from Kr-threshold photoionization were performed on: 2- C_4F_6 (Chutjian et al. 1984); c- C_6F_{10} , c- C_6F_{12} , C_8F_{16} , 1,1,2- $C_2F_3Cl_3$ (Alajajian and Chutjian 1986); c- C_4F_6 , c- C_4F_8 , c- C_7F_{14} (Chutjian and Alajajian 1987); 1,1,1- $C_2F_3Cl_3$ (Alajajian et al. 1988). Beam experiments were carried out in: C_2F_6 , C_3F_8 , c- C_4F_8 (Harland and Franklin 1974); c- C_5F_8 , c- C_6F_{10} , c- C_6F_{12} , C_7F_8 , C_8F_{16} (Pai et al. 1979); perfluoroalkanes n- C_nF_{2n+2} (n=1-6) and i- C_4F_{10} (Spyrou et al. 1983); C_3F_8 and C_2H_3Cl (Chantry and Chen 1989); fluoroethylenes C_2F_4 , trans-1,2- $C_2H_2F_2$,

cis-1,2-C₂H₂F₂, 1,1-C₂H₂F₂ (Oster *et al.* 1989); C₂H₅Cl (Pearl and Burrow 1993); normal chloroalkanes $C_nH_{2n+1}Cl$ (n=2–9) by Pearl and Burrow (1994); C₂F₆, C₂F₄ (Weik and Illenberger 1995). Pulse-radiolysis microwave experiments were performed on: 1,1,1-C₂F₃Cl₃, 1,1,1-C₂H₃Cl₃ (Shimamori *et al.* 1992); in Rydberg atoms quenching: 1,1,1-C₂F₃Cl₃, C₅F₈ and c-C₇F₁₄ (Zollars *et al.* 1984, Dunning 1987); in flowing plasma afterglow: C₂H₃Cl₃, C₂F₃Cl₃ (Smith *et al.* 1989).

Dissociative attachment only was observed in C_2F_6 (and CF_4), while in C_3F_8 (and $n\text{-}C_4F_{10}$) also non-dissociative attachment was measured (Hunter and Christophorou 1984, Spyrou and Christophorou 1985a, b, Datskos and Christophorou 1987, Hunter et al. 1987). The thermal attachment rate constant is below 10^{-16} cm³ s⁻¹ in CF_4 and C_2F_6 , while it amounts to about $4 \cdot 10^{-11}$ cm³ s⁻¹ in $n\text{-}C_4F_{10}$ and is not greater than $3 \cdot 10^{-13}$ cm³ s⁻¹ in C_3F_8 (Hunter et al. 1987).

In C_2F_6 three fragment negative ions were observed: F^- , CF_3^- and $C_2F_5^-$, with relative peak intensities less than 1000, 316, and 1 (Spyrou *et al.* 1983). The F^- and CF_3^- peaks are centered around 3.9–4.0 eV, while the $C_2H_5^-$ peak is at 4.8 eV. A recommended value of $0.14 \cdot 10^{-20}$ m², *i.e.* about 0.5% of the TCS (Szmytkowski *et al.* 2000), has been given for the total electron attachment CS at 4.0 eV (Christophorou *et al.* 1998a).

In C_3F_8 five fragment negative ions were observed: $C_3F_7^-$, $C_2F_5^-$, $C_2F_3^-$, CF_3^- and F^- , with relative peak intensities of 1.7, 66, 1.8, 54 and 1000 (Spyrou *et al.* 1983). The F^- peak is centered at about 2.9 eV while the others at 3.2–3.7 eV. The signal from CF_3^- extends up to 7 eV (Spyrou *et al.* 1983). A recommended value of $0.2 \cdot 10^{-20}$ m², *i.e.* about 0.7% of the TCS (Tanaka *et al.* 1999) has been given for the total electron attachment CS at 2.5 eV (Christophorou *et al.* 1998b).

Sum check. The sum of the elastic and vibrational CS (Takagi et al. 1994) in C_2F_6 at the maxima of the resonant vibrational excitation (4.0 and 8.5 eV) is somewhat lower than the recent TCS (Szmytkowski et al. 2000) and agrees better with the TCS of Sanabia et al. (1998), see Table VIII. On the other hand, at 15 eV the elastic CS (Takagi et al. 1994) exceeds the TCS of Sanabia et al. At 22 eV the sum of the elastic CS (Takagi et al. 1994) and the dissociation (neutrals + ions) CS (Winters and Inokuti 1982) agrees within the experimental uncertainty with the TCS of Szmytkowski et al. (2000). At 100 eV, the difference between the TCS (Tanaka et al. 1999) and the summed ionization (Nishimura et al. 1999) and elastic (Tanaka et al. 1999) CS amounts to 10% of the TCS and corresponds possibly to the electronic excitation CS.

No absolute values of vibrational CS were given by Tanaka et~al.~(1999) for C_3F_8 , so no sum check is possible at low energies, see Table IX. At 6–15 eV the elastic CS (Tanaka et~al.~1999) exceeds the two (Tanaka et~al.~1999, Sanabia et~al.~1998) TCS determinations. Similarly to in C_2F_6 , at 22 eV the sum of the elastic CS (Tanaka et~al.~1999) and the dissociation (neutrals + ions) CS (Winters and Inokuti 1982) agrees very well with the TCS (Tanaka et~al.~1999). At higher energies the TCS is 10–15% higher than the sum of the ionization (Nishimura et~al.~1999) and elastic CS; the missing part can correspond to the electronic and/or dissociation-into-neutrals CS.

Other fluorocarbons. The C_2F_4 elastic and vibrational CS have been studied in the 1.5–100 eV energy range by Suzuki et al. (1999) and Okamoto et al. (1999a, b). The C_2F_4 elastic DCS at 2 eV (Suzuki et al. 1999) and 10 eV (Okamoto et al. 1999b) are unusually low compared to C_2H_4 . The DCS for vibrational excitation in C_2F_4 indicate two broad resonances at 3 eV and 6 eV (Suzuki et al. 1999).

The total cross-section in c-C₄F₈ was measured at 2–3000 eV by Nishimura (1999) and by Sanabia et~al. (1998) below 20 eV. The TCS shows a minimum of about $21 \cdot 10^{-20}$ m² at 3.5–4.0 eV and an almost a constant value of 38– $40 \cdot 10^{-20}$ m² at 9–20 eV (Sanabia

Table VIII. – Integral cross-sections for electron scattering on C_2F_6 (in 10^{-20} m² units).

Energy (eV)	Elastic	Vibrational	Dissociation	Ionization	Sum	Total
2.0	15.5 T					19.0 Sz 17.0 SA
4.0	19.1 T	5.0 T1			24.1	29.5 Sz 23.5 SA
8.5	24.3 T	4.2 T1			28.5	34.0 Sz 28.3 SA
15	28.0 T					29.0*Sz 24.8 SA
20	28.1 T					31.1*Sz
22	27.0*T		3.1 W	(0.98*N)	30.1	31.9*Sz
30	25.3 T			2.65*N	28.0	33.1 Sz
60	21.5 T			6.76*N	28.3	33.0 Sz
72	20.2*T		(8.1) W	7.60*N	27.8	31.2*Sz
100	16.1 T		(8.5) W	8.39 N	24.5	27.1 Sz
125			(8.6) W	8.77 N		24.7*Sz
200			(8.1) W	8.57 N		20.0 Sz

^{*} Interpolated values

et al.). The values of Nishimura agree well in shape with those of Sanabia et al. (1998) but are about 10–15% higher; small humps can be seen around 9, 10 and 40 eV; at 3000 eV the TCS amounts to about $4.5 \cdot 10^{-20}$ m².

Elastic CS in c-C₄F₈ at 3–15 eV were reported by Okamoto et~al. (1999b). The DCS for this molecule (Okamoto et~al.) resemble much, both in shape and in absolute values, those for C₂F₆ (Takagi et~al. 1994). Partial ionization CS and neutral dissociation CS in c-C₄F₈ from threshold to 250 eV were measured by Toyoda et~al. (1997), radical production rates by Sasaki et~al. (1998) and Sasaki and Kadota (1999), swarm coefficients by Yamaji et~al. (1999). Itoh et~al. (1991) gave a set of cross-sections up to 100 eV in c-C₄F₈.

Resonances in ethylene C_2H_4 and five fluoroethylenes up to C_4F_4 were studied by Chu et al. (1979). The position of resonances observed in transmission current rises from 1.78 eV in C_2H_4 to 3.0 eV in C_2F_4 . Fluorination effects on the inner-shell spectra of unsaturated molecules, including five fluoroethylenes, were studied by Robin et al. (1988).

^() Not used for summation

N - Nishimura et al. (1999)

T - Takagi *et al.* (1994)

T1- Takagi et al. (1994), normalized to elastic, given at 4.0 and 8.5 eV

W - Winters and Inokuti (1982)

Sz - Szmytkowski et al. (2000)

SA - Sanabia et al. (1998)

Table IX. – Integral cross-sections for electron scattering on C₃F₈ (in 10⁻²⁰ m² units).

Energy (eV)	Elastic	Dissociation	Ionization	Sum	Total
1.5	19.8 TA				25.3 T
2.0	20.8 TA				27.8 T
5.0	37.5 TA				37.0 T
10	44.3 TA				41.6 T
15	39.15 TA				38.0 T
20	37.6 TA		1.13 N	38.7	41.8 T
22	37.1*TA	4.4 W	(1.81) N	41.5	42.4 T
30	32.9 TA		4.10 N	37.0	42.5 T
60	18.8 TA		10.4 N	29.2	40.0 T
72	17.1*TA	(11.0) W	10.6*N	27.7	39.0 T
100	13.0 TA	(11.6) W	12.8 N	25.8	33.8 T
125		(11.8) W	13.3 N		31.6 T
200		(11.1) W	12.8 N		25.1 T
300		(10.0) W	11.2 N		20.0 T

^{*} Interpolated data

- () Not used for summation
- N Nishimura et al. (1999)
- TA Tanaka *et al.* (1999)
- T Tanaka et al. (1999), read from their figure
- W Winters and Inokuti (1982)

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3.7. Hexafluorobenzene (C_6F_6) . – Hexafluorobenzene (C_6F_6) is a molecule strongly attaching electrons at low energies. Most of the existing measurements are concerned with this phenomenon. We are not aware of measurements of partial CS. A set of attachment cross-sections is reported in fig. 19 in the 1 meV–2 eV energy range.

Total cross-sections between 0.6 and 250 eV have been measured in a linear transmission experiment by Kasperski et al. (1997). The TCS exhibits a very weak maximum of $33 \cdot 10^{-20}$ m² at 1.1 eV, a shallow minimum at 3.5 eV and a broad maximum of $60 \cdot 10^{-20}$ m² at 25–30 eV. A shoulder structure is visible on the left side of this maximum, at about 15 eV. The TCS dependence from energy is similar to that in C_6H_6 , but is shifted to higher energies, compare fig. 2.

The TCS below 1 eV in monohalogenated benzenes, C_6H_5F , C_6H_5Cl , C_6H_5Br , C_6H_5I were studied using a synchrotron-radiation electron source (Lunt *et al.* 1999). Both integral and "backward scattering" CS were determined. In the zero-energy limit the integral CS rise, in good accord with the Born approximation for rotational excitation (Lunt *et al.* 1999), and at 0.1 eV amount to $500-800 \cdot 10^{-20}$ m² for the $C_6H_5F-C_6H_5I$ series. Weak resonant structures in TCS (about $5 \cdot 10^{-20}$ m²) at 0.7-1.0 eV were observed in C_6H_5F , C_6H_5Cl and C_6H_5Br (Lunt *et al.* 1999).

Electronic excitation. Energy loss spectra in fluorobenzenes for the 3.4–8.5 eV energy

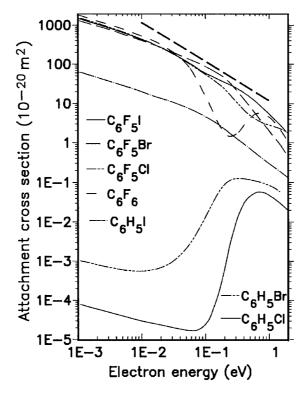


Fig. 19. – Electron attachment cross-sections for monohalomethanes C_6H_5I , C_6H_5Br , C_6H_5Cl (Shimamori et~al.~1995) and pentafluorobenzenes C_6F_5I , C_6F_5Br , C_6F_5Cl and C_6F_6 (Shimamori et~al.~1994) from pulse-radiolysis microwave-cavity measurements. The uppermost straight line indicates the theoretical limit for the s-wave capture (Wigner 1948).

loss were reported by Frueholz et al. (1977, 1979). Transition energies are similar in the whole series C_6H_6 - C_6F_6 . In C_6F_6 these energies are 3.86 eV for the excitation to the first triplet state ($\tilde{X}^1A_{1g} \to 1^3B_{1u}$, the symmetry designations correspond to the transitions in benzene), 4.80, 6.36 and 7.10 eV for the excitations to singlet states (1^1B_{2u} , 1^1B_{1u} , 1^1E_{1u} , respectively). The excitation to another triplet, 1^3E_{1u} state, like that in C_6H_6 at 5.59 eV, has not been observed for C_6F_6 ; instead, the excitation to a singlet state at 5.32 eV was seen (Frueholz et al. 1977). The electron excitation in monosubstituted benzene derivatives (including C_6H_5F) was studied by the electron-scavenger method (Christophorou et al. 1974).

Ionization. Diebler et al. (1957) reported relative ion abundances for C_6H_6 and C_6F_6 at 70 eV collision energy. Fragment ions, in particular $C_5X_3^+$ and CX^+ , are generally more abundant in C_6F_6 than in C_6H_6 . The $C_5F_5^+$ and $C_5F_4^+$ ions have no analogs in C_6H_6 .

Electron attachment. C_6F_6 , similarly to SF_6 , forms a quasi-stable parent anion, $C_6F_6^-$. The lifetime of the transient vibrationally excited $C_6F_6^{-*}$ ion formed by electron attachment at thermal energies is about 12 μs (Naff et al. 1968, Gant and Christophorou 1976, Christophorou et al. 1977). The ion can be stabilized in collisions with a buffer gas (N₂, Datskos et al. 1993) or in clustering reactions (Hiraoka et al. 1990). The electron affinity of C_6F_6 is +0.86 eV (Jordan et al. 1976b, Chen et al. 1985, Wentworth et al. 1987, see also Weik and Illenberger 1995 and Chowdhury et al. 1986). The electron attachment has been studied by beam (Fenzlaff and Illenberger 1984, Fenzlaff and Illenberger 1989, Oster et al. 1989, Weik and Illenberger 1995), swarm (Gant and Christophorou 1976, Spyrou and Christophorou 1985, Datskos et al. 1993), flowing afterglow (Adams et al. 1985), threshold photoionization (Chutjian and Alajajian 1985), Rydberg atoms quenching (Hildebrandt et al. 1978, Marawar et al. 1988) techniques. Woodin et al. (1980) used the ion cyclotron resonance spectroscopy to determine the non-dissociative attachment rate coefficients for C_6F_6 (and C_7F_8 , c- C_4F_8 , C_7F_{14} , CCl_4). They obtained in C_6F_6 attachment rate coefficients a factor of ten lower than in SF_6 .

Studies of attachment by the Rydberg atoms technique (Hildebrandt *et al.* 1978, Marawar *et al.* 1988) showed that the rate coefficient is independent of the Rydberg electron velocity distribution. This indicates an s-wave capture process. The attachment rate, determined by quenching potassium atoms with principal quantum numbers up to 100, amounts to about $0.45 \cdot 10^{-7}$ cm³ s⁻¹ which corresponds to a cross-section value of about $240 \cdot 10^{-20}$ m² at 1.0 meV (Marawar *et al.* 1988). Gant and Christophorou (1976) reported attachment CS of $123 \cdot 10^{-20}$ m² at 30 meV and of $8.04 \cdot 10^{-20}$ m² at the 0.73 eV peak. Chutjian and Alajajian (1985) studied electron attachment at very low energies for C_6F_6 (and $2-C_4F_8$, $c-C_5F_8$, C_7F_8) using the Kr photoionization technique. They normalized the attachment CS to the rate coefficient $1.5 \cdot 10^{-7}$ cm³ s⁻¹ at 0.05 eV mean electron energy as extracted from swarm experiments (Spyrou and Christophorou 1985). Chutjian and Alajajian gave the following analytical formula for the CS:

(3)
$$\sigma_{\rm A} = N \left[a E^{-0.5} * \exp[-E^2/\lambda^2] + \exp[-E/\gamma] \right]$$

with E in eV, $N=389\cdot 10^{-20}$ m², $a=8.14\cdot 10^{-2}$ eV^{1/2}, $\lambda=4.33\cdot 10^{-3}$ eV, $\gamma=3.57\cdot 10^{-2}$ eV. According to these parameters, the attachment CS would amount to $380\cdot 10^{-20}$ m² at 1 meV. Note that a slightly modified formula was given in Chutjian

and co-workers' corrigendum of the C₂H₂Br₄ data (Alajajian et al. 1996):

(4)
$$\sigma_{\rm A} = N \left[a E^{-0.5} + \exp[-E^2/\lambda^2] + \exp[-E/\gamma] \right].$$

While the attachment CS in c-C₅F₈ and C₇F₈ approaches the theoretical limit for the s-wave electron capture, the C₆F₆ cross-section is by a few folds lower.

The apparent low-energy attachment coefficient rate in C_6F_6 decreases with increasing gas temperature (Spyrou and Christophorou 1985, Adams *et al.* 1985). This observation has been explained by the rise of the detachment coefficient rate with the increase of temperature (Chen *et al.* 1985, Christophorou 1985, Knigton *et al.* 1992, Datskos *et al.* 1993).

A second peak in electron attachment CS, observed in swarm (Gant and Christophorou 1976, Datskos et al. 1993) and pulse radiolysis (Shimamori et al. 1994) experiments at about 0.7–0.8 eV, has been associated with the electron capture to the π_6 -orbital (Datskos et al. 1993). Two more maxima were observed in electron-beam experiments. Around 4.5 eV F⁻ ions and, in less amount, $C_6F_5^-$ (25%) ions were observed (Fenzlaff and Illenberger 1989, Oster et al. 1989). In a broad peak (from 8 to 13 eV) for the F⁻ ion production, three sub-structures are visible. At 8.5 eV, $C_6F_5^-$ ions are also produced, while at higher energies only F⁻ is produced (Fenzlaff and Illenberger 1989, Oster et al. 1989, Weik and Illenberger 1995). The 8–13 eV structure also shows other fragment anions, like $C_6F_3^-$ (Fenzlaff and Illenberger 1984).

Electron attachment to gases in the pentafluorobenzene series C_6F_5X (X = F, Cl, Br, I) was studied by Herd et~al.~(1989) and Shimamori et~al.~(1993, 1994), and in the monohalobenzene series C_6H_5X (X = F, Cl, Br, I) by Shimamori et~al.~(1995). For all the four pentafluorobenzenes the same rate constant of $2 \cdot 10^{-7}~\rm cm^3~\rm s^{-1}$ (independently of the buffer gas used) was measured at thermal energies (Shimamori et~al.~1993, 1994). The parent anions $C_6F_6^{-*}$, $C_6F_5Cl^{-*}$, $C_6F_5Br^{-*}$, $C_6F_5l^{-*}$ have lifetimes long enough to be stabilized in the collisions with the buffer gas (Shimamori et~al.~1993, 1994). The electron attachment CS falls monotonically with energy in the 1 meV–1 eV range from about $1.5 \cdot 10^{-17}~\rm m^2$ at 1 meV for all the three C_6F_5Cl , C_6F_5Br , C_6F_5I molecules (Shimamori et~al.~1994), see fig. 19. The fall, up to 0.01 eV, follows approximately an $E^{-1/2}$ dependence, in agreement with Wigner's (1948) threshold law. At about 0.1 eV the CS in all four gases C_6F_6 , C_6F_5Cl , C_6F_5Br , C_6F_5I approaches the s-wave limit for attachment (see fig. 19). In C_6F_6 at 1 meV, according to the measurements of Shimamori et~al.~(1994), the CS is $1.7 \cdot 10^{-17}~\rm m^2$ and shows a maximum at about 0.75 eV, as described before.

In monohalogenated benzenes C_6H_5X , no parent anions were observed but only dissociated X^- ions (Shimamori et~al.~1995). Only the C_6H_5I dissociative attachment CS shows a monotonic dependence starting from a $6.4 \cdot 10^{-19}~\text{m}^2$ value at 1 meV, similar to that in the C_6F_5X group. For C_6H_5Cl and C_6H_5Br a sharp rise of the electron attachment CS by 2–3 orders of magnitude was observed between 0.05 and 0.5 eV (Shimamori 1995 et~al.), see fig. 19. The maximum CS value in C_6H_5Cl and in C_6H_5Br is $5.5 \cdot 10^{-22}~\text{m}^2$ and $1.2 \cdot 10^{-21}~\text{m}^2$ at 0.7 and 0.3 eV, respectively (Shimamori et~al.~1995). Swarm measurements (Gant and Christophorou 1976, Datskos et~al.~1993) indicated a maximum of attachment CS in these two gases at about 0.8 eV. As estimated by Shimamori et~al.~(1995), the dissociative attachment in C_6H_5I is exothermic by 0.24 eV and in C_6H_5Br , C_6H_5Cl , C_6H_5F endothermic by 0.14, 0.54 and 2.02 eV, respectively.

Several works made a systematic comparison of temporary negative-ion resonance

energies for benzene and fluorine substituted benzenes (see, for example, Hahndorf and Illenberger 1997). Naff et al. (1968) determined that each fluorine atom added to the benzene ring increases the electron affinity by about 0.4 eV. More recent works (Jordan et al. 1976b, Frazier et al. 1978) and calculations (Hinde et al. 1978, Birch et al. 1980) reported a smaller increase of the electron affinity (0.26 to 0.30 eV); for example from -1.15 eV in C_6H_6 to about -0.85 eV in C_6H_5F (Jordan et al. 1976b, Frazier et al. 1978). A slightly higher increase is observed for other halogen-atom-substituted benzenes, like C_6H_5Cl , C_6H_5Br (Jordan et al. 1976b).

Resonances. Frazier et al. (1978) used an electron transmission technique to determine the position of resonances in the C_6H_6 – C_6F_6 series. They reported two resonant states in C_6F_6 : at 0.42 and 4.50 eV. None of these states was seen in the TCS (Kasperski et al. 1997). The positions of these resonances are somewhat closer to those in C_6H_6 , i.e. 1.35 and 4.80 eV (Frazier et al. 1978) than the theoretical predictions, 0.13 and 4.02 eV.

Effects in resonant states due to the substitution of hydrogen atoms in C_6H_6 with halogen atoms and different functional groups were also studied by Compton *et al.* (1968), Christophorou *et al.* (1974), Burrow *et al.* (1976), Jordan *et al.* (1976a, b), Giordan and Moore (1983) and those due to the substitution of carbon atoms in C_6H_6 with nitrogen by Pisanias *et al.* (1972) and Nenner and Schulz (1975).

Lunt et al. (1999) observed shape resonances in the TCS of C_6H_5F , C_6H_5Cl and C_6H_5Br at about 0.7–1.0 eV. These resonances form a family which, including that in C_6H_6 at 1.17 eV, moves down in energy, illustrating increasing stabilization of the molecular orbital with decreasing electronegativity of the substituent (Olthoff et al. 1985). Resonance energies are 0.87, 0.75 and 0.69 eV for C_6H_5F , C_6H_5Cl and C_6H_5Br , respectively (Lunt et al. 1999). An additional structure in the C_6H_5F resonance has been attributed (Lunt et al. 1999) to a superposition of outgoing and reflected wave functions formed during the resonance lifetime.

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WOODIN R. L., FOSTER M. S. and BEAUCHAMP J. L., J. Chem. Phys., 72 (1980) 4223.

 $3^{\circ}8$. Sulphur hexafluoride (SF₆). – The very high cross-section for electron attachment at near-to-zero energies promotes the extensive use of SF₆ in electrical switches and high-voltage power lines. For this reason, a number of works have been devoted to modeling SF₆ plasmas and discharges and numerous semiempirical cross-section data sets have been published (see, e.g., Kline et al. 1979, Dincer and Govinda Raju 1983, Hayashi and Nimura 1984, Novak and Fréchette 1984, Phelps and Van Brunt 1988, Shimamura 1989, Itoh et al. 1988, 1991, 1993, Meyyappan 1997). In particular, numerical sets of electron scattering CS in SF₆ are also available in internet (Morgan 1999, Phelps 1999). Experimental total and partial CS are shown in fig. 20.

Total cross-sections have been measured with use of the time-of-flight absolute method by Ferch et al. (1982) in the 0.036–1.0 eV energy range, by Kennerly et al. (1979) using the same method at 0.5–100 eV, by Wan et al. (1993) at 0.2–12 eV using a trochoidal monochromator, by Romanyuk et al. (1984) in the 0.25–25 eV energy range using an electron trap and normalizing their data to those of Kennerly et al. (1979) at 7 eV, by Dababneh et al. (1988) at 1–500 eV who used a magnetically guided electron (or positron) beam and by Zecca et al. (1992) at 75–4000 eV using an absolute transmission method in a modified Ramsauer configuration. The agreement between different results is rather good, as compared in detail by Kasperski et al. (1997). The data in the latter paper agree well with the measurements of Dababneh et al. (1988) at low energies; at 100 eV the data of Kasperski et al. are 10% higher than the measurements of Dababneh et al. but coincide with those of Zecca et al. (1992). The data of Kennerly et al. (1979)

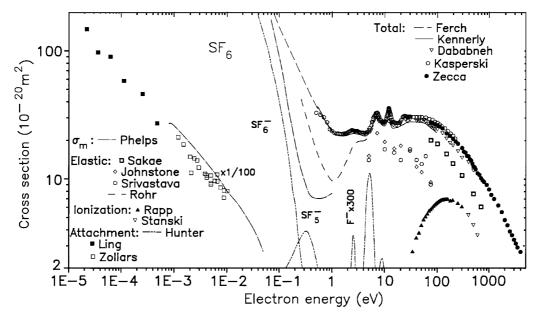


Fig. 20. – Integral cross-sections for electron scattering on SF₆. Total: Ferch et al. (1982); Kennerly et al. (1979); Dababneh et al. (1988); Kasperski et al. (1997); Zecca et al. (1992). Elastic: Rohr (1979); Johnstone and Newell (1991); Srivastava et al. (1976); Sakae et al. (1989). Momentum transfer: Phelps (1999), based on Phelps and Van Brunt (1988). Electron attachment: Ling et al. (1992) and Zollars et al. (1985), Rydberg atoms; Hunter et al. (1989). Ionization: Rapp and Englander-Golden (1965); Stanski and Adamczyk (1983).

above 20 eV are slightly lower than the measurements of Dababneh et al. (1988). The difference between the two data sets increases with rising energy. This can be partially attributed to scattering on exit orifices, as stated for measurements from Bonham's group for CH₄ and for chlorofluorocarbons (Jones 1986). At 1–12 eV the data of Kennerly et al. (1979) and those of Wan et al. (1993) coincide; below 1 eV the TCS of Wan et al. is somewhat lower, but the error can also be due to an energy shift. Romanyuk et al. (1984) obtained at the cross-section maxima slightly higher values than Dababneh et al. (1988) and observed an additional peak at 17 eV. The latter structure does not appear in any other measurements; therefore the data of Romanyuk et al. are not shown in fig. 20. At energies above 100 eV the data of Dababneh et al. (1988) are somewhat lower than the measurements of Zecca et al. (1992). As discussed by Dababneh et al., their data can suffer from a few per cent angular resolutions error in the high-energy limit.

The TCS rises in the limit of zero energy reaching a value of $400 \cdot 10^{-20}$ m² at 0.037 eV (Ferch et al. 1982). In the whole 1–100 eV range the TCS in SF₆ exceeds the value of $20 \cdot 10^{-20}$ m². Several maxima observed in the low-energy region have been attributed to resonant states. Their positions have been determined both in TCS measurements and in a transmission experiment using a trochoidal spectrometer (Kennerly et al. 1979). The width of the structures in the TCS, a weak one centered at 2.52 eV and two stronger ones, centered at 7.05 and 11.87 eV, is of about 1 eV. No other fluorine-containing molecule (CF₄, SiF₄, GeF₄, C₂F₆, C₃F₈, see figs. 7, 15, 17, 18) shows such narrow maxima in TCS.

Stricklett and Burrow (1991) examined the variation of the TCS for scattering on laser-excited SF_6 molecules. At energies below 1 eV they observed, similarly as it was seen in CO_2 (see subsect. 4.3 in part II), higher cross-sections for scattering on vibrationally excited states.

Swarms. Early measurements of electron diffusion coefficients in SF₆ were reviewed by Gallagher et al. (1983) and analyzed by Phelps and Van Brunt (1988). More recently, the electron drift velocity at intermediate reduced fields (280–700 \cdot 10⁻²¹ Vm²) have been measured by Nakamura (1988) and at high reduced fields (560–3500 \cdot 10⁻²¹ Vm²) by Lisovskiy and Yegorenkov (1999). This last measurement yielded at 2800 \cdot 10⁻²¹ Vm² a lower (8 \cdot 10⁷ cm/s) drift velocity than the semiempirical model of Itoh et al. (1990) but in agreement with the model of Phelps and Van Brunt (1988).

A scattering length of $-9.15a_0$ for SF₆, corresponding to the zero-energy cross-sections of $73.6 \cdot 10^{-20}$ m² (not shown in fig. 20), has been obtained by the Fano method (Evans et al. 1998).

The elastic cross-section below 6 eV has been obtained by Rohr (1979), between 5 and 75 eV by Srivastava (1976) with 15–55% declared errors, between 5 eV and 75 eV by Johnstone and Newell (1991), between 75 and 500 eV by Sakae et al. (1989). The integral elastic CS exhibits a minimum of about $10 \cdot 10^{-20}$ m² at 0.5 eV, then rises to about $20 \cdot 10^{-20}$ m² at 4 eV (Rohr 1979). The sum of the integral elastic and vibrational CS (Rohr 1979) at 0.5–6 eV, i.e. below the threshold for electronic excitation (Sze and Brion 1990) coincides with the TCS of Kennerly et al. (1979). On the other hand, the integral CS of Srivastava et al. (1976) are systematically lower by 30% than the elastic CS of Rohr (1979). This latter discrepancy could be caused by the normalization procedure (to the He cross-section) in the measurements of Srivastava et al. (1976). The elastic CS of Johnstone and Newell (1991) between 5 and 75 eV follows generally the descending trend of Srivastava's et al. data and is substantially lower at 75 eV than the result of Sakae et al. (1989). The recent integral CS measurements (Cho et al. 2000) between 2.7-75 eV agree within the error bar with the results of Rohr (1979) at low energies and

with those of Sakae et al. (1989) at 75 eV.

The low-energy DCS (Rohr 1979) indicate a direct scattering mechanism at 1 eV, while in the 2–3 eV range and around 7 eV the scattering is dominated by resonances. Enhancement of the elastic DCS (at 20° scattering angle) was also reported at 12 eV (Johnstone and Newell 1991).

The very low-energy DCS at 90° has been studied in the Daresbury synchrotron-light laboratory (Randell $et\ al.$ 1992). The 90° DCS possibly shows a maximum at 0.15 eV. Randell $et\ al.$ pointed out that this could happen because of the channel-coupling effects: the low-energy rise of the CS for the SF $_6^-$ formation depresses the elastic channel by denying it the s-wave.

The geometrical structure of the molecule and the interference effects were studied in numerous high-energy scattering experiments (Miller and Fink 1981, Goates and Bartell 1982, Miller and Fink 1992, Meier *et al.* 1993). The elastic scattering at near-to-zero angles was studied by Miller and Fink (1992) at 30 keV. The asymptotic value of the DCS at zero degrees amounts to $257 \cdot 10^{-20}$ m²/sr.

Vibrational excitation has been studied by Rohr (1977 and 1979) at energies below 10 eV and by Randell et al. (1992) at fixed 90° angle between 0.05 and 1 eV. At energies below 1 eV a forward-peaked angular distribution (Rohr 1979) indicates a direct excitation process. The energy loss spectrum at 0.4 eV at 90° (Randell et al. 1992) indicates that as much as 50% of the TCS at this energy can be due to the vibrational excitation of the unresolved ν_1 (Raman, strong) and ν_3 (infrared-active, strong) modes and their overtones (see Randell et al. for assignment of the states). A sharp peak was observed 60 meV above the vibrational excitation threshold (Rohr 1977). Randell et al. stated that the near-to-threshold excitation of the ν_1 mode follows the Born approximation but the ν_3 mode is excited via a resonance mechanism.

At 1 eV the ν_1 mode and its overtones up to n=7 dominate the vibrational excitation (Rohr 1977). An enhancement of the vibrational excitation was observed in the 2–3 eV and 7 eV regions (Rohr 1979) and at 12 eV (Trajmar and Chutjian 1977), in coincidence with resonant peaks in the TCS.

Electronic excitation spectra at 20–30 eV impact energies were studied by Trajmar and Chutjian (1977) and at 400 eV by Simpson et al. (1966). DCS at 20 eV for the lowest, 9.8 eV energy loss structure is uniform in angle, while the higher, 11.0, 11.6, 12.8 and 13.3 eV energy loss structures are somewhat forward-centered (Trajmar and Chutjian 1977). A rough evaluation of the integral CS at 20 eV from the data of Trajmar and Chutjian gives a value of about $0.8-1.0 \cdot 10^{-20}$ m².

The inner shell excitations were studied by Hitchcock and Brion (1978), Ying et al. (1993), Francis et al. (1995), Turci et al. (1995). The valence and inner-shell excitations in SF_6 , SeF_6 and TeF_6 at high impact energy (2.0–3.7 keV) and zero degree scattering angle were studied by Sze and Brion (1990) but no absolute intensities were given. Electron-ion coincidence measurements at 160-230 eV impact energy were done by Hitchcock et al. (1978). Theoretical values of low-angles DCS for some optically allowed transitions were given by Fomonung et al. (1996).

The optical emission cross-section for the molecular continuum between 200 and 340 nm amounts to as little as $0.29 \cdot 10^{-22}$ m² at 200 eV (Jabbour et al. 1988). The overall emission in the 620–790 nm range due to transitions between the highest and the lowest excited states $(2p^43p) \rightarrow (2p^43s)^{2,4}P$ of the fluorine atom equals $1.6 \cdot 10^{-22}$ m² at 100 eV, compared to $3.1 \cdot 10^{-22}$ m² for CF₄ (Blanks et al. 1987, see also Roque et al. 1991). The emission spectrum in the 200–600 nm range due to 200 eV electron excitation consists of a broad feature from 200 to 320 nm and several line emissions from excited fluorine

fragments (Blanks and Becker 1987).

Dissociation into neutrals was studied by Ito et al. (1995) up to 195 eV. The yield of SF_2 radicals reaches a maximum at 145 eV; the relative yields for SF_3 and SF radicals are 0.5 and 0.05 of the one for SF_2 . Metastable radicals production was studied by Corr et al. (1987). Onsets for F_2 and F production in Rydberg states were observed at 18.7 and 35.0 eV, respectively; no cross-sections were given.

Ionization. The total ionization CS up to 300 eV has been given by Rapp and Englander-Golden (1965) and the normalized partial cross-sections up to 600 eV by Stanski and Adamczyk (1983). The SF_5^+ and SF_3^+ yields amount to about 50% and 15% of the total ionization CS; no stable SF_6^+ ions were observed. Peculiar, with respect to gases like CH₄ or CF₄, is the high-energy (150 eV) at which the ionization CS reaches the maximum; this reflects a high ionization threshold in SF_6 (15.8 eV).

Doubly differential CS for ionization at incident energies of 100 and 200 eV have been measured by Al-Nasir *et al.* (1996).

Electron attachment. SF₆ forms both parent SF₆ ions and dissociated anions. Different experiments (Rapp and Briglia 1965, Stamatovic and Schulz 1968) proved that electrons are attached to SF₆ at collision energies close to zero. The cross-section for SF₆ yield derived from Rydberg's atoms quenching (Weast et al. 1976, Foltz et al. 1977, Hildebrandt et al. 1978, Zollars et al. 1984, 1985, 1986, Dunning 1987, Ling et al. 1992) or laser photoelectron attachment (Chutjian 1981, Chutjian and Alajajian 1985, Klar et al. 1992b, 1994) follows, in the zero-energy limit (0.1–10 meV), the threshold $E^{-1/2}$ law for s-wave capture on a polarization potential (Vogt and Wannier 1954, Klots 1976, Gauyacq and Herzenberg 1984). These data merge well at energies above 20 meV with the cross-sections derived from swarm (Hunter et al. 1989), threshold photoionization (Chutjian and Alajajian 1985), flowing afterglow (Smith et al. 1984, Smith and Španěl 1994), microwave conductivity (Shimamori et al. 1992) and beam (Wan et al. 1993) experiments. At energies of the order of 0.1 eV (Klar et al. 1992b, Wan et al. 1993) the slope of the electron attachment CS changes as E^{-1} . Some disagreement regards CS at energies between 0.2–0.3 eV, where a peak for SF₆ occurs. While Kline et al. (1979) and Hunter et al. (1989) observed a quick fall of the attachment CS in this region, Wan et al. (1993) reported a smoothly changing CS. Wan et al. argued that the quick fall observed in previous experiments was an artifact, caused by decay in flight of SF₆ ions formed in excited vibrational states. Popple et al. (1992) in a Rydberg atoms experiment showed that at very low energies the SF_6^{-*} ion can be stabilized in post-collisional interactions between ions.

The attachment rates at sub-thermal energies, about $4\cdot 10^{-7}$ cm³ s⁻¹, is lower for SF₆ than for CCl₄ (about $7\cdot 10^{-7}$ cm³ s⁻¹, Harth et~al.~1989, Dunning 1987); the same is found at thermal energies (300 K): $2.3\cdot 10^{-7}$ cm³ s⁻¹ and $4.0\cdot 10^{-7}$ cm³ s⁻¹, respectively (Shimamori 1992). Qualitatively the same result was obtained in flowing afterglow plasma experiments (Smith et~al.~1984, Smith and Španěl 1994): at 300 K the attachment rate coefficient amounts to $3.1\cdot 10^{-7}$ cm³ s⁻¹ for SF₆ and $3.9\cdot 10^{-7}$ cm³ s⁻¹ for CCl₄; at 590 K: $4.0\cdot 10^{-7}$ cm³ s⁻¹ and $3.5\cdot 10^{-7}$ cm³ s⁻¹, respectively (Smith et~al.~1984). The electron attachment CS in SF₆ reaches lower values at the same energy than that in CCl₄, compare figs. 11 and 20. This, according to the model of Vogt and Wannier (1954), reflects the dependence on molecular polarizability which is $6.54\cdot 10^{-30}$ m³ for SF₆ and $10.8\cdot 10^{-30}$ m³ for CCl₄.

Klar et al. (1992a, 1994), studying dissociative attachment via a high-resolution photoelectron attachment technique, noticed resonant structures in the cross-section at energies coinciding with threshold energies for vibrational excitation of the ν_1 and ν_3 modes.

Table X. – Integral cross-sections for electron scattering on SF₆ (in 10^{-20} m² units).

Energy	Elastic	Vibrational	Electronic	Ionization	Summed	Total
5.0	21.6 RO 14 S	1.0 RO			22.6	22.5 K
	15.1 J				16.1	
10	17 S					25.7 K
	19.4 J					28.0 KA
20	14 S		1.0 S1		15	26.0 K
	18.1 J				19.1	28.2 KA
30	17 S		1.5** S1	1.93 R	20.4	27.8 K
	16.4 J				19.8	30.5 KA
50	13 S		1.2** S1	4.35 R	18.6	27.5 K
75	20.0 SA		1.0** S1	5.77 R	26.8	27.7 Z
	9.0 S					29.1 KA
100	18.7 SA		0.8** S1	6.53 R	26.0	26.3 Z
						27.4 KA
150	15.6 SA		0.6** S1	6.97 R	23.2	22.7 Z
200	13.1 SA		0.7** S1	6.83 R	20.4	20.6 Z
						21.7 KA
300	10.6 SA		0.4** S1	6.09 R	17.1	17.4 Z
500	7.59 SA		0.3** S1	4.34 ST	12.2	12.5 Z

^{**} Extrapolated values using Born-Bethe approximation

We recall a similar observation in the CH₃I molecule, see subsect. 3.3.

The SF $_5^-$ yield exhibits a maximum of about $3\cdot 10^{-20}$ m 2 at 0.3 eV (McCorkle et al. 1980, Hunter et al. 1989), while the production of other negative ions (F $^-$, F $_2^-$, SF $_3^-$, SF $_3^-$, SF $_2^-$) shows peaks at 2.5, 5.0, 9.0, 12 eV (Kline et al. 1979, Hunter et al. 1989, Fenzlaff et al. 1988, Rao and Srivastava 1993).

Little or no dependence on the temperature was observed (Spence and Schulz 1973, Datskos et al. 1993, Smith et al. 1984, Miller et al. 1994) for the SF_6^- yield, and a significant rise with temperature for the SF_5^- yield at low energies (Miller et al. 1994, Smith et al. 1995, Matejcik et al. 1995). This effect has been explained by Matejcik et

J - Johnstone and Newell (1991)

K - Kennerly et al. (1979)

KA - Kasperski et al. (1997)

R - Rapp and Englander-Golden (1965)

RO - Rohr (1979)

S - Srivastava et al. (1976), 15–55% declared errors

S1 - A rough evaluation (present work, $\pm 50\%$ possible error) from DCS at 20 eV by Srivastava and Trajmar (1977) and Born-Bethe extrapolation at higher energies

SA - Sakae *et al.* (1989)

ST - Stanski and Adamczyk (1983)

Z - Zecca et al. (1992)

al. (1995) as due to the endothermic character of the SF_5^- formation (0.12 eV).

Resonances. Shape resonances have been observed in electron attachment (see Hunter et al. 1989), in differential elastic (Rohr 1979), in vibrational excitation (Trajmar and Chutjian 1977), and in total CS (Kennerly et al. 1979, Romanyuk et al. 1984). Similarly to chlorofluoromethanes, see subsect. 3.2, the positions of these resonances seen in the attachment CS are shifted towards lower energies compared to those in the TCS (see fig. 20). The following symmetries, predicted by Dehmer et al. (1978), can be assigned to the peaks in transmitted current at the following energies (Kennerly et al. 1979): A_{1g} at 2.52 eV, T_{1u} at 7.01 eV, T_{2g} at 11.88 eV, and a broad structure centered at 25 eV of E_{g} symmetry. The last resonance was also studied in threshold photoelectron spectroscopy (Yencha et al. 1997). Stricklett and Burrow (1991) noticed that the T_{1u} (and T_{2g}) resonance shifts to a lower energy (and enlarges, respectively) for scattering on vibrationally excited molecules.

Sum check. As shown by Rohr (1979), the elastic CS in the 0.5–5.0 eV region sums up with the vibrational CS (Rohr 1979) within the experimental errors to Kennerly's et al. (1978) TCS. It is remarkable that at 1 eV the vibrational and elastic cross-sections are of a similar magnitude, $10 \cdot 10^{-20}$ m². At 20–50 eV the sum of elastic values by Srivastava et al. (1976) with the inelastic processes is significantly lower than the TCS values (Kennerly et al. 1979), see Table X. To reach agreement between the summed and experimental TCS, the elastic values of Srivastava et al. (1976) should be increased by approximately 40%. The elastic CS of Johnstone and Newell (1991) are higher than those of Srivastava et al. at 5–20 eV but lower at 30 and 50 eV.

At energies above 75 eV the sum of the elastic cross-sections of Sakae *et al.* (1989) and the ionization cross-sections of Rapp and Englander-Golden (1965) is almost equal to the absolute TCS values of Zecca *et al.* (1992).

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ZOLLARS B. G., WALTER C. W., KU F., JOHNSON C. B., SMITH K. A. and DUNNING F. B., J. Chem. Phys., 84 (1986) 5589. 3.9. Other hexafluorides (UF₆, WF₆). – Uranium hexafluoride (UF₆). The differential and integral CS, the elastic CS at 10–75 eV and electronic excitation CS at 10, 20, 40 eV in UF₆ have been measured by Cartwright et al. (1983) superseding the data of Srivastava et al. (1976). The integral elastic CS reaches a maximum of $28.1 \cdot 10^{-20}$ m² at 20 eV falling to $7.5 \cdot 10^{-20}$ m² at 75 eV; the overall electronic excitation CS at 20 eV amounts merely to $0.4 \cdot 10^{-20}$ m² (Cartwright et al. 1983).

Ionization of UF₆ was studied experimentally by Compton (1977) up to 1 keV and semiempirically by Margreiter et~al.~(1990). The experimental value at 80 eV is $16 \cdot 10^{-20}~\text{m}^2$; this value summed with the partial cross-sections of Cartwright et~al.~(1983) — the elastic value (7.5 · $10^{-20}~\text{m}^2$ at 75 eV) and the electronic excitation one (0.33 · $10^{-20}~\text{m}^2$ at 40 eV) — would give an upper limit on TCS of about $24 \cdot 10^{-20}~\text{m}^2$ at 80 eV. This is much lower than the experimental TCS value in WF₆, $34.4 \cdot 10^{-20}~\text{m}^2$ at the same energy (Szmytkowski et~al.~2000).

Tungsten hexafluoride (WF₆). The TCS for this molecule has recently been measured by Szmytkowski et al. between 1.2–250 eV and by Karwasz et al. (2000) between 75–3500 eV. The total cross-section in WF₆ is similar in shape to that in C₆F₆ (and C₆H₆), see Kasperski et al. (1997), but is shifted to higher energies. The TCS in WF₆ rises from $24 \cdot 10^{-20}$ m² at 8 eV, reaches a broad maximum of about $38 \cdot 10^{-20}$ m² at 40 eV and shows some shoulder structure at about 20 eV. The low-energy maximum in WF₆ at 3 eV is much higher $(31 \cdot 10^{-20} \text{ m}^2)$ than that in C₆F₆ at 1 eV.

The two TCS measurements in WF₆ disagree somewhat in the energy overlap, the data of Szmytkowski et~al. being 25% higher at 100 eV and 10% at 200 eV than the data of Karwasz et~al. The data of Karwasz et~al. can be somewhat overestimated in their low-energy limit, due to beam instabilities; on the other hand, the data of Szmytkowski at 300 eV can suffer from a higher angular resolutions error than those of Karwasz et~al. (2000). However, all these possibilities do not explain the discrepancy. In the high-energy range the TCS falls from $28 \cdot 10^{-20}$ m² at 200 eV to $5.0 \cdot 10^{-20}$ m² at 3000 eV.

Formation of negative ions in WF₆ was studied by Thynne and Harland (1973), de Wall and Neuert (1977), Hildebrand (1977), Dispert and Lackmann (1977). Electronic excitation energy loss spectra were studied by Rianda $et\ al.$ (1979).

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